

Petrological and Geochemical Evaluation of the Paleocene-Eocene Lithofacies in Dahomey Embayment, Southwestern Nigeria

Olugbenga A. Boboye, Olusegun J. Omotosho

Department of Geology, University of Ibadan, Ibadan, Nigeria Email: oa.boboye@mail.ui.edu.ng, boboyegbenga@yahoo.com

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Abstract

The detail mapping and logging of the Ibese area within the Dahomey Basin has revealed the total thickness of the limestone to range between 12.0 m to 15.0 m. This study is based on qualitative lithologic, spectrochemical analyses via X-ray fluorescence and petrographic indices to understand the lithofacie, chemistry, viability of the sequences and depositional environment. This consists of milky to grey and nodular to massive fossiliferous limestone units while the sand content increases with depth. The composition of the major oxides in the thirty-seven (37) samples from six exploratory wells revealed that CaO (51% - 56%), SiO₂ (2.6% - 10.56%), Fe₂O₃ (0.33% - 0.94%), MgO (0.78% - 1.02% and Al_2O_3 (0.72% - 0.98%) were the most abundant elements. The trend of the geochemical indices show increase in the percentage composition of CaO, and MgO down the sequences with high values which defined the depth of diagenetic re-distribution. The percentage composition of Al₂O₃, Fe_2O_3 and K_2O are significant indicating continental influence during diagenesis and diverse depositional episodes. The three different microfacies recognized based on depositional texture are shelly biomicrite, shelly biosparmicrite and alga biosparmicrite. The predominance of micrite as the cementing matrix revealed that the rock was deposited in a quiet shallow inner shelf environment occasioned by storm waves.

Keywords

Environment, Petrography, Geochemistry, Diagenesis, X-Ray Fluorescence

1. Introduction

The Dahomey Basin is an arcuate extensive basin located within the Gulf of Guinea, West Africa (Figure 1). It extends from the eastern part of Takoradi



Figure 1. Map of Africa showing the location of Nigeria and the generalized geological map of the eastern. Dahomey Basin (modified after Gebhardt *et al.* [27]).

Height in Ghana through Togo, Republic of Benin and then terminates at the Okitipupa ridge, south western Nigeria separating it from the Niger Delta. It runs parallel to the coastal states of these countries covering a total area of approximately 49,209 km² [1]. The states within the basin in the Nigerian sector include Lagos, Ogun, Ondo and Benin and the sediments' thicknesses increase toward the sea (**Figure 1**). The northern part of the basin in Nigeria is the exposure of the Abeokuta group which unconformably overlying the basement while towards the sea, there is a progressive increase in the thickness of chemical and biochemical sediments such as gypsum, glauconite, limestone and other associated marine deposits which is then capped by coastal sands [2] [3]. Numerous works have been carried out on the Ewekoro and Akinbo formations which are exposed at different sections of eastern part of the Dahomey Basin [3] [4] [5].

Omatsola and Adegoke described the structure and stratigraphy of the Abeokuta formation [6]. Ogbe proposed that the overlying finely laminated shale at the Ewekoro quarry is Akinbo Formation [7]. Ekweozor and other workers studied the origin of tar sands, organic geochemical investigation of bitumen, properties and source rock evaluation of interbedded shale of southwestern Nigeria [8] [9]. They concluded that the tar sand deposits represent the product of reservoir transformation of hydrocarbon by anaerobic micro-organisms. The study of the subsurface in Dahomey Basin of upper Cretaceous and lower Tertiary of Bodashe-1 and Ile paw wells was carried out and concluded that the sediments belong to the Araromi Formation [10], while the use of pebble morphometry associated with the siliciclastic sediments of Abeokuta group was used to deduce their depositional environment [11] [12]. Boboye and Raji reported that the biostratigraphy and petrography of the sequences were deposited in quiet shallow marine environments and that the presence of Toweiuscallosus and Coccolithusformosus indicated that Oshosun and Ewekoro Formations were Ypresian and Thanetian to Ypresian age [3]. This rock is a potential hydrocarbon reservoir for the overlying shale while this character of the rock is seen in the Benin Republic sector of the basin, where petroleum is produced at marginal level [3] [13] [14]. The aspect geochemical evaluation, depositional environment and reserve estimation of limestone deposit were also reported around Orimogija and Okeluse areas of eastern Dahomey Embayment [15] [16]. However, this study focus on the distribution of diverse lithofacies, genetic description of the Ewekoro Formation with emphasis on the limestone and what characterize the environment of its deposition, the extent of diagenesis, biological influence on the chemistry of the units, major elemental composition, post depositional processes and their viability in determining its economic values.

2. Stratigraphy of the Dahomey Basin

The Dahomey Basin is described by an extensive wedge of Cretaceous to Recent sediments which lies unconformably on the basement [1]. Theses rocks are upper Cretaceous Post Santonian which also outcrops in the River Niger, Bida Basin and the eastern sector part of the Anambra Basin down to the Okitipupa high. In the Dahomey Basin the three stage tectonic evolution allows the stratigraphic sequence to be divided into three sequences namely Precambrian to Triassic intracratonic rocks and Jurassic to lower Cretaceous, Continental to marginal marine rocks representing the pre-transform stage, the lower Cretaceous to late Albian rocks representing the syn-transform stage and Cenomanian to Holocene rocks representing the post-transform stage [9].

The lower Ise Formation which overlies the basement contains conglomerates, coarse to medium grain sandstones and shales that were deposited in continental and deltaic environments. The conglomerates are imbricated and at some locations, ironstones occur [10]. The conglomerates are poorly sorted, micaceous and more or less clayey [1]. This formation which is part of the Abeokuta group can be classified under the groups of rocks formed in the pre-transform stage between Upper Jurassic to Neocomian, [9] [11] [12]. This age is more representative on the offshore part of the basin which may reach about 2000 m thick though this has not been reached by drilling but geological interpretation have been inferred from various seismic data [13]. This formation also is represented in the Neocomian. Cross bedding is noticeable while the rock is soft and friable [1].

Overlying the Ise Formation of the Abeokuta group is the Afowo Formation which is made up coarse to medium grained sands and interbedded shales, siltstones and claystone with the sandy facies bearing tar while the shales are organic rich [12] [18] (Figure 2).

The lower part of this formation which overlie the Ise Formation is transitional with mixed brackish to marginal horizons that alternate with well sorted, sub-rounded sands indicating a littoral or estuarine near shore environment of



Age	Formation	Lithology
Pleistocene-Oligocene	Coastal Plain Sands	
Eocene	llaro Oshosun	
Palaeocene	Akinbo Ewekoro	
Maastrichtian-Neocomian	Araromi Afowo Ise	
Precambrian	Crystalline Basement	

Figure 2. Stratigraphy of Nigerian eastern sector of Dahomey Basin (modified after Omatsola and Adegoke, [4]).

deposition [2]. Brownfield and other workers described fairly sorted sandstone interbedded with shale beds overlying a shale and siltstone sequence deposited as a reworked fan delta in a marginal marine to inner shelf environment [19]. This unit corresponds to the beginning of the post transform stage in the basin and is dated as Turonian. The upper part of this formation grades into Maastrichtian in different parts of the basin [19].

