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Feasibility Study of Melon Seed Oil as a Source of Biodiesel

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Abstract

Melon seed oils were extracted at a very high yield of 52.2%. The extracted oil was subjected to oil quality tests and subsequently transesterified to give fatty acid methyl esters or biodiesel. The biodiesel was also subjected to fuel quality tests. The results showed that the extracted oil had specific gravity of 0.91 and moisture content of 0.90% indicating that the oil is a very good energy source, a good candidate for transesterification and will not be easily susceptible to microbial attack and autooxidation. The fuel quality parameters of the produced biodiesel showed that it conforms to standards for biodiesel and compares well with a standard petrodiesel.

Keywords

Biodiesel, Melon, Petrodiesel, Transesterification

1. Introduction

The need for alternatives to fossil fuel sources has become more obvious today than ever and the reasons are not far-fetched. They include increasing cost of fossil fuels, their unrenewable or closed carbon cycle nature, emissions of combustion-generated pollutants, etc [1]. The alternatives are found in biofuels. Among the different biofuels, biodiesel seems to be the most studied and the most promising competitor to the currently used petrodiesel.

Biodiesel has been described as a fuel composed of monoalkyl esters of long-chain fatty acids derived from renewable vegetable oils and fats [2]. Biodiesel production apart from providing a good alternative to petrodiesel, will encourage the market for increased production of vegetable oils and animal fats, reduce global warming

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and improve the lubricating properties of diesel engines [3] [4].

Chemically, biodiesels are produced from the transesterification of triglycerides found in vegetables oils and fats to form the monoalkyl esters which are the primary molecules of biodiesel [5] [6].

The formation of the methyl esters (ME) involves a step-wise reaction with intermediate formation of digly-cerides (DG) and monoglycerides (MG) from the starting triglyceride (TG) with the production of glycerol (GL) in the final step.

$$TG + ROH \rightleftharpoons DG + R'CO_{2}R$$
 (1)

$$DG + ROH \rightleftharpoons MG + R'CO_{2}R$$
 (2)

$$MG + ROH \rightleftharpoons GL + R'CO_2R$$
 (3)

The overall reaction is therefore given as:

$$TG + 3ROH \rightleftharpoons 3R'CO_2R + GL$$

Of course the reaction is catalyzed by either a base or an acid [7]-[9].

The hydrocarbon moiety of the methyl esters are of comparable carbon number with that of the petrodiesel. Also the cetane number, energy content, viscosity and phase changes of biodiesel are akin to that of petrodiesels [10].

The vegetable oils that have been utilized as sources of biodiesel include those from rape seed, cotton seed, soybean seed, fluted pumpkin seed, sunflower seed, palm kernel seed and fruit, corn, linseed etc. [11] [12].

Melon seed (*Citrullus lanatus. thumb. mausf*) oil provides another useful source of biodiesel production. The melon seed (*Citrullus lanatus. thumb. mausf*) popularly known as egusi in Nigeria is from the family cucurbifaceae and mainly cultivated as a soup thickener and snack. The seed is rich in oil, low in cholesterol and contains essential and unsaturated fatty acids [13] [14].

The Local utilization of this seed in biodiesel production will no doubt improve its production/market, generate employment and may earn Nigeria the much needed foreign exchange.

2. Materials and Method

2.1. Oils Extraction

The melon seed samples used for this study were obtained from Ariaria market, Aba in Abia State, Nigeria. The samples were air-dried, weighed, ground and extracted in a soxhlet extractor using hexane as the solvent. The extracted oil was subsequently analyzed for some oil quality parameters as shown in **Table 1**.

2.2. Biodiesel Production

50 ml of methanol was added to 3.0 g of potassium hydroxide KOH in a 100 ml conical flask. The mixture was stirred with gentle heating. The KOH/methanol mixture was subsequently added to 150 ml of the melon seed oil in 1000 ml conical flask and stirred gently for more than 10 hrs. The entire mixture was then transferred to a separatory funnel and allowed to stand overnight to give two distinct layers of lower glycerol and upper methyl esters (biodiesel). The separated biodiesel was subsequently washed severally with water and subjected to fuel

Table 1. Oil quality parameter of oil extracted from melon seed.

