

Green Synthesis of Carvenone by Montmorillonite-Catalyzed Isomerization of 1,2-Limonene Oxide

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ABSTRACT

Montmorillonite was considered as a good heterogeneous catalyst for the isomerization of 1,2-limonene oxide into carvenone under solvent-free condition. Both conventional heating and green activations were tested in this research. The microwave-assisted isomerization afforded carvenone in high yield within only 6 minutes.

Keywords: Carvenone; Isomerization; Green Chemistry; Montmorillonite; Solvent-Free Synthesis

1. Introduction

Carvenone, *p*-menth-3-en-2-one, has been considered as a useful constituent for perfume, flavor, and fragrance [1,2]. This compound is one of the major products achieved in Brønsted acid-catalyzed isomerization of different precursors, e.g. dihydrocarvone [3,4] and 1,2-limonene oxide. Because the latter compound is easily prepared from naturally available limonene [5], the interest for direct syntheses of carvenone from 1,2-limonene oxide has been increasing [6-21]. However, most of the related reports showed that the selectivity towards carvenone was not good as expected. Arata's group developed a pathway in which 1,2-limonene oxide was isomerized over H₂SO₄/SiO₂ or SiO₂-Al₂O₃ to give carvenone in only modest yields [16]. Ravasio, *et al.* reported that a quantitative isomerization of limonene oxide can be carried out over copper supported on activated SiO₂-Al₂O₃ in toluene, but this was not the selective route to prepare carvenone [21]. The most selective synthesis of carvenone directly from 1,2-limonene oxide was reported by Yoshito, *et al.* [2]. In this procedure, carvenone was prepared by dropwise addition of 1,2-limonene oxide into a tubular reactor filled with CaA zeolite at 200°C and 10 mm Hg to give a mixture rich of carvenone.

Herein, we reported a mild and simple pathway to selectively obtain carvenone by the isomerization of 1,2-li-

monene oxide over montmorillonite prepared from Vietnamese natural clay. Various techniques including conventional heating (CH), ultrasonic irradiation (US), and microwave irradiation (MW) were used to induce the reaction under solvent-free condition.

2. Results and Discussion

The isomerization of 1,2-limonene oxide **1** over Vietnamese montmorillonite under solvent-free conditions resulted in various products (**Table 1** and **Scheme 1**) that were produced from the isomerization (**6**, **7**, **8**, **9**, **10**, **12**, and **13**), aromatization (**3**), and other processes (**2**, **4**, and **5**). Their presence as well as percentage in the final mixture depended on which technique was used. **Table 1** summarized the influence of activation techniques on reaction outcomes. Besides being directly produced as one of the isomerization products from 1,2-limonene oxide, the formation of carvenone **12** was also induced due to the shift of hydrogen in the framework of the other isomerization products. Therefore, it was not surprising that carvenone **12** made up the largest proportions of the final mixtures, especially the samples obtained under the CH and MW techniques (**Table 1**). As a result, it was strongly believed that the isomerization of 1,2-limonene oxide into carvenone **12** is an endothermic process. This again was confirmed through the results in **Table 2** which showed the relationship between the temperature and the selectivity towards carvenone **12** for the CH

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Table 1. The isomerization of 1,2-limonene oxide over Vietnamese montmorillonite^a.