3. Location of the Study Area and Geology

The study area is located in Ibese, southwestern Nigeria about 40km southwest of Abeokuta where quarrying activities have revealed rocks of diverse ages and lithologies within the basin (**Figure 1**). The geology as exposed at different road cuts and quarry exposures show the occurrence of lateritic sandy horizon which is the topmost layer. The underlying lithology in this sequence is sandy shale with lateritic concretions which is succeeded by the occurrence of shale. At other sections where weathering is minimal, this layer show glauconitic nodules followed by the occurrence of partially weathered poor fossiliferous limestone units succeeded by a fossiliferous horizon which extends to a significant depth before terminating on marl. Intercalating the marl units are sandy shale and poorly sorted sandstones.

4. Materials and Methods

4.1. Field Study

The field mapping was carried out in order to determine different rock units and their association within the area. The samples constitute limestone, shale and sandstone. A systematic mapping and sampling was adopted at five meters' intervals. The lithostratigraphic descriptive geology of the study area is well described in the quarry site where exposures of the rocks can be studied (Figures **3-9**). The rocks are structurally undeformed and strike east west running parallel to the coastline. The rocks association were defined from core samples with the uppermost unit extending up to about 0.7 m.

4.2. Laboratory Analyses

The samples were disaggregated and subjected to lithological description and petrography in the laboratory while the samples for the XRF analysis were pulverized into a fine powder. Between 10 mg - 15 mg of the samples were subjected to X-ray fluorescence Spectrometric analysis which was done using the Rigaku XRF Spectrometer with a rhodium (Rh) target. This runs at 50 KV/50 mA with full vacuum and a 25 mm mask for the analysis of various major oxides with the detectable limits for the major elements which were 0.01% for SiO₂, Al₂O₃, CaO and 0.04% for Fe₂O₃ both at 100% upper limit. The petrographic slides were prepared at the thin section laboratory, where thin sections approximately 30 µm thick of the samples were produced for microscopic study.

5. Results and Discussion

5.1. Lithostratigraphy

The lithofacies that constitute the study area are sandstone, clay, shale, limestone and marl. The sandstones are fine grain with the ferruginized sandy clay units



Figure 3. A lithostratigraphic section as exposed in Ibese.



location 1

coordinates: 06°48'46.6" 03°7'55.32" elevation: 50.55 m



Figure 4. Lithostratigraphic profile of well-1.

Coordinates: 06°48'58.6" 03°8'45.02"

Elevation: 50.60 m Total Depth: 20.50 m



Figure 5. Lithostratigraphy profile of well-2.



Coordinates: 06°46'46.6" 03°11'27.32"

Elevation: 50.24 m Total Depth: 18.50 m



Figure 6. Lithostratigraphy profile of well-3.

Coordinates: 06°48'46.6" 03°12'15.32" Elevation: 49.84 m



Figure 7. Lithostratigraphy profile of well-4.



Coordinates: 06°48'46.6" 03°12'59.32"

Elevation: 49.32 m



Figure 8. A lithostratigraphic profile of well-5.

Coordinates: 06°49'04.6" 03°13'47.12" Elevation: 50.60 m Total Depth: 20.50 m



Figure 9. Lithostratigraphy profile of well-6.



significantly difference from the upper units. The clay and shale units are lateritized occurring as concretions ranging from brown to reddish brown and grey in colour respectively. The shale units are rich in glauconite and the limestone units constitute the weathered, poorly fossiliferous partly to the telltale of weathering on the layer and unweathered units which grades from the upper layer in a diffuse manner. The diagenetic process has had its toll on this layer due to pore spaces (mouldic and vuggy porosities) which shows casts of the disintegrated fossils (**Figures 3-9**). The marl units are described as the impure limestone. The deeper section of this unit shows contact between the basal weathered marl and a fresh top layer with the former occurring in the form of lateritic clay concretions.

5.2. Geochemistry

The major elements from the six wells within the Ewekoro Formation put the average percentages of major oxides across the wells as CaO (51% - 56%), SiO₂ (2.6% - 10.56%), Fe₂O₃ (0.33% - 0.94%), MgO (0.78% - 1.02% and Al₂O₃ (0.72% - 0.98%) (Appendix: **Tables 1-6**). The percentage concentration of CaO increases down the formation to approximately 13.0 m depth, where the highest value was recorded which tapper down the formation to the limestone-marl contact. The average percentage composition of CaO is 52.6% which indicate the dominancy of this oxide. This corroborate with the petrographic studies with the occurrence of carbonate bioclasts within the lime mud matrix [3] [19] [20]. The high ratio observed in calcium and magnesium defines a high calcite purity which is significance in the cement production. The presence of magnesium in carbonate rocks is a function of the temperature of deposition and time due to its removal by interstitial solution while dolomitization process could also contribute to high magnesium content [21].

The increase in the amount of CaO at different depths clearly defines diverse carbonate lithologic units. At 13 m depth, there is a significant increase in the amount of MgO, this significant enrichment is attributed to increase in the amount of coralline algae and/or had an initial dolomitization where the water in the basin may be partially isolated under conditions of aridity and hence enriched in magnesium by continual inflow of sea water and subsequent precipitation of carbonate. Enrichment in this manner would form a heavy brine solution that moved down through porous sediments below and dolomitize those carbonates with which it came in contact with [21] [22] [23]. This depth of replacement serves as the datum of skeletal accumulations at water thermodynamic condition. At depth ranging from 12 m to about 14 m, there is an increase in fossil content with large primary porosity which creates the avenue for replacement. At other deeper depths there is an inversion of aragonite to calcium and also consequent shedding of carbonate sediments from above [3].

The geochemistry of well-1 defined the depth of carbonate diagenetic redistribution complemented by the amount of SiO_2 measured from various horizons (Appendix: **Table 2, Figure 10** and **Figure 11**). At 12 m to 14 m depth,



Figure 10. A lithofacies correlation of study wells showing diverse horizons.

MgO peaked showing the depth of favoured carbonate deposition and/or aragonite-calcite inversion which also correspond with the increase in the value of SiO_2 . This helps in inferring the environment of deposition and the diagenetic redistribution [21] [24]. Fairly high SiO_2 percentage range between 2.4% to 10.38% (average of 4.363%), suggest that the continental areas have an effect on the limestone though diagenesis and leaching tend to minmize such effects. Feldspars containing silica had been broken down during transportation before the commencement of diagenetic alterations. The presence of alumina (Al_2O_3), confirms the occurrence of alumino silicates (feldspars) and micas in the adjacent areas that have being subjected to partial weathered prior diagenesis [3] [18] [22]. The average percentage composition of Al_2O_3 in the shale units (12.63%) and



Figure 11. A Ternary plot showing degree of diagenesis in the study wells.

limestone (0.693%) has constituted a slight alumina contaminant to the upper limestone units viametoric water percolation.

The modal composition of Fe_2O_3 in the Ewekoro Formation is low compared to other oxides in the well. The low Fe^{3+} confirms the suboxic character of the palaeoenvironment responsible for the deposition. The Fe/Mg ratios measured

in the upper limestone horizon with an average of 0.40 is an indication of continental influence [22]. The moderate amount of both seems to reflect in the ferruginous syn-sedimentary and post depositional inputs of these sediments by surface meteoric waters flushing rapidly through ferromagnesian rich minerals (shale and glauconite) overlying the limestone. The rocks are stained at some parts with reddish brown colouration when exposed and it is seen that some of the allochems exhibit some iron oxide stained under the microscope [3]. However, the results have shown that the limestone sequence bounded by thick shales, correspond to higher value of Fe/Mg ratio, due to the occurrence of shalev glauconitic bands and leaching of smectite clavs into the unit. The result of the major oxides in well-2 has shown that the depth at which Mg occur in this well tend to become shallower to that measured in Well-1 suggesting the possible increase in water level thus causing the deposition of the carbonate at a shallower depth. The depth of SiO₂ still remain at 13.5 m indicating a common diagenetic depth for both horizons. Silica percentage (SiO₂) is low and range between 2.4% to 10.38% (average of 4.46%). This is an indication that the continental areas have a mild effect on the limestone through diagenesis [21]. The presence of alumina with average value of 14.43% (shale) and 0.66% (limestone) confirms the presence of little alumino silicates in the adjacent areas that had not been completely weathered prior diagenesis and also that which dripped on the upper limestone horizons with high alumina value of 2.44% in the limestone [21]. The modal composition of Fe_2O_3 is low relative to others hence suggesting low availability of oxygen to oxidize Fe²⁺ present in heavy minerals to Fe³⁺ during diagenesis with little effect of post depositional processes on the overlying glauconite (Appendix: Table 2, Figure 11 and Figure 12).

