

Chemometric Evaluation of the Heavy Metals in Urban Soil of Fallujah City, Iraq

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Abstract

To evaluate the concentrations of heavy metals and identification of their sources, 20 composite soil samples were collected randomly in Fallujah city, Iraq. To assess the pollution level, we used the soil metal index (SMI) and enrichment factor. We investigated the sources of heavy metal using the chemometric techniques such as correlation analysis, principal component analysis (PCA) and cluster analysis (CA). The concentrations of Pb and Co have a normal distribution in soil of the study area. Zn and Cd have an approximately normal distribution, and Ni, Cu, and Cr have a non-normal distribution. Our results revealed that the urban soils in Fallujah city were unpolluted by heavy metals and showed a significant to extremely high enrichments of heavy metals controlled by anthropogenic activities. The moderate to strong positive correlations among Cd, Co, Cu, Cr, Ni, Pb, and Zn suggest that these metals have common sources. The chemometric techniques identify that the source of Cu, Co and Pb is controlled by lithogenic origin and that the source of Cr, Ni and Cd is controlled by mixed sources (lithogenic and anthropogenic).

Keywords

Chemometric, Heavy Metal, Urban Soil, Iraq

1. Introduction

Urban soil is a soil material having a non-agricultural, man-made surface layer more than 50 cm thick that has been produced by mixing, filling, or by contamination of land surfaces in urban and suburban areas [1]. Urban soils have peculiar characteristics, such as vertical and spatial variability, unpredictable layering, poor structure, low organic matter content, high content of anthropogenic materials, high concentrations of heavy metals, and

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modified soil organisms population and activity [2]-[4]. The urban soil is highly modified by human activities like terrestrials and subterranean constructions, infrastructures reconstructions, the maintenance of parks and green zones, in which the local soil is mixed with soils from other locations [5] [6]. Heavy metals in urban soils have been shown to be very useful tracers of environmental pollution [7]. As heavy metals are non-biodegradable and accumulative in nature, the elevated metal emissions and their deposition over time can lead to anomalous enrichment and the contamination of the surface environment [8]. The most important sources of metal pollution are mine tailings, smelter emissions, waste incineration, atmospheric deposition and traffic emissions [9]-[11]. Due to proximity to humans, accumulation of harmful substances in urban soils is of great concern [12] [13]. Studies on heavy metals in urban soils were carried out in different regions around the world [14]-[24].

Chemometric techniques widely applied in atmospheric studies for identifying sources of pollutants or for determining the importance of contaminant source contributions to a particular site, have been little use in soil studies, but may be effectively employed in such investigations [25]. Chemometric techniques such as correlation analysis, principal component analysis (PCA) and cluster analysis (CA) are employed to identify sources contributing to observing heavy metal pollution of soils [26]-[33].

The aim of this study is to apply the chemometric techniques of correlation analysis, principal component analysis (PCA), and cluster analysis (CA) to evaluate the heavy metal content of urban soil in Fallujah city and identify sources of heavy metal pollution.

2. Materials and Methods

Fallujah city located roughly 69 km west of Baghdad on the Euphrates River, 33°21'13"N, 43°46'46"E (Figure 1). The city grew from a small town in 1947 to a population of 326,471 inhabitants in 2010. The region has been inhabited for many millennia. There is evidence that the area surrounding Fallujah is inhabited in Babylonian times. Geologically, Fallujah city is situated on the quaternary unconsolidated sediment of the Mesopotamian plain formed mainly from river sediments (sand, silt and clay).

The sampling sites were randomly distributed in the study area (Figure 1). The soil (0 - 0.9 m) was collected. Each of the soil samples consisted of 3 subsamples obtained in different depths (0 - 0.3 m, 0.3 - 0.6 m and 0.6 - 0.9 m) using a hand auger. Locations of the sampling sites were recorded using Garmon 72 GPS. For each sampling site, a composite sample was made by mixing the three subsamples. The soil samples were kept in plastic bags. They were oven-dried in the laboratory at 105°C for 24 h, sieved through a 2-mm stainless steel sieve to remove large debris, gravel-sized materials, plant roots and other waste materials and they were homogenized with porcelain pestle and mortar. Then these samples passed through a 2-mm stainless steel sieve and stored in closed plastic bags until analysis. Soil samples were digested with a 3:2:2 mixture of HNO₃-H₂SO₄-HCl [34]. The digested solutions were analyzed by Atomic Absorption Spectrometry (AAS Phoenix-986). All of the soil samples were analyzed for their total concentrations of Cu, Zn, Ni, Pb, Fe, Cd, Co and Cr. The absolute concentrations of heavy metals in the urban soil of the study area were listed in Table 1.

The soil pollution level and its variation along the sites are determined using soil metal index (SMI). This index is a quick tool to compare the pollution status of different sites. For calculation the SMI for each sampling site, the ratio of heavy metal concentration in the soil sample to the background concentration of the metal is calculated first, for each metal. These values are then summed up and divided with number of metals investigated in this study for the background concentrations are used and multiplied by 100. The following equation was used to calculate the SMI:

$$SMI = (C_1/B_1 + C_2/B_2 + C_3/B_3 + \dots + C_n/B_n) / n \quad (1)$$

where C , in mg/kg, is the measured concentration of a metal for a particular sampling site; B , in mg/kg, is the background (reference) concentration of a metal; and n is a number of metals ($n = 7$ in this study). Because there are no data about the uncontaminated soils in the study area, we used the Dutch reference concentrations of a metal [35] as background concentrations of metals. These reference concentrations were listed in Table 2. The SMI above 100% pointed out the polluted soils while the SMI below 100% indicated the unpolluted soils.