PARAMETERS	VALUE
Acid value (mg KOH/g)	4.18
Iodine value (g/100g)	121.65
Peroxide value (meq/kg)	16.07
Specific gravity (g/ml)	0.91
Kinematic viscosity (mm ² /s)	21.65
Moisture content (%)	0.90
Oil yield (%)	52.20
Color	Yellow

quality tests to determine its suitability or otherwise, as an alternative to petro-diesel. The results are presented in Table 2.

3. Result and Discussion

The result of the oil quality parameters of the extracted melon seed oil is presented in **Table 1**.

The result of the oil quality parameters shows that melon seed oil could serve as an economic feedstock for biodiesel production given the percentage oil yield of 52.2. The iodine value is not too high showing that biodiesel obtained from melon seed oil may not be too susceptible to oxidation and quick rancidity.

Given the threshold acid value of 1.0 mg KOH/g for fresh oils, it is obvious that melon seed oil (acid value 4.18) will have catalysis problem during esterification (catalyst deactivation) and formation of soaps as side reaction. These problems can be taken care of with the use of heterogeneous catalysts or the addition of bases like sodium hydroxide.

The specific gravity of 0.91 shows that the oil is a good energy source. The very low moisture content of the melon seed oil of 0.90 is an indication that hydrolysis of the produced esters during esterification and its consequent soap formation is not likely to occur if melon seed oil is used in transesterification reaction. The low moisture content will also prevent microbial attack and autoxidation which leads to oil rancidity.

The result of the fuel quality parameters of the melon seed oil methyl esters are shown in Table 2.

It is obvious that transesterification has drastically reduced the moisture content and acid value of melon seed oil giving the produced biodiesel better fuel characteristics. There is also a sharp reduction of the kinematic viscosity of the melon seed oil from 21.65 to about 5.8 mm²/s in the biodiesel, a value that is comparable to that of the petrodiesel.

The suitability of the produced methyl esters as a biodiesel is better appreciated when these parameters are compared with that of a standard petrodiesel (ASTM D975) and a standard biodiesel (ASTM D6751) as shown in **Table 3**.

The specific gravity of the produced biodiesel compares well with that of the standard biodiesel (D6751) and

Table 2. Fuel quality parameters of methyl esters produced from melon seed oil.

PARAMETERS	VALUE
Specific gravity (g/ml)	0.86
Kinematic viscosity (mm ² /s)	5.80
Flash Point (°C)	132.00
Diesel Index	46.60
Iodine Value (mg KOH/g)	120.40
Acid Value (mg KOH/g)	0.26
Peroxide value (meq O ₂ /kg)	51.12
Heat of combustion (mJ/kg)	37.36
Moisture Content (%)	0.04
Color	Light Yellow

Table 3. Fuel quality parameters of a standard biodiesel, a petrodiesel and melon seed oil methyl esters.

PARAMETERS	ASTM D975 (PETRODIESEL)	ASTM D6751 (BIODIESEL)	MELON SEED OIL METHYL ESTERS
Specific gravity (g/ml)	0.85	0.88	0.86
Kinematic Viscosity (mm ² /s)	1.9 - 4.1	1.9 - 6.0	5.8
Flash Point (°C)	60 - 80	100 - 170	132.0
Diesel Index	40 - 55	47 - 60	46.0
Acid Value		0.80	0.20
Heat of combustion (mJ/kg)	44.80	37.80	37.36
Moisture Content (%)	0.05	0.05	0.04

petrodiesel (D975). The flash point is higher than that of a standard biodiesel making it safer for storage and transportation. The acid value and moisture content is highly reduced and the diesel index similar to that of the standard biodiesel and petrodiesel.

4. Conclusions

The high percentage oil yield of melon seed shows that it is a viable oil source. The oil quality parameters show that the oil from melon seed with a moderate degree of unsaturation is not susceptible to oxidative rancidity. The low moisture content makes it a good candidate for transesterification reaction.

The fuel quality parameters of the methyl esters derived from the melon seed oil show that it conforms to standards for biodiesel and compares very well with a standard petrodiesel. So melon seed oil can serve as a good feedstock for biodiesel production.

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