The amount of MgO relative to depth in well-3 defined the depth of carbonate primary replacement complemented by the amount of SiO₂ measured from various horizons which showed different peaks (Figure 11). Though there exists fluctuations at diverse horizons but the depth of high MgO and SiO₂ concentration is put at 12.0 m to 14.0 m. MgO peaked showing the depth of favoured carbonate deposition of aragonite - calcite inversion which also correspond with an increase in the amount of SiO₂. Two peaks were recorded for this horizon (Figure 11). An average of 64.91% and 4.89% were measured for the shale and limestone respectively. Relative to previous wells, a high percentage of SiO₂ was recorded at 13.0 m depth which indicate the presence of authigenic quartz and minor detrital quartz [21] (Appendix: Table 3). An average of 11.97% and 0.96% of alumina were recorded for the shale and limestone. A local enrichment of 4.2% was measured at the upper limestone horizons while a gradual decrease in Al₂O₃ with depth was recorded. This indicated the rate of Al₂O₃ percolating into the limestone from the shale unit. The K_2O value (0.06%) measured from the top of the limestone unit confirm with the shaly contamination. Average values of Fe₂O₃ (7.864% and 0.675%) were recorded for the shale and limestone while a peak of 8.36% at the shale-limestone contact suggest the partial oxidation of glauconite from Fe²⁺ to Fe³⁺.





Figure 12. Plots of MgO and SiO₂ percentage against depth for the study wells.

The ratio of MgO with depth in well-4 defined the depth of carbonate diagenetic re-distribution which is complemented by the amount of SiO₂ measured from various horizons with different peak (Figure 10). At 12 m to 14 m depth MgO peaked showing the depth of favoured carbonate deposition or aragonite-calcite inversion which also correspond with an increase in the amount of SiO_2 . SiO₂ percentage is fairly high and ranged from 1.45% to 10.56% (average of 3.87%). The rock can be described as chemically pure indicating that the continental areas have little effect on the limestone [24] (Appendix: Table 4). Feldspar containing silica had been weathered during transportation prior the commencement of diagenesis. The values of Al₂O₃ in the shale (11.8%) and limestone (0.81%) lithofacies confirms the presence of little alumino silicates in the adjacent areas that are partially weathered prior the diagenetic processes in the overlying shale facie. Significant shale contamination is well defined by the increase in percentage of K₂O (0.11%) measured at the shale-limestone contact [21]. The percentage composition of Fe_2O_3 measured in the shale (7.3%) increased compared to other oxides thus helps to infer the amount of oxygen available to oxidize Fe²⁺ present in heavy minerals to Fe³⁺. The increase in the amount of Fe₂O₃ in this horizon describes the intense chemical weathering of the continental areas (Appendix: Table 4).

In well-5 at 13.0 m depth, there is a sharp increase in both MgO and SiO₂ (Figure 12). This graphical behaviour is suggestive of an increase in the thickness of the deposit towards the continent, which showed a decrease in the extent of continental inputs as the deposits progrades on shore. Silica (SiO_2) percentage ranged from 1.4% to 10.2% (average of 5.1%) (Appendix: Table 5). This is an indication of the Continental areas having little effect on the limestone [22]. Alumina, (Al_2O_3) , increases in this horizon confirms the presence of alumino-sili- cates in the adjacent formations. The effect of shale contaminante in the limestone units is minimal. This corroborates with the average percentage of $K_{2}O$ (0.1%). The amounts of Fe decreases down the formation making it obvious that it is post depositional. This is due to the percolation of meteoric waters through shale and glauconitic layers into the underlying limestone units.

5.3. Carbonate Petrography

Three microfacies recognized are shelly biomicrite (bioclastic wackestone), shelly biosparmicrite (bioclastic wackestone-packstone) and alga biosparmicrite (Alga bioclastic wackestone-packestone) [21] [22] [25] (Appendix: Table 7 and Figures 13-15).

5.3.1. Shelly Biomicrite

This represents the topmost portion of the limestone layers. The rock is micrite (mud) supported and contain a substantial amount of bioclasts (grain). At different sections, the rock defines about 1/4 of spar. Bioclasts described from this layer include pelecypods which take a higher percentage of bioclasts population, gastropods, echinoids, cup corals exhibiting radial structure and coralline algae.



(b)

Figure 13. (a): Photomicrographs of a shelly biomicrite (Sample A1). Allochems present include mollusk which are gastropod (G) pelecypod (Pe), spar ite (S). Note the internal micritization of the fossils especially the mollusk, brachiopod and pelecypod. (*Mag. X40*). (b): Photomicrographs of shelly biomicrite. The allochems present include echinoid (E) pelecypods (P), foraminifera (F), broken shell fragment (X), prismatic pelecypod, spar fill brachiopod (Bs), gastropods (G) and micrite (M), sparite (S). Note the internal micritization of the fossils and spar fill echinoid. (*Mag. X40*).

Internal micritization of some grains were observed especially the mollusks, gastropods, echinoids and brachiopods. Micritic corrosion of the surface of coralline algae and pelecypods gave rise to more degenerated structures. Internal recrystallization of bioclasts is well defined. Non bioclastic components are intraclasts, quartz grains and occasional pellets.

5.3.2. Shelly Biosparmicrite

This microfacies succeeds the top biomicrite due to the increase in intraclasts, spar and bioclasts populations. It clearly define the increase in population of coralline and rod-like algae compared to others while there is an increase in population of prismatic mollusks, intraclasts shell fragments and cup algae. It shows sorting and compaction relative to the first microfacies and a decrease in internal structure of bioclasts. Internal sparitization increases in this facies while the



Figure 14. (a): Photomicrographs of sorted shelly biosparmicrite (Sample B2). Allochems present include coralline algae (CA), intraclasts (I), ooids (O), brachiopods (B), pelecypods (Pe), foraminifera (F), algae (A). Note the spar cement around coralline algae and brachiopods and internal micritization of the bioclasts. (Mag. X40). (b): A photomicrographs of shelly biosparmicrite. The allochems present include micritizedalgae (Am), algae (A), intraclasts (I), sparite (S). Note the predominance of internal micritization of thebioclasts and also the binding action of sparite within the lime mud. (Mag. X40).

spars crystal increase outward from shell boundaries, however, the compaction increases grain to grain contact.