The enrichment factor (EF) of metals is a useful indicator reflecting the status and degree of environmental contamination [36]. The EF calculations compare each value with a given background level, either from the local site, using older soils formed under similar conditions, but without anthropogenic impact, or from a regional or global average composition [37] [38]. The EF was calculated using the method proposed by [39] as follows:

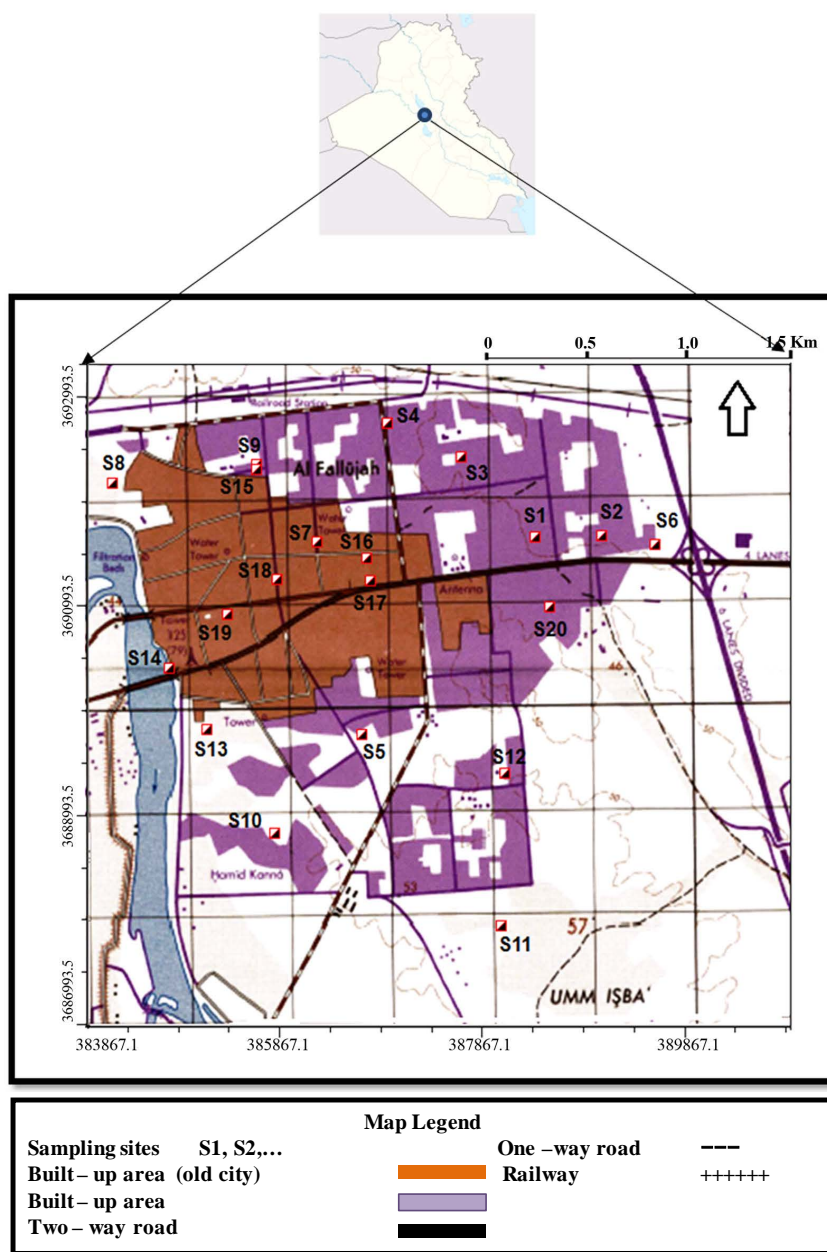


Figure 1. Map of the Fallujah city with location of sampling site of soil (After Salah *et al.* [22]).

$$\text{Enrichment Factor (EF)} = \frac{\left(\frac{\text{Me}}{\text{Fe}}\right)_{\text{sample}}}{\left(\frac{\text{Me}}{\text{Fe}}\right)_{\text{background}}} \tag{2}$$

where (Me/Fe) sample is the metal to Fe ratio in the sample of interest; (Me/Fe) background is the natural background value of metal to Fe ratio. As we do not have metal background values for our study area, we used the values from world average data [40]. Iron was chosen as the element of normalization because natural sources (1.5%) vastly dominate its input [41]. Five contamination categories are recognized on the basis of the enrichment factor, where EF < 2 is deficiency to minimal enrichment; EF 2 - 5 is moderate enrichment; EF 5 - 20 is significant enrichment; EF 20 - 40 is very enrichment, and EF > 40 is extremely high enrichment [42].

Table 1. Concentrations (in mg/kg) of the heavy metals in the sampling sites of the study area.