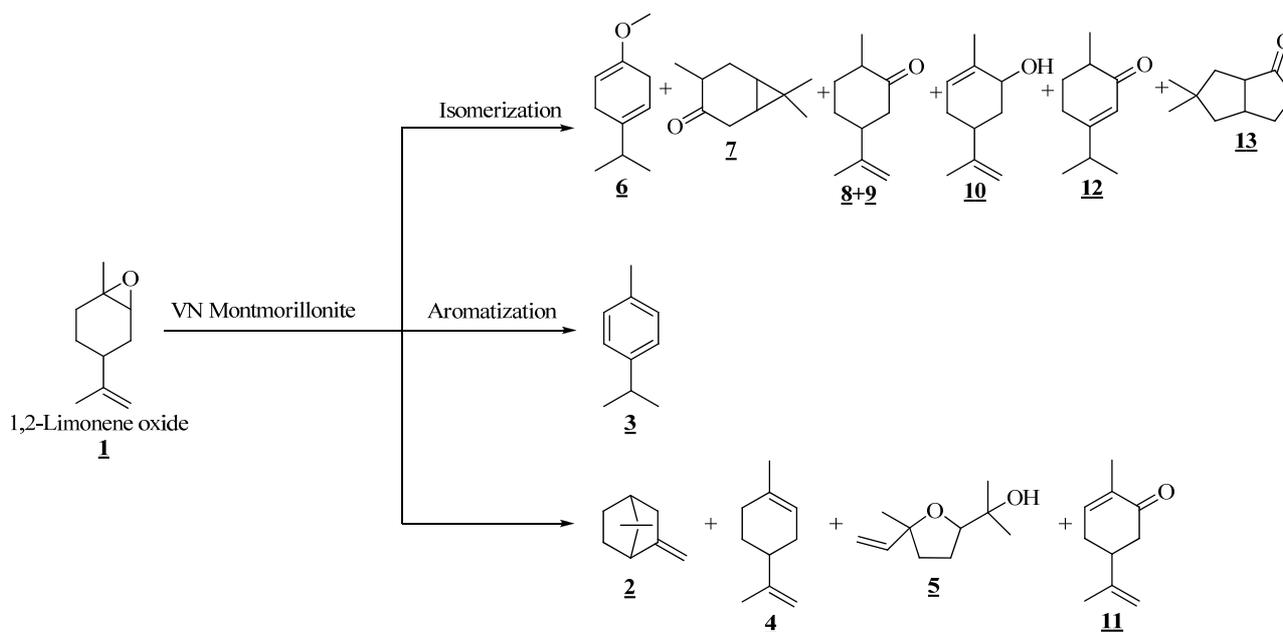
Compounds	Selectivity (%)		
	CH ^b	US ^c	MW ^d
α -Fenchene (2)	-	-	0.75
<i>p</i> -Cymene (3)	8.97	-	6.89
Limonene (4)	-	3.64	-
<i>trans</i> -Linalool oxide (5)	1.61	2.04	0.84
1-Methoxy-4-(1'-methylethyl)cyclohexa-1,4-diene (6)	3.27	12.09	0.81
<i>trans</i> -3-Caranone (7)	0.99	1.27	1.65
<i>cis</i> -Dihydrocarvone (8)	-	19.69	-
<i>trans</i> -Dihydrocarvone (9)	-	9.99	-
<i>trans</i> -Carveol (10)	-	2.71	-
Carvone (11)	-	-	1.07
Carvenone (12)	85.16	47.70	88.00
7,7-Dimethylbicyclo [3.3.0] octan-2-one (13)	-	0.88	-
Yield of carvenone (%)	81	40	82

a. 1,2-limonene oxide:montmorillonite = 1:0.08 (w/w); b. 140°C, 60 min; c. 111 μ m (amplitude), 75°C - 77°C, 8 min; d. 455 W (power), 138°C - 141°C, 6 min.

Table 2. The influence of temperature on the reaction under CH^a.

Compounds	Selectivity (%)				
	40°C	60°C	80°C	100°C	140°C
8 + 9	47.32	44.77	44.06	30.24	-
12	6.67	11.61	28.41	48.18	84.75
Others	46.01	43.62	27.71	21.59	15.25
Yield of carvenone (%)	6	11	26	45	80

a. 1,2-limonene oxide:montmorillonite = 1:0.1 (w/w), reaction time: 60 min.

**Scheme 1. Vietnamese montmorillonite-catalyzed isomerization of 1,2-limonene oxide.**

method. Obviously, there was a significant increase in the amount of the product carvenone **12** from only 6.67% at 40°C to 85.16% at 140°C.