5.3.3. Alga Biosparmicrite

This microfacies is the lowermost section of the limestone sequence in Ibese. It is synonymous to the alga bioclastic wackestone-packstone [25]. This facies is characterized by abundant stromatolitic algae which defines sediment binding. Other bioclasts include coiled gastropods, micritized echinoids, foraminifera, mollusks and spherulitic corals. Non bioclastic components include pellets and intraclasts which make about 1/3 of various portions of this microfacies. The radial pattern show internal rim cement (microspore calcite cement) within micritic ground mass. Other depositional texture that can be inferred from this layer is oncolitic grainstone defined by corals and algae complemented by increase





Figure 15. (a): Photomicrographs of alga biosparmicrite (Sample C2). The allochems present include brachiopod (B), pelecypods (Pe), intraclasts (I), gastropods (G), pellets (P) coralline algae (C). alga mat (Am). Note the internal micritization and sparitization and its binding action on the bioclasts and iron mineral infill. (*Mag. X*40). (b): Photomicrographs of alga biosparmicrite. The allochems present include Brachiopod (B), Sponges (Sp), intraclasts (I) sparitized nacreous pelecypod (nPs). Notice the internal micritization of the fossils and Iron mineral (Ir) infill also note the growth sparry calcite within the alga mat and also the rarity of micrite within the mat. (*Mag.* X40).

in the amount of skeletal debris. Grain dissolution and interparticle porosity are significant as they reflect the degree of diagenetic process, hence affecting their physical properties.

5.4. Depositional Palaeoenvironment

Based on the sedimentological indices, the shale units of Akinbo Formation are fine with no sand, silt or carbonates, grey, flaggy and slabby indicating low energy quiet water setting. The palaeobathymetry of the fossil assemblage defined from the earlier studies indicated benthic dwellers except for some shell fragments whose origin could not be ascertain [3] [20] [22] [26].

As the processes of micritization and sparitization are diagenetic processes which may involve the recrystallization of solid solutions of carbonate into coarse calcitic crystals or the mechanical degradation of shells into carbonate mud which may eventually alter the initial depositional fabric of the deposit, sequel to this, there has been decrease in the total porosity of the rocks where the topmost biomicrite defines mouldic, fenestrael, intraparticle and little interparticle porosities. The shelly biosparmicrite and alga biosparmicrite microfacies are characterized by increase in grains' components which increase the grain to grain contacts of the rock units thus enhances interparticle porosity [27].

The magnesium content of the carbonate rock is low when compared to calcium, while the average percentage composition across the wells ranges between 0.78% - 1.02%. This low composition is attributed to an increase in organism complexities (advanced phyla) based on this premise, a shallow marine environment of deposition is suggested. The presence of SiO₂, K₂O and Al₂O₃ is an indication of clastic input and degree of diagenesis on the lithofacies [21] [22] [28]. The downhole increment of SiO_2 in the formation indicates an increase in the continental influence. In well-1, the relative increase in the amount of SiO₂ defines the degree of continental influence which suggest that the paleobathymetry during the formation of the carbonate to be within the range of 12.0 m to 14.0 m depth suggesting a shallow water environment (Figure 10 and Figure 11). The average Fe/Mg in well-3 measured in the limestone (1.24) indicated gradual increase suggesting a continental influence. The moderate amount of both reflect the ferruginous syn-sedimentary and post-depositional inputs of these sediments by suface meteoric waters flushing rapidly through ferromagnesian rich minerals overlying the limestone.

6. Conclusion

Detailed mapping and logging of the study area showed the total thickness of the limestone ranged from 12.0 m to 15.0 m. The limestone consists of milky to grey and nodular to massive fossiliferous units. It is evident that the sand units increase with depth. The major element percentages revealed an increase in CaO with depth and an average of 56% and also a corresponding increase in MgO with average of 0.86% which show the calcitic nature of the deposit. The six wells showed significant peaks in MgO and SiO₂ at depth approximately 13.0 m. The increase in the Mg component suggests diagenetic processes and the significant occurrence of Al₂O₃ and Fe₂O₃ in the sequence is an indication of continental influence in the diagenetic history of the sedimentation. Three different microfacies were recognized based on depositional texture, they are shelly biomicrite, shelly biosparmicrite and alga biosparmicrite. The predominance of micrite as the cementing matrix revealed that the rock was deposited in a quiet shallow inner shelf environment occasioned by storm waves. The elemental composition of the rock units suggested the suitability of deposit for cement production however; further studies could look into the lime saturation factor (LSF), silica modulus (SR) and alumina modulus (AR) in making up for corrections to meet production specifications.

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Appendix

 Table 1. Major elemental geochemistry of the Akinbo and Ewekoro Formations in Well 1.