Sampling Site	Cu	Zn	Ni	Pb	Mn	Fe	Cd	Co	Cr
S1	2.050	8.925	12.05	4.625	25.875	417.700	0.800	3.825	21.225
S2	0.925	3.800	5.575	3.475	10.225	296.350	0.575	2.275	12.325
S3	1.325	5.175	10.475	4.575	21.950	332.725	0.825	3.625	14.300
S4	1.050	5.350	10.375	3.975	22.200	352.950	0.650	3.525	16.725
S5	1.425	6.950	10.525	3.000	20.450	275.400	0.575	3.150	14.900
S6	1.450	3.225	6.400	3.150	18.400	201.875	0.600	2.650	9.450
S7	0.875	4.750	8.400	2.675	27.225	189.325	0.750	3.250	11.600
S8	1.000	6.425	7.375	2.900	21.225	205.750	0.625	3.100	11.300
S9	1.450	4.925	7.525	2.625	22.675	203.300	0.525	2.950	10.725
S10	1.000	2.750	6.275	2.750	13.200	209.250	0.525	2.800	9.400
S11	1.600	3.100	7.550	3.550	16.425	215.525	0.575	2.525	9.350
S12	2.175	3.800	7.475	3.300	19.625	218.650	0.475	2.500	7.900
S13	3.825	3.825	11.800	4.350	31.325	200.125	0.775	4.125	10.100
S14	3.050	8.450	12.225	4.750	31.450	231.625	0.825	3.400	9.975
S15	2.325	6.775	11.775	4.050	30.950	228.125	0.875	3.950	9.425
S16	1.425	4.500	7.750	3.725	22.550	194.675	0.550	3.525	9.450
S17	2.500	5.125	10.025	4.400	31.975	203.450	0.575	4.600	12.25
S18	3.125	6.775	10.675	4.700	36.300	145.025	0.525	4.225	9.550
S19	4.975	10.400	8.150	5.300	28.175	200.475	0.750	4.725	11.675
S20	2.750	5.000	6.876	4.600	29.775	193.175	0.575	4.050	10.275

Table 2. Descriptive statistics parameters for the heavy metals in soil of study area.

Metal	Mean (mg/kg)	Median	Min.	Max.	Range	Standard Deviation	Coefficient Of Variance %	Standard Error	World Average Data	Dutch Soil Standards
Cd	0.64	0.58	0.47	0.87	0.40	0.12	19.03	0.02	0.4	0.8
Co	3.43	3.46	2.27	4.72	2.45	0.70	20.56	0.15	8	20
Cr	11.59	10.50	7.90	21.22	13.32	3.13	27.07	0.70	70	100
Cu	2.01	1.52	0.87	4.97	4.10	1.09	54.24	0.24	30	36
Ni	8.96	8.27	5.57	12.22	6.65	2.14	23.97	0.48	50	35
Pb	3.82	3.85	2.62	5.30	2.67	0.81	21.35	0.18	35	85
Zn	5.50	5.06	2.75	10.40	7.65	2.05	37.33	0.45	90	140
Fe	235.77	207.50	145.02	417.70	272.67	66.29	28.11	14.82	40,000	-

Chemometric techniques provide both qualitative and quantitative information about the source of the contamination [43]. The use of one chemometric techniques alone on soil data can still lead to extremely useful information on the soil quality [44]-[48]. In this study we employed two chemometric techniques: Principal Component Analysis (PCA) and Cluster Analysis (CA).

The aims of PCA are to determine underlying information from multivariate raw data. PCA analyzes a data table representing observations described by several dependent variables, which are, in general, inter-correlated. The goal of PCA is to extract the important information as a set of uncorrelated (*i.e.*, orthogonal) variables. These variables are called principle components, factors eigenvectors, singular vectors, or loadings. Each unit is

also assigned a set of scores which correspond to its projection on the components. The results of the analysis are often presented with graphs plotting the projections of the units onto the components, and the loading of the variables. The importance of each component is expressed by the variance (*i.e.*, eigenvalue) of its projection or by the proportion of the variance explained [49]. Varimax rotation was applied because the orthogonal rotation minimizes the number of variables with a high loading on each component and facilitates the interpretation of relationships and the hypothetical source of metals, lithogenic or anthropogenic [50] [51]. The multivariate statistical method of a small set of data together with other information can provide valuable insight in the context of the sources of soil pollutants [27] [52].

Cluster analysis (AC) is an exploratory data analysis for solving classification problems. Its object is to sort cases, data, or objects into groups or clusters. The resulting clusters of objects should exhibit high internal (within-cluster) homogeneity and high external (between-clusters) heterogeneity [53]. Hierarchical CA, the most common approach, starts with each case in a separate cluster and joins clusters together step by step until only one cluster remains [54]. The results of application of the cluster analysis are best described using a dendrogram or binary tree. Salah *et al.* [55] explained in more detail this technique.

In order to study the pollution sources of the urban soil in Fallujah city by heavy metals, Pearson's correlation, PCA and CA were conducted using Statistica 7 software [56].