The US method greatly enhanced the reactivity due to releasing the energy by cavitation collapsing the bubbles, especially for reactions involving solids. And as expected, the conversion of 1,2-limonene oxide could reach to the completion after a short irradiation time (8 min) at 75°C - 77°C (**Table 1**). However, a poor selectivity towards carvenone **12** was detected because the ultrasound energy was not strong enough to shift hydrogen in the isomerization process.

Due to the effective absorbability for microwave energy of 1,2-limonene oxide and montmorillonite to supply heat for the total reaction mixture, the endothermic isomerization into carvenone **12** was favored under the MW. A complete conversion was obtained after only 6 min at 138°C - 141°C to afford carvenone **12** in the nearly same yield compared with the CH.

Unlike Yoshito's procedure [2], the isomerization of 1,2-limonene oxide using the modified montmorillonite as a Lewis acid catalyst can furnish carvenone **12** under solvent-free condition. Besides, good selectivity towards carvenone **12** could be obtained when using the CH and MW techniques. Consequently, the synthesis of carvenone can be accomplished in 6 minutes with 100% selectivity and 82% yield.

3. Experimental

3.1. General

All chemicals, unless otherwise noted, were purchased from commercial sources and used without further purification. The catalyst montmorillonite was prepared from Vietnamese natural clay according to Pesco method [23]. The analyses for quantification and identification were performed on GC-FID and GC-MS, respectively. GC-FID analyses were recorded on Agilent 6890N GC, equipped with a fused silica capillary column DB-5 (30 m × 0.32 mm × 0.25 μm). GC-MS spectra were obtained with Agilent 7890A GC coupled to 5973C VL MSD Triple-Axis selective detector and equipped with a capillary column HP-5MS (30 m × 0.25 mm × 0.25 μm).

3.2. Preparation of 1,2-Limonene Oxide

- Extraction of sweet orange peel oil: The fresh sweet orange peels were subjected to hydrodistillation for 6 h. The resulting water-oil samples were extracted with diethyl ether. The combined ethereal extracts were dried over anhydrous sodium sulfate for 24 h, filtrated, and rotary evaporated under vacuum to afford the essential oil as a colorless liquid.
- Isolation of limonene: Limonene was isolated from

sweet orange peel oil by vacuum fractional distillation using Vigreux apparatus. The limonene fraction was collected at 55°C - 56°C, 12 mm Hg in the recovery yield of 83%. The purity of limonene was checked by GC-MS analysis.

- Synthesis of 1,2-limonene oxide: The epoxidation of limonene was performed in the same way as previously reported by Mandelli and Silva (**Scheme 1**) [5,22]. The product was identified as a mixture of *cis/trans*-1,2-limonene oxide (69%).

3.3. General Procedure for Synthesis of Carvenone

A 100 mL round-bottom flask was charged with 1,2-limonene oxide (1.0 g) and montmorillonite (0.08 g). The reaction mixtures were then subjected to different techniques for organic synthesis:

- CH: The reaction flask was fitted with a condenser, immersed in an oil bath, heated and simultaneously stirred at 140°C for 60 min.
- US: The reaction mixture was sonicated for 8 min at the amplitude corresponding to the wavelength of 111 μm emitted by a Cole Parmer GE130 Ultrasonic Processor equipped with a microtip and a probe fabricated from titanium alloy.
- MW: The reaction mixture was heated at the power of 455 W for 6 min in a modified domestic microwave oven KENWOOD SJW17.

Upon completion, the resulting mixtures were cooled to room temperature and diluted with diethyl ether. The solid catalyst was removed by filtration. The ethereal solutions were washed with water, dried over anhydrous sodium sulfate, and concentrated under reduced pressure to afford the mixture of products whose constituents were identified and quantified by GC-MS and GC-FID, respectively. The calculated yields of carvenone were 81, 40, and 82% for the CH, US, and MW techniques, respectively.

4. Conclusion

From the natural materials, such as clay and *Citrus* peel essential oil, carvenone can be efficiently prepared in high yield via the isomerization of 1,2-limonene oxide under solvent-free condition. Particularly, in the presence of microwave irradiation, the reaction can be greatly enhanced under mild conditions.

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