Depth (m)	CaO (%)	SiO ₂ (%)	Al ₂ O ₃ (%)	Fe ₂ O ₃ (%)	MgO (%)	K ₂ O (%)	SiO ₂ (%)	Mg/Ca	Fe/Ca	Si/Ca	Fe/Mg	Si/Al	Ca/Mg	Mg/Ca	Fe/Ca	Si/Ca
1.00	0.22	77.49	9.50	4.20	0.19	0.77	77.49	0.863636	19.09091	352.2273	22.10526	8.156842	1.157895	0.863636	19.09091	352.2273
1.50	0.17	65.27	14.71	6.20	0.36	0.70	65.27	2.117647	36.47059	383.9412	17.22222	4.437118	0.472222	2.117647	36.47059	383.9412
2.00	0.18	65.94	13.58	6.75	0.39	0.82	65.94	2.166667	37.50000	366.3333	17.30769	4.855670	0.461538	2.166667	37.50000	366.3333
2.50	0.15	65.25	14.22	7.12	0.4	0.88	65.25	2.666667	47.46667	435.0000	17.80000	4.588608	0.375000	2.666667	47.46667	435.0000
3.00	0.18	64.32	1.87	9.16	0.38	0.84	64.32	2.111111	50.88889	357.3333	24.10526	34.39572	0.473684	2.111111	50.88889	357.3333
3.50	0.19	62.96	15.32	5.84	0.47	0.84	62.96	2.473684	30.73684	331.3684	12.42553	4.109661	0.404255	2.473684	30.73684	331.3684
4.00	0.2	65.32	14.80	5.45	0.41	0.82	65.32	2.050000	27.25000	326.6000	13.29268	4.413514	0.487805	2.050000	27.25000	326.6000
4.50	0.21	64.19	14.20	5.13	0.38	1.54	64.19	1.809524	24.42857	305.6667	13.50000	4.520423	0.552632	1.809524	24.42857	305.6667
5.00	0.20	63.82	15.22	4.98	0.43	0.87	63.82	2.150000	24.90000	319.1000	11.58140	4.193167	0.465116	2.150000	24.90000	319.1000
5.50	51.10	4.44	0.82	0.44	1.13	0.17	4.44	0.022114	0.008611	0.086888	0.389381	5.414634	45.22124	0.022114	0.008611	0.086888
6.00	52.43	3.61	0.59	0.39	1.10	0.27	3.61	0.020980	0.007438	0.068854	0.354545	6.118644	47.66364	0.020980	0.007438	0.068854
6.55	51.80	4.94	0.55	0.40	1.44	0.08	4.94	0.027799	0.007722	0.095367	0.277778	8.981818	35.97222	0.027799	0.007722	0.095367
7.00	-	-	washout	-	-	-	-	-	-	-	-	-	-	-	-	-
7.50	52.72	3.48	0.84	0.48	0.84	0.06	3.48	0.015933	0.009105	0.066009	0.571429	4.142857	62.76190	0.015933	0.009105	0.066009
8.05	52.74	3.32	0.57	0.52	1.06	0.72	3.32	0.020099	0.009860	0.062950	0.490566	5.824561	49.75472	0.020099	0.009860	0.062950
8.50	51.66	4.82	0.87	0.52	1.11	0.07	4.82	0.021487	0.010066	0.093302	0.468468	5.540230	46.54054	0.021487	0.010066	0.093302
9.00	51.74	4.16	0.82	0.50	1.68	0.08	4.16	0.032470	0.009664	0.080402	0.297619	5.073171	30.79762	0.032470	0.009664	0.080402
9.50	52.16	3.27	0.65	0.36	106	0.07	3.27	2.032209	0.006902	0.062692	0.003396	5.030769	0.492075	2.032209	0.006902	0.062692
10.00	52.58	3.37	0.59	0.39	0.98	0.17	3.37	0.018638	0.007417	0.064093	0.397959	5.711864	53.65306	0.018638	0.007417	0.064093
10.50	52.52	3.90	0.47	0.32	0.83	0.16	3.90	0.015804	0.006093	0.074257	0.385542	8.297872	63.27711	0.015804	0.006093	0.074257
11.00	52.16	3.51	0.52	0.35	0.70	0.06	3.51	0.013420	0.006710	0.067293	0.500000	6.750000	74.51429	0.013420	0.006710	0.067293
11.50	51.80	4.36	0.85	0.42	1.18	0.12	4.36	0.022780	0.008108	0.084170	0.355932	5.129412	43.89831	0.022780	0.008108	0.084170
12.00	51.85	4.06	0.90	0.45	1.11	0.19	4.06	0.021408	0.008679	0.078303	0.405405	4.511111	46.71171	0.021408	0.008679	0.078303
12.30	-	-	washout	-	-	-	-	-	-	-	-	-	-	-	-	-
12.50	51.84	4.90	0.54	0.38	0.98	0.05	4.90	0.018904	0.007330	0.094522	0.387755	9.074074	52.89796	0.018904	0.007330	0.094522
13.00	52.56	3.75	0.45	0.44	1.01	0.05	3.75	0.019216	0.008371	0.071347	0.435644	8.333333	52.03960	0.019216	0.008371	0.071347
13.50	44.31	10.38	1.65	1.14	4.84	0.10	10.38	0.109230	0.025728	0.234259	0.235537	6.290909	9.154959	0.109230	0.025728	0.234259
14.00	53.10	2.40	0.45	0.38	1.06	0.14	2.40	0.019962	0.007156	0.045198	0.358491	5.333333	50.09434	0.019962	0.007156	0.045198
14.50	51.74	4.43	0.57	0.42	1.02	0.06	4.43	0.019714	0.008118	0.085620	0.411765	7.771930	Na	0.019714	0.008118	0.085620
15.00	52.54	3.02	0.49	0.32	0.90	0.06	3.02	0.017130	0.006091	0.057480	0.355556	6.163265	Na	0.017130	0.006091	0.057480
15.50	53.19	2.98	0.42	0.38	0.91	0.04	2.98	0.017108	0.007144	0.056026	0.417582	7.095238	Na	0.017108	0.007144	0.056026
16.00	53.14	2.47	0.48	0.40	0.72	0.05	2.47	0.013549	0.007527	0.046481	0.555556	5.145833	Na	0.013549	0.007527	0.046481
16.50	51.59	4.48	0.94	0.53	1.18	0.06	4.48	0.022873	0.010273	0.086839	0.449153	4.765957	Na	0.022873	0.010273	0.086839
17.00	52.37	3.28	0.84	0.55	0.96	0.05	3.28	0.018331	0.010502	0.062631	0.572917	3.904762	Na	0.018331	0.010502	0.062631
17.50	49.31	7.32	0.75	0.56	1.12	0.06	7.32	0.022713	0.011357	0.148449	0.500000	9.760000	Na	0.022713	0.011357	0.148449
18.00	51.7	4.13	0.48	0.42	1.15	0.14	4.13	0.022244	0.008124	0.079884	0.365217	8.604167	Na	0.022244	0.008124	0.079884
18.50	49.16	8.67	0.92	0.47	0.82	0.04	8.67	0.016680	0.009561	0.176363	0.573171	9.423913	Na	0.016680	0.009561	0.176363

Table 2. Major elemental geochemistry of the Akinbo and Ewekoro Formations in Well-2.

Depth (m)	CaO(%)	SiO ₂ (%)	$Al_2O_3(\%)$	Fe ₂ O ₃ (%)	MgO(%)	K ₂ O(%)	Mg/Ca	Fe/Ca	Si/Ca	Fe/Mg	Si/Al	Ca/Mg
0.50	0.16	72.36	10.16	5.38	0.15	0.10	0.937500	33.62500	452.2500	35.86667	7.122047	1.066667
1.00	0.14	6.77	15.68	6.65	0.40	0.80	2.857143	47.50000	48.35714	16.62500	0.431760	0.350000
1.50	0.15	62.85	14.76	5.57	0.19	0.29	1.266667	37.13333	419.0000	29.31579	4.258130	0.789474
2.00	0.13	63.82	14.72	6.48	0.31	0.62	2.384615	49.84615	490.9231	20.90323	4.335598	0.419355
2.50	0.12	62.94	15.37	5.94	0.37	0.76	3.083333	49.50000	524.5000	16.05405	4.094990	0.324324
3.00	0.14	60.85	14.13	5.30	0.26	0.39	1.857143	37.85714	434.6429	20.38462	4.306440	0.538462
3.50	0.14	62.86	14.12	7.24	0.36	0.79	2.571429	51.71429	449.0000	20.11111	4.451841	0.388889
4.00	0.19	63.75	14.45	4.98	0.47	0.90	2.473684	26.21053	335.5263	10.59574	4.411765	0.404255
4.50	0.17	62.93	14.10	6.18	0.14	0.79	0.823529	36.35294	370.1765	44.14286	4.463121	1.214286
5.00	0.19	65.27	16.26	5.30	0.47	0.94	2.473684	27.89474	343.5263	11.27660	4.014145	0.404255
5.50	0.18	63.79	15.15	5.18	0.38	0.92	2.111111	28.77778	354.3889	13.63158	4.210561	0.473684
6.00	0.16	67.62	11.14	5.30	0.35	0.35	2.187500	33.12500	422.6250	15.14286	6.070018	0.457143
6.50	0.19	62.90	15.92	5.17	0.39	0.97	2.052632	27.21053	331.0526	13.25641	3.951005	0.487179
6.90	0.30	65.72	13.47	5.31	0.29	1.79	0.966667	17.70000	219.0667	18.31034	4.878990	1.034483
7.50	49.43	8.12	2.27	1.27	1.15	0.13	0.023265	0.025693	0.164273	1.104348	3.577093	42.98261
8.00	52.45	2.76	0.49	0.35	0.67	0.09	0.012774	0.006673	0.052622	0.522388	5.632653	78.28358
8.50	53.31	2.15	0.35	0.22	0.50	0.06	0.009379	0.004127	0.040330	0.440000	6.142857	106.6200
9.00	53.35	2.74	0.45	0.35	0.52	0.06	0.009747	0.006560	0.051359	0.673077	6.088889	102.5962
9.50	49.63	6.22	0.97	0.65	0.77	0.05	0.015515	0.013097	0.125327	0.844156	6.412371	64.45455
10.00	51.87	4.80	0.45	0.37	0.66	0.06	0.012724	0.007133	0.092539	0.560606	10.66667	78.59091
10.50	53.55	2.32	0.39	0.32	0.75	0.05	0.014006	0.005976	0.043324	0.426667	5.948718	71.40000
11.00	52.25	3.80	0.47	0.38	0.65	0.06	0.012440	0.007273	0.072727	0.584615	8.085106	80.38462
11.50	53.34	2.30	0.47	0.32	0.67	0.05	0.012561	0.005999	0.043120	0.477612	4.893617	79.61194
12.00	52.48	3.26	0.38	0.28	0.52	0.04	0.009909	0.005335	0.062119	0.538462	8.578947	100.9231
12.50	53.49	2.16	0.42	0.25	0.49	0.05	0.009161	0.004674	0.040381	0.510204	5.142857	109.1633
13.00	53.44	2.90	0.44	0.36	0.56	0.38	0.010479	0.006737	0.054266	0.642857	6.590909	95.42857
13.50	52.57	3.67	0.45	0.34	0.52	0.05	0.009892	0.006468	0.069812	0.653846	8.155556	101.0962
14.00	53.88	2.40	0.73	0.35	0.68	0.06	0.012621	0.006496	0.044543	0.514706	3.287671	79.23529
14.60	51.90	4.25	0.60	0.40	1.02	0.07	0.019653	0.007707	0.081888	0.392157	7.083333	50.88235
15.00	-	-	washout	-	-	-	-	-	-	-	-	-
15.50	48.67	7.40	0.80	0.59	2.52	0.09	0.051777	0.012122	0.152044	0.234127	9.250000	19.31349
16.00	51.33	4.29	0.45	0.37	1.32	0.08	0.025716	0.007208	0.083577	0.280303	9.533333	38.88636
16.50	54.39	1.36	0.58	0.47	0.95	0.06	0.017466	0.008641	0.025005	0.494737	2.344828	57.25263
17.00	53.99	2.19	0.52	0.32	0.76	0.06	0.014077	0.005927	0.040563	0.421053	4.211538	71.03947
17.50	53.23	2.59	0.56	0.49	0.57	0.05	0.010708	0.009205	0.048657	0.859649	4.625000	93.38596
18.00	51.18	4.33	0.83	0.44	1.10	0.07	0.021493	0.008597	0.084603	0.400000	5.216867	46.52727
18.60	51.99	4.44	0.68	0.39	1.28	0.03	0.024620	0.007501	0.085401	0.304688	6.529412	40.61719
19.50	49.23	8.44	0.52	0.45	1.23	0.13	0.024985	0.009141	0.171440	0.365854	16.23077	40.02439
20.00	47.13	10.24	1.54	1.20	1.22	0.12	0.025886	0.025461	0.217271	0.983607	6.649351	38.63115
20.50	48.94	7.52	0.82	0.67	0.71	0.06	0.014508	0.013690	0.153658	0.943662	9.170732	68.92958