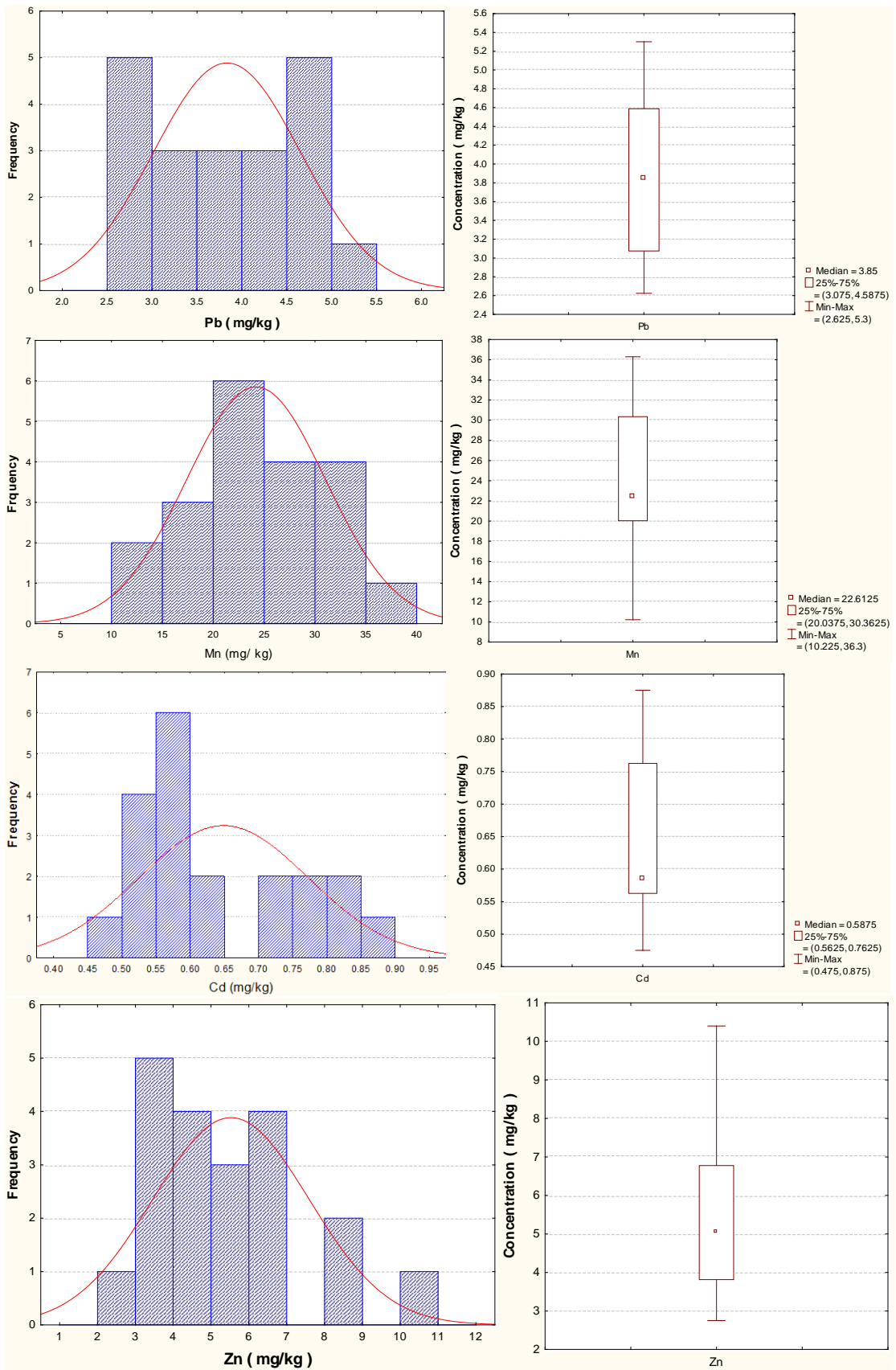
3. Results and Discussion

The descriptive statistics of the measured variables were given in **Table 2**. Based on mean value, the metals follow the decreasing concentration order: Fe > Cr > Ni > Zn > Pb > Co > Cu > Cd. For a symmetrical distribution such as normal, the mean and median are coincide at the mid-point of the distribution. The lower coefficient of variation (CV%) of the variables suggests that the more symmetrical is the distribution. From **Table 2**, the mean value of Pb, Co, and Cd was closer to the median values of them, and they have lower CV%. This result suggests that the distribution of these metals is relative homogeneous in the study area. The distribution of Zn and Cr was less symmetrical relative to distribution of Pb, Co, and Cd. The distribution of Ni and Cu was not symmetry. More homogeneous distribution of Pb, Co, and Cd may be attributed to non-point sources, such as automobile exhaust, manufacturing and building emissions, and urban dust blown by the wind. The asymmetrical distribution of Ni and Cu suggests point sources of contamination, such as dumping contaminated materials (paint, oil, and household cleaners), smelters and industrial workshops.

Histograms and box-plots are very useful for the rough estimation of element distribution [57] as shown in **Figure 2**. The concentrations of Pb and Co have a normal distribution in soil of the study area, Zn and Cd have an approximately normal distribution, and Ni, Cu, and Cr have a non-normal distribution. Parameters such as the 25th percentile, median, 75th percentile and whiskers are shown in the box-plots. In the cases of Pb, Co, and Zn, the concentration median is symmetrical with no extreme outliers. In the cases of Cd, Cr, Ni, and Cu the concentration median is asymmetrical with no extreme outliers.

The results of the Pearson's correlation coefficients and their significant levels ($P < 0.01$) are shown in **Table 3**. Generally, a correlation coefficient > 0.7 is interpreted as a strong relationship between two parameters, whereas values between 0.5 and 0.7 represent a moderate relationship [58]. Correlation analysis provides an effective way to reveal the relationships between multiple variables and thus have been helpful for understanding the influencing factors as well as the sources of chemical components [18]. The relationship between heavy metals can provide important information on heavy metal sources and pathways [12]. The concentration of Cu showed a strong positive relationship with Co (0.70), and Pb (0.75). Geochemically, Cu and Pb are classified as chalcophile and Cu and Co as siderophile. This suggests these heavy metals may originate from a natural common source. The Pb concentration has a strong positive relationship with Co (0.76), and moderate with Zn (0.58) suggesting that these metals come from similar sources. Pb and Zn are chalcophile elements and Co is siderophile but it shows affinity for chalcophile group. Additionally, the correlations between Ni and Cd (0.65) and Ni and Co (0.56) are moderately positive. Ni and Co are classified as siderophile elements and Ni shows affinity for chalcophile group that include Cd. The correlation between Cd and Ni indicates that both are from common source. The concentration of Cr showed a weak to very weak correlations with other metals suggesting that Cr was from different source than other metals. Cr is classified as lithophile element.

The obtained SMI values were below 100% suggesting that the soils of study area are unpolluted with heavy metals. The spatial variation of SMI (**Figure 3**) reflects effect of the anthropogenic activities in the enrichment of heavy metals. The spatial distribution of heavy metals (**Figure 4**) was in a good agreement with that of SMI.



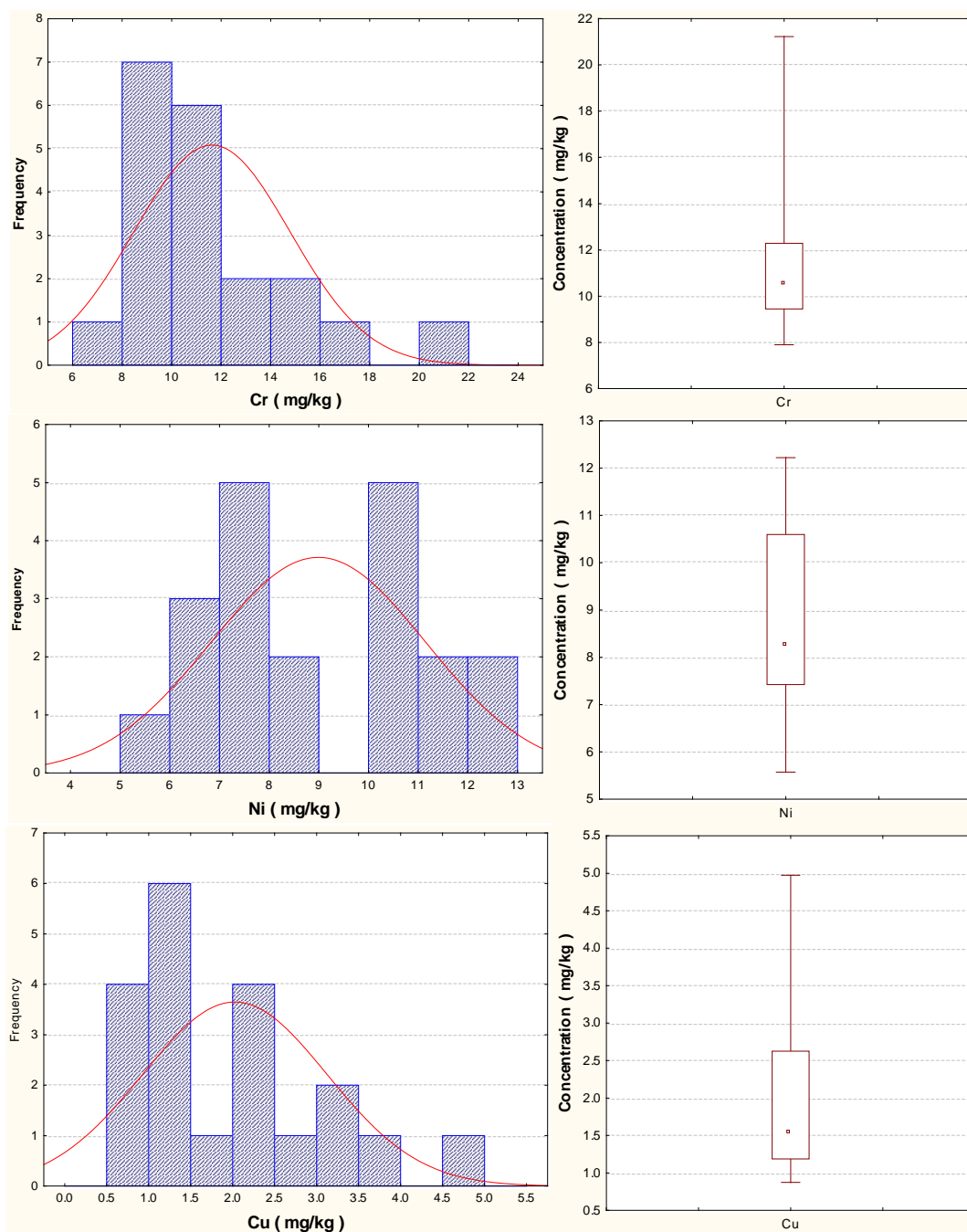


Figure 2. Histograms and box-plot of heavy metals in the soil.

This indicates that we can evaluate the heavy metals distribution depending on the soil pollution level in terms of soil metal index (SMI).