Depth (m)	CaO(%)	SiO ₂ (%)	$Al_2O_3(\%)$	Fe ₂ O ₃ (%)	MgO(%)	K ₂ O(%)	Mg/Ca	Fe/Ca	Si/Ca	Fe/Mg	Si/Al	Ca/Mg
1.00	0.18	72.79	10.65	6.25	0.19	0.12	1.055556	34.72222	404.3889	32.89474	6.834742	0.947368
1.50	0.13	63.42	12.89	9.27	0.28	0.55	2.153846	71.30769	487.8462	33.10714	4.920093	0.464286
2.00	0.1	64.42	12.19	11.14	0.31	0.71	3.100000	111.4000	644.2000	35.93548	5.284660	0.322581
2.50	0.14	62.57	12.86	8.17	0.33	0.75	2.357143	58.35714	446.9286	24.75758	4.865474	0.424242
3.00	0.16	63.92	11.14	7.32	0.29	0.71	1.812500	45.75000	399.5000	25.24138	5.737882	0.551724
3.50	0.44	65.95	12.9	4.26	0.21	1.38	0.477273	9.681818	149.8864	20.28571	5.112403	2.095238
4.00	0.28	64.52	10.57	7.44	0.2	1.27	0.714286	26.57143	230.4286	37.20000	6.104068	1.400000
4.50	0.12	63.94	11.27	8.57	0.27	1.93	2.250000	71.41667	532.8333	31.74074	5.673469	0.444444
5.00	0.11	62.67	13.32	8.36	0.26	2.79	2.363636	76.00000	569.7273	32.15385	4.704955	0.423077
5.50	52.4	3.30	0.75	0.60	0.57	0.06	0.010878	0.011450	0.062977	1.052632	4.400000	91.92982
6.00	-	-	-	washout	-	-	-	-	-	-	-	-
6.50	40.21	18.2	4.22	2.72	1.13	0.04	0.028102	0.067645	0.452624	2.407080	4.312796	35.58407
7.00	52.69	3.76	0.84	0.46	0.56	0.07	0.010628	0.00873	0.071361	0.821429	4.476190	94.08929
7.50	52.12	3.68	0.65	0.54	0.48	0.06	0.00921	0.010361	0.070606	1.125000	5.661538	108.5833
8.00	53.74	2.5	0.34	0.29	0.38	0.06	0.007071	0.005396	0.046520	0.763158	7.352941	141.4211
8.50	52.44	3.34	0.52	0.42	0.32	0.05	0.006102	0.008009	0.063692	1.312500	6.423077	163.8750
9.00	54.19	1.43	0.34	0.28	0.40	0.08	0.007381	0.005167	0.026389	0.700000	4.205882	135.4750
9.50	53.37	2.17	0.59	0.36	0.54	0.09	0.010118	0.006745	0.040660	0.666667	3.677966	98.83333
10.00	52.22	3.55	0.75	0.55	0.52	0.05	0.009958	0.010532	0.067982	1.057692	4.733333	100.4231
10.50	-	-	-	washout	-	-	-	-	-	-	-	-
11.00	53.94	2.27	0.56	0.32	0.32	0.06	0.005933	0.005933	0.042084	1.000000	4.053571	168.5625
11.50	54.15	1.25	0.45	0.32	0.36	0.16	0.006648	0.005910	0.023084	0.888889	2.777778	150.4167
12.00	53.94	2.66	0.44	0.21	0.49	0.06	0.009084	0.003893	0.049314	0.428571	6.045455	110.0816
12.50	53.72	2.41	0.2	0.36	0.51	0.11	0.009494	0.006701	0.044862	0.705882	12.05000	105.3333
13.00	53.51	2.82	0.45	0.32	0.64	0.04	0.011960	0.005980	0.052700	0.500000	6.266667	83.60938
13.50	52.32	3.63	0.58	0.45	0.92	0.04	0.017584	0.008601	0.069381	0.489130	6.258621	56.86957
13.70	48.42	7.69	1.55	1.03	3.71	0.06	0.076621	0.021272	0.158819	0.277628	4.961290	13.05121
14.20	-	-	-	washout	-	-	-	-	-	-	-	-
14.50	48.42	7.69	1.55	1.03	3.71	0.06	0.076621	0.021272	0.158819	0.277628	4.961290	13.05121
15.00	48.84	5.32	1.85	1.10	2.36	0.06	0.048321	0.022523	0.108927	0.466102	2.875676	20.69492
15.50	51.75	4.37	0.87	0.45	1.32	0.05	0.025507	0.008696	0.084444	0.340909	5.022989	39.20455
16.00	51.96	4.03	1.48	1.20	1.12	0.08	0.021555	0.023095	0.077560	1.071429	2.722973	46.39286
16.60	48.73	6.73	1.48	1.20	1.12	0.08	0.022984	0.024625	0.138108	1.071429	4.547297	43.50893
17.70	-	-	-	washout	-	-	-	-	-	-	-	-
18.00	48.87	8.13	0.92	0.68	1.07	0.04	0.021895	0.013914	0.166360	0.635514	8.836957	45.67290
18.50	49.37	8.24	0.84	0.74	0.82	0.05	0.016609	0.014989	0.166903	0.902439	9.809524	60.20732
19.00	48.85	8.22	0.82	0.58	0.49	0.06	0.010031	0.011873	0.168270	1.183673	10.02439	99.69388

 Table 3. Major elemental geochemistry of the Akinbo and Ewekoro Formation in Well-3.