The enrichment factor (EF) was used to quantify the level of pollution and the potential anthropogenic effects in urban soils of Fallujah city. Basically, as the EF values increase, the contributions of the anthropogenic sources also increase. If an enrichment factor is greater than unity, this indicates that abundance of the heavy metals in soil may not come from the local soil background but other natural and/or anthropogenic sources in urbanized areas, including vehicle emissions, industrial discharge and other activities [59] [60]. However, enrichment factors less than 5 may be not considered significant although they are an indicator of metal accumulation, because such small enrichments may arise from differences in the composition of local soil material

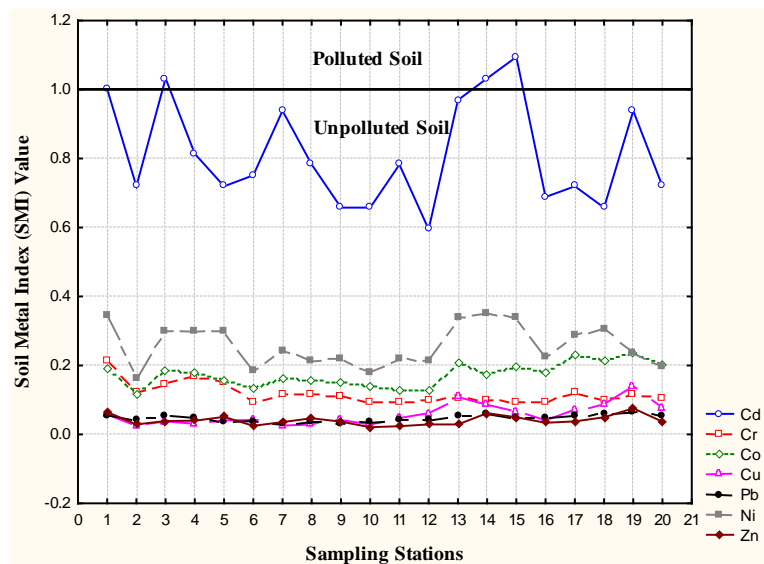


Figure 3. Spatial variation of soil metal index (SMI).

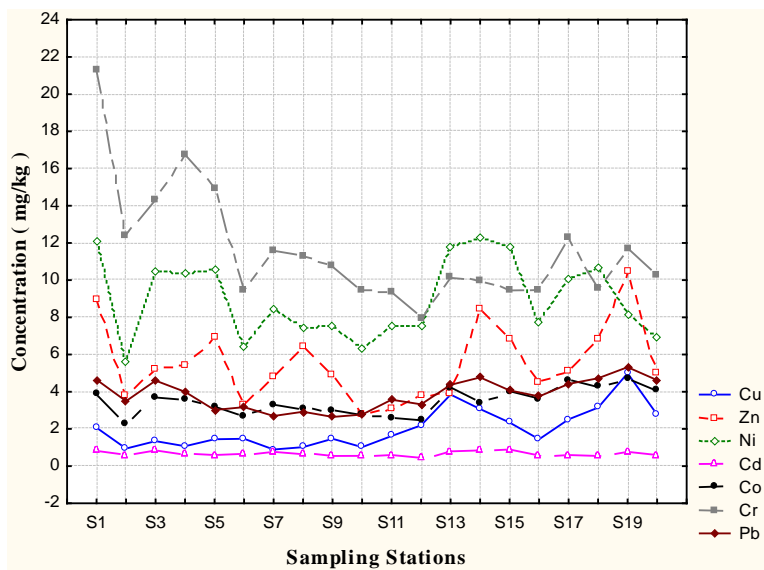


Figure 4. Spatial variation of concentrations of heavy metals in the soil.

Table 3. Pearson's correlation coefficient of heavy metals in soil samples of Fallujah City.

Variable	Cu	Zn	Ni	Cd	Co	Cr	Pb
Cu	1.00						
Zn	0.53	1.00					
Ni	0.37	0.54	1.00				
Cd	0.29	0.51	0.65	1.00			
Co	0.70	0.58	0.56	0.41	1.00		
Cr	-0.17	0.43	0.40	0.33	0.19	1.00	
Pb	0.75	0.58	0.53	0.47	0.76	0.21	1.00

*Marked correlations are significant at $p < 0.01$.

and the reference background value used in the EF calculations [14]. If the EF values are greater than 5, they are considered to be soil contamination for related metals. The mean values of enrichment factors are greater than 5 for Cd (287.13), Co (78.13), Ni (31.88), Cr (28.31), Pb (19.62), Cu (12) and Zn (10.87). The results of EF indicated that the urban soils in Fallujah city were extremely high enriched with Cd and Co, very high enriched with Ni and Cr and Pb, Cu and Zn exhibit significant enrichment. It can be concluded that Cd followed by Co was the most serious enriched heavy metals in the urban soil of Fallujah. The high enrichment factor for Cd and Pb indicate the importance of industrial and traffic emission by fossil fuel and atmospheric deposition as sources of this metal. The significant sources of Co in soils are derived from the deliberate applications of Co salts or Co treated phosphate fertilizers [61]. Because of Fallujah city suffered military operations and bombing in the past years and are still being, we suggest that the military operations can be important source for enrichment heavy metals such as (Pb, Cu, Zn, Ni) in the study area. The sources of the heavy metals in the battlefields and bombed cities are shells, bullets, bombs, unexploded mines, cartridge cases, damaged vehicles, leaking fuel and burning building [62]. The spatial distribution of EF values showed that high enrichment for heavy metals occurs at sampling sites (7 and 13 - 19) located in the built – up area of the old city (Figure 5). This result can be attributed to intensive anthropogenic activities in these sites.

The principal component analysis (PCA) was performed to investigate the relationship among the seven metals analyzed. The principal components with eigenvalues larger than 1 were extracted with loadings rotated for the maximum variance. Two factors were extracted, accounting for 75.315% of the total variance (Table 4). The variance of F1 and F2 was 43.110% and 32.205%, respectively. Spatial representation of the two rotated components is shown in (Figure 6). Factor 1 had strong loadings on Co, Cu and Pb, and moderate positive loading on Zn. Copper, Pb and Zn are chalcophile elements while Co is siderophile element which is primary rock forming element. Correlation analysis showed strong relationship between Cu, Co and Pb, and this confirmed the above result. The association of Cu, Pb, and Zn in F1 may be attributed to their lithogenic source. Factor 2 comprised Cr and Ni with high loadings and Cd and Zn with moderate loadings. The good relation between Ni and Cd (0.65) confirmed this result. As shown in (Table 4) Zn was distributed in F1 and F2. The association of Cr, Ni, and Cd in F2 suggests that these metal come from mixed sources.

Researches indicates that Cr and Ni come from the same mineral, and they always associate with each other [63] [64]. The possible sources of Ni and Cd in the soil of study area are agricultural fertilizers, residues from combustion of fossil fuels and waste water.

The dendrogram of the hierarchical cluster analysis of metal contents in the study area is shown in (Figure 7). Two distinct clusters were identified. The first cluster (I) contained Cu, Co, Pb, Cd, and Zn. The second cluster (II) included Ni and Cr. The elements of cluster I are considered to be mixed in origin. Cu, Zn and Pb may be

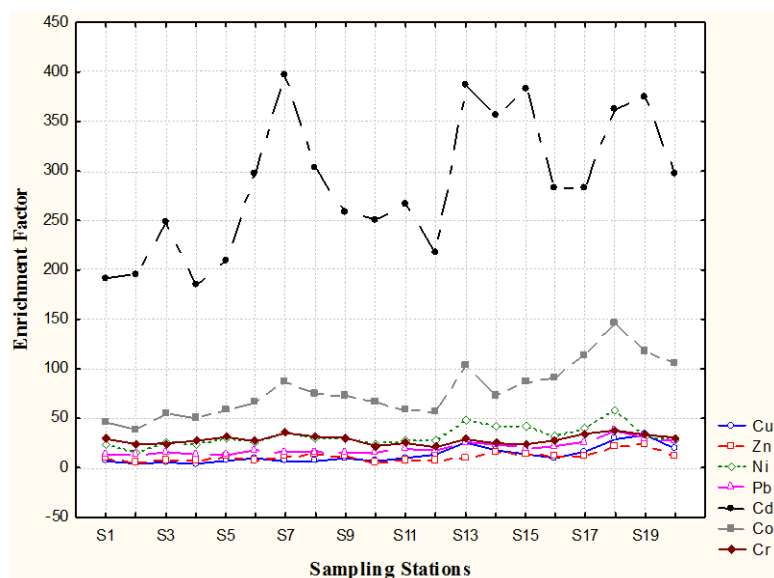


Figure 5. Spatial variation of enrichment factor of heavy metals of the urban soil in Fallujah city.

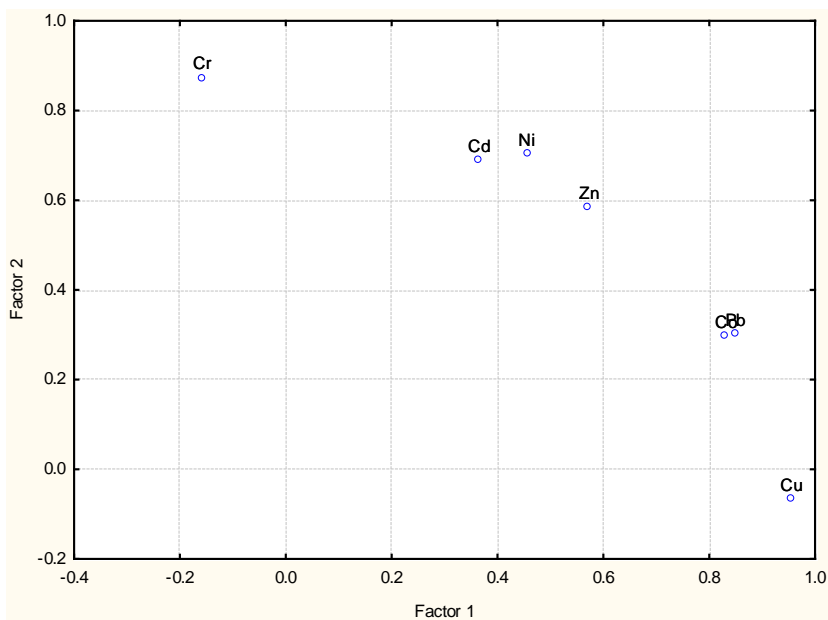


Figure 6. Principal component analysis loading parts for the rotated components.