Table 4.	Major elemental	geochemistry	of the Akinbo	and Ewekoro	Formations in Well-4.

Depth (m)	CaO(%)	SiO ₂ (%)	Al ₂ O ₃ (%)	Fe ₂ O ₃ (%)	MgO(%)	K ₂ O(%)	Mg/Ca	Fe/Ca	Si/Ca	Fe/Mg	Si/Al	Ca/Mg
1.00	0.13	72.7	10.34	5.16	0.18	0.19	1.384615	39.66154	559.2308	28.64444	7.030948	0.722222
1.50	0.18	72.31	11.31	5.31	0.16	0.24	0.888889	29.50000	401.7222	33.1875	6.393457	1.125000
2.00	0.19	65.83	12.81	8.35	0.30	4.68	1.578947	43.94737	346.4737	27.83333	5.138954	0.633333
2.50	0.16	64.25	12.75	10.23	0.27	3.44	1.687500	63.93750	401.5625	37.88889	5.039216	0.592593
3.00	51.26	4.15	0.82	0.62	0.69	0.11	0.013461	0.012095	0.080960	0.898551	5.060976	74.28986
3.30	53.73	2.24	0.55	0.34	0.51	0.06	0.009492	0.006328	0.041690	0.666667	4.072727	105.3529
4.00	52.84	3.44	0.49	0.38	0.54	0.65	0.010220	0.007192	0.065102	0.703704	7.020408	97.85185
4.50	53.49	2.31	0.65	0.53	0.51	0.06	0.009534	0.009908	0.043186	1.039216	3.553846	104.8824
5.00	54.68	1.47	0.56	0.45	0.50	0.08	0.009144	0.008230	0.026884	0.900000	2.625000	109.3600
5.50	54.39	1.45	0.49	0.28	0.34	0.05	0.006251	0.005148	0.026659	0.823529	2.959184	159.9706
6.00	53.63	2.60	0.52	0.35	0.35	0.13	0.006526	0.006526	0.048480	1.000000	5.000000	153.2286
6.50	53.84	2.49	0.64	0.52	0.34	0.07	0.006315	0.009658	0.046248	1.529412	3.890625	158.3529
7.00	53.17	2.94	0.52	0.40	0.25	0.08	0.004702	0.007523	0.055294	1.600000	5.653846	212.6800
7.50	54.62	1.69	0.40	0.24	0.29	0.06	0.005309	0.004394	0.030941	0.827586	4.225000	188.3448
8.00	54.36	1.59	0.45	0.35	0.38	0.05	0.006990	0.006439	0.029249	0.921053	3.533333	143.0526
8.50	48.11	7.17	1.95	0.86	0.97	0.11	0.020162	0.017876	0.149033	0.886598	3.676923	49.59794
9.00	48.35	6.75	1.65	1.00	1.20	0.21	0.024819	0.020683	0.139607	0.833333	4.090909	40.29167
9.50	51.40	4.31	0.55	0.42	1.37	0.09	0.026654	0.008171	0.083852	0.306569	7.836364	37.51825
10.00	52.29	3.34	0.72	0.65	0.89	0.05	0.017020	0.012431	0.063875	0.730337	4.638889	58.75281
10.50	52.79	3.45	0.60	0.44	0.88	0.05	0.016670	0.008335	0.065353	0.500000	5.750000	59.98864
11.00	53.12	2.93	0.67	0.47	0.92	0.05	0.017319	0.008848	0.055158	0.510870	4.373134	57.73913
11.50	53.19	2.30	0.57	0.45	0.68	0.05	0.012784	0.008460	0.043241	0.661765	4.035088	78.22059
12.00	52.37	3.21	0.54	0.38	0.86	0.05	0.016422	0.007256	0.061295	0.441860	5.944444	60.89535
12.50	52.24	3.62	0.54	0.42	0.87	0.08	0.016654	0.008040	0.069296	0.482759	6.703704	60.04598
13.00	51.69	4.24	0.62	0.45	0.91	0.04	0.017605	0.008706	0.082027	0.494505	6.838710	56.80220
13.50	51.33	4.50	0.92	0.82	0.8	0.05	0.015585	0.015975	0.087668	1.025000	4.891304	64.16250
14.00	51.92	4.32	0.82	0.62	0.77	0.04	0.014831	0.011941	0.083205	0.805195	5.268293	67.42857
14.50	50.05	5.36	0.96	0.85	1.03	0.05	0.020579	0.016983	0.107093	0.825243	5.583333	48.59223
15.00	48.21	8.26	1.27	0.85	0.98	0.05	0.020328	0.017631	0.171334	0.867347	6.503937	49.19388
15.50	46.48	10.56	2.54	1.24	1.11	0.12	0.023881	0.026678	0.227194	1.117117	4.157480	41.87387



Table 5. Major elemental geochemistry of the Akinbo and Ewekoro Formations in Well-5.

Depth (m)	CaO(%)	SiO ₂ (%)	Al ₂ O ₃ (%)	Fe ₂ O ₃ (%)	MgO(%)	K ₂ O(%)	Mg/Ca	Fe/Ca	Si/Ca	Fe/Mg	Si/Al	Ca/Mg
1.00	0.14	30.47	8.62	4.34	0.29	0.54	2.071429	31.00000	217.6429	14.96552	3.534803	0.482759
1.50	0.18	70.77	12.37	6.04	0.22	0.17	1.222222	33.55556	393.1667	27.45455	5.721099	0.818182
2.00	0.10	68.47	1362	7.14	0.29	0.24	2.900000	71.40000	684.7000	24.62069	0.050272	0.344828
2.55	0.15	64.84	12.6	8.30	0.17	0.22	1.133333	55.33333	432.2667	48.82353	5.146032	0.882353
3.00	54.10	1.45	0.49	0.35	0.44	0.06	0.008133	0.006470	0.026802	0.795455	2.959184	122.9545
3.50	53.91	2.80	0.56	0.39	0.43	0.08	0.007976	0.007234	0.051938	0.906977	5.000000	125.3721
4.00	53.22	2.16	0.54	0.43	0.45	0.16	0.008455	0.008080	0.040586	0.955556	4.000000	118.2667
4.40	-	-	-	washout	-	-	-	-	-	-	-	
4.50	53.22	2.16	0.54	0.43	0.45	0.16	0.008455	0.008080	0.040586	0.955556	4.000000	118.2667
5.00	54.36	4.31	0.48	0.38	0.38	0.06	0.006990	0.006990	0.079286	1.000000	8.979167	143.0526
5.50	52.94	3.46	0.75	0.47	0.48	0.05	0.009067	0.008878	0.065357	0.979167	4.613333	110.2917
6.00	52.46	3.37	0.64	0.54	0.59	0.08	0.011247	0.010294	0.064239	0.915254	5.265625	88.91525
6.50	51.70	4.32	0.62	0.45	0.96	0.09	0.018569	0.008704	0.083559	0.468750	6.967742	53.85417
7.50	49.39	6.60	0.88	0.45	1.14	0.08	0.023082	0.009111	0.133630	0.394737	7.500000	43.32456
8.00	52.68	3.34	0.79	0.44	0.69	0.05	0.013098	0.008352	0.063402	0.637681	4.227848	76.34783
8.50	52.44	3.20	0.71	0.48	0.83	0.06	0.015828	0.009153	0.061022	0.578313	4.507042	63.18072
9.00	53.57	2.65	0.55	0.43	0.57	0.04	0.010640	0.008027	0.049468	0.754386	4.818182	93.98246
9.50	51.8	4.44	0.84	0.65	0.63	0.06	0.012162	0.012548	0.085714	1.031746	5.285714	82.22222
10.00	51.96	4.80	0.65	0.35	0.90	0.05	0.017321	0.006736	0.092379	0.388889	7.384615	57.73333
10.50	52.8	3.40	0.84	0.56	0.73	0.04	0.013826	0.010606	0.064394	0.767123	4.047619	72.32877
11.00	51.50	4.90	0.82	0.65	0.32	0.14	0.006214	0.012621	0.095146	2.031250	5.975610	160.9375
11.50	52.79	3.67	0.68	0.54	0.39	0.12	0.007388	0.010229	0.069521	1.384615	5.397059	135.3590
12.00	51.47	4.73	0.73	0.48	1.29	0.15	0.025063	0.009326	0.091898	0.372093	6.479452	39.89922
12.50	43.46	15.2	3.75	2.10	0.29	0.07	0.006673	0.048320	0.349747	7.241379	4.053333	149.8621
13.00	51.62	4.32	0.62	0.50	0.30	0.13	0.005812	0.009686	0.083688	1.666667	6.967742	172.0667
13.50	37.95	21.4	3.25	2.75	1.45	0.20	0.038208	0.072464	0.563900	1.896552	6.584615	26.17241
14.00	51.36	4.73	0.96	0.84	0.62	0.15	0.012072	0.016355	0.092095	1.354839	4.927083	82.83871
14.50	50.81	5.78	0.68	0.43	0.35	0.22	0.006888	0.008463	0.113757	1.228571	8.500000	145.1714

Depth (m)	CaO(%)	SiO ₂ (%)	Al ₂ O ₃ (%)	Fe ₂ O ₃ (%)	MgO(%)	K ₂ O(%)	Mg/Ca	Fe/Ca	Si/Ca	Fe/Mg	Si/Al	Ca/Mg
1.00	0.16	69.21	12.45	8.88	0.15	0.14	0.937500	55.50000	432.5625	59.20000	5.559036	1.066667
1.50	0.14	70.5	10.81	6.26	0.17	0.10	1.214286	44.71429	503.5714	36.82353	6.521739	0.823529
2.00	0.10	60.75	15.6	9.63	0.18	0.29	1.800000	96.30000	607.5000	53.50000	3.894231	0.555556
2.50	0.11	61.78	16.22	9.02	0.19	0.30	1.727273	8200000	561.6364	47.47368	3.808878	0.578947
3.00	54.13	1.29	0.49	0.34	0.37	0.06	0.006835	0.006281	0.023832	0.918919	2.632653	146.2973
3.10	54.12	1.82	0.38	0.29	0.25	0.06	0.004619	0.005358	0.033629	1.160000	4.789474	216.4800
4.00	54.12	1.82	0.38	0.29	0.25	0.06	0.004619	0.005358	0.033629	1.160000	4.789474	216.4800
4.50	53.40	2.90	0.42	0.37	0.67	0.13	0.012547	0.006929	0.054307	0.552239	6.904762	79.70149
5.00	41.63	19.00	1.98	1.39	0.78	0.08	0.018736	0.033389	0.456402	1.782051	9.595960	53.37179
5.50	54.24	1.42	0.42	0.25	0.33	0.04	0.006084	0.004609	0.026180	0.757576	3.380952	164.3636
6.00	51.40	4.37	0.62	0.42	0.6	0.12	0.011673	0.008171	0.085019	0.700000	7.048387	85.66667
6.50	0.23	64.84	11.5	5.88	0.14	0.08	0.608696	25.56522	281.9130	42.00000	5.638261	1.642857
7.10	51.12	4.27	0.59	0.42	0.22	0.06	0.004304	0.008216	0.083529	1.909091	7.237288	232.3636
8.50	53.34	2.25	0.38	0.25	0.71	0.35	0.013311	0.004687	0.042182	0.352113	5.921053	75.12676
9.50	52.72	3.56	0.54	0.42	0.62	0.05	0.011760	0.007967	0.067527	0.677419	6.592593	85.03226
10.00	53.12	2.25	0.63	0.54	0.66	0.10	0.012425	0.010166	0.042357	0.818182	3.571429	80.48485
10.50	51.24	4.21	0.52	0.44	0.96	0.15	0.018735	0.008587	0.082162	0.458333	8.096154	53.37500
11.00	52.67	3.2	0.64	0.46	0.82	0.06	0.015569	0.008734	0.060756	0.560976	5.000000	64.23171
11.40	52.85	3.71	0.8	0.47	0.42	0.08	0.007947	0.008893	0.070199	1.119048	4.637500	125.8333
12.00	52.02	3.14	0.65	0.52	0.69	0.15	0.013264	0.009996	0.060361	0.753623	4.830769	75.39130
12.50	51.10	4.19	0.74	0.46	1.2	0.14	0.023483	0.009002	0.081996	0.383333	5.662162	42.58333
13.00	51.25	4.25	0.48	0.37	0.92	0.05	0.017951	0.007220	0.082927	0.402174	8.854167	55.70652
13.50	51.20	4.71	0.62	0.38	0.72	0.06	0.014063	0.007422	0.091992	0.527778	7.596774	71.11111
14.00	50.61	5.98	0.56	0.44	0.6	0.13	0.011855	0.008694	0.118158	0.733333	10.67857	84.35000
14.50	43.72	18.18	0.79	0.63	0.63	0.26	0.014410	0.014410	0.415828	1.000000	23.01266	69.39683
15.00	53.29	2.27	0.36	0.24	0.82	0.07	0.015388	0.004504	0.042597	0.292683	6.305556	64.98780
15.50	48.12	11.7	0.58	0.41	0.68	0.18	0.014131	0.008520	0.243142	0.602941	20.17241	70.76471
16.00	48.14	9.34	0.57	0.33	0.53	0.28	0.011010	0.006855	0.194017	0.622642	16.38596	90.83019

Table 6. Major elemental geochemistry of the Akinbo and Ewekoro Formations in Well-6.



Sample	Depth	Allochems (and% C	Smarrita	Miguito	Summant			
No	(m)	Bioclasts	Intraclasts	Pellets	Ooids	Sparite	Micrite	Support
A2	12.0	Sponge spicules, pelecypods, algae, echinoids, shell fragments, gastropods (48%)	Intraclasts (2%)	Few (2%)	Few (2%)	Present as infill within bioclasts, (6%)	Present as matrix, (40%)	Micrite
B2	15.0	Coralline algae, gastropods, algae, corals, pelecypods, foraminifera, shell fragments, sponge spicules, (35%)	Present (15%)	Present (3%)	Few (2%)	Present as infill and matrix components, (10%)	Present as matrix, (35%)	Micrite
C2	19.0	Gastropods foraminifera, algae mat, echinoid, shell fragments, sponge spicules, (36%)	Present (10%)	Present (6%)	Few (2%)	Present as infill and matrix components, (6%)	Present as matrix, (40%)	Micrite

Table 7. Summary of carbonate petrography of the limestone.

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