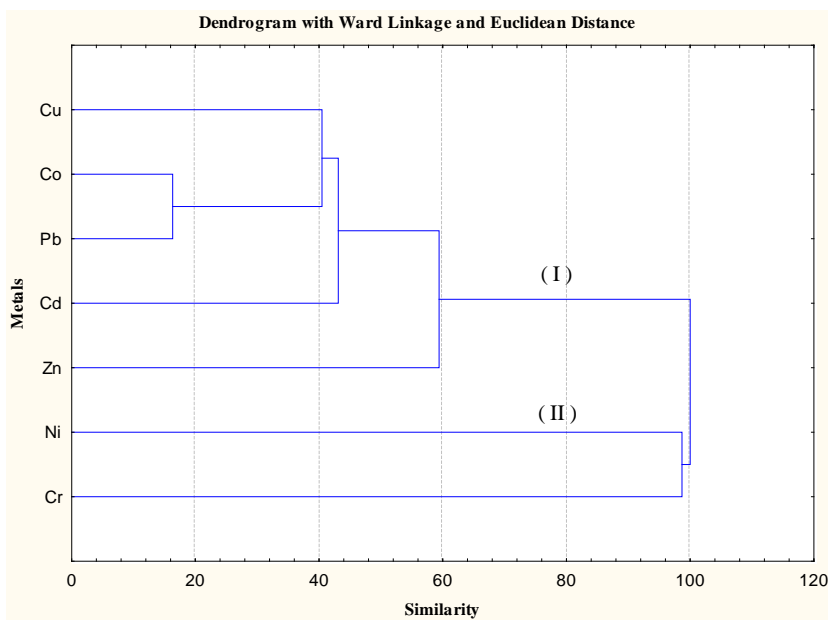


Figure 7. Dendrogram of the cluster analysis of the urban soil in Fallujah city.

the result of weathering of sulfide minerals such as galena and sphalerite. The elements of cluster II are of lithogenic origin. The results of cluster analysis confirmed the results of PCA except for Cd. Cluster analysis is often used to confirm the results of principle component analysis [45] [48] [65] [66].

The groupings from hierarchical cluster analysis showed sampling stations ordered in two main clusters, I and II (Figure 8). The cluster I includes S1 only. High concentrations of heavy metals were reported in this station as shown in Figure 4. Cluster II contained other stations (S2-S19). The cluster II was classified into three groups: A, B, and C. The group A includes (S3-S5 and S18), group B (S7, S13-S15, S17, S20), and group C (S2, S6, S8-S12, S16). The spatial distribution of heavy metals concentrations confirmed these results (Figure 4). These findings indicate that members of each group were distributed based on whether the sources of enrichment were point or nonpoint.

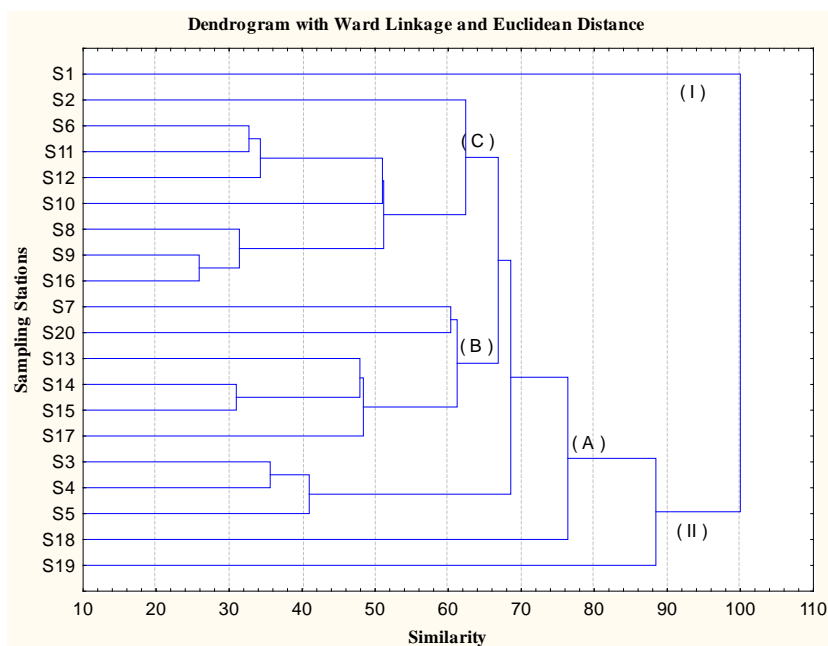


Figure 8. Dendrogram of spatial clustering of sampling stations.

Table 4. Factor loadings (varimax rotation) on heavy metals in soil samples in Fallujah city.

Element	Factor 1	Factor 2
Cd	0.365	0.689
Co	0.831	0.295
Cr	-0.157	0.873
Cu	0.954	-0.067
Ni	0.455	0.701
Pb	0.848	0.301
Zn	0.572	0.584
Eigenvalue	3.017	2.254
Variance explained %	43.110	32.205
Cumulative variance %	43.110	75.315

4. Conclusions

From the obtained results, we concluded the following:

- 1) According to the soil metal index, the urban soils of Fallujah city is unpolluted.
- 2) The industrial and traffic emission by fossil fuel, atmospheric deposition, and the military operations and bombing caused significant to extremely high enrichments of heavy metals, such as Cd, Co, Ni, Cr, Pb, Cu, and Zn.
- 3) The moderate to strong positive correlations among Cd, Co, Cu, Cr, Ni, Pb, and Zn suggest that these metals have common sources.
- 4) The results of a PCA performed on seven heavy metals confirmed by the cluster analysis (CA), identified two factors controlling their variability in urban soil of Fallujah city. The variability of Cu, Co, Pb, and Zn is associated in the same factor and is controlled by lithogenic origin. The variability of Cr, Ni, and Cd is controlled by mixed origin (lithogenic and anthropogenic).
- 5) This study also reconfirms that chemometric techniques such as PCA and CA can be used to identify sources of soil pollution.

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