

# Caffeoyl Derivatives: Major Antioxidant Compounds of Some Wild Herbs of the *Asteraceae* Family

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# ABSTRACT

The polyphenolic composition and antioxidant activity of the aerial parts of 18 medicinal or food plants of the Asteraceae family were studied. Five main caffeoyl derivatives were determined individually by HPLC and compared with levels determined by colorimetry for total dihydroxycinnamic derivatives and total phenolics. The aim of this study was to assess the contribution of these constituents to the antioxidant activity of the herbs determined by DPPH radical scavenging tests. Significant correlations were found between total phenolic ( $R^2 = 0.8904$ ), total dihydroxycinnamic derivative ( $R^2 = 0.8529$ ) and total caffeoyl derivative ( $R^2 = 0.7172$ ) concentrations and the DPPH-scavenging ability of all herbs. The antioxidant activity of the main constituents, including chicoric acid ( $EC_{50} = 8.24 \mu mol/l$ ) or 3,5-dicaffeoylquinic acid ( $EC_{50} = 7.62 \mu mol/l$ ), was very high compared to vitamin C ( $EC_{50} = 15.66 \mu mol/l$ ). Thus, for each species, antioxidant activity mainly involves the major caffeoyl derivatives. The contribution to antioxidant activity were assessed as 48.92% for 3,5-dicaffeoylquinic acid in Tanacetum parthenium (30.08 g/kg), and 68.96% for chicoric acid in Taraxacum officinale (34.08 g/kg). The main caffeoyl derivatives among polyphenols can be considered as the major antioxidant compounds of the studied Asteraceae herbs.

Keywords: Plants, Asteraceae, Caffeoyl Derivatives, Antioxidant

## 1. Introduction

Polyphenols, particularly flavonoids, have attracted a great deal of research on their broad distribution in plants, their physiological (including antioxidant) activities, and their health effects [1-3]. In the Asteraceae family, flavonoid composition has long been established, but much less is known about their phenolic acid derivatives composition. The most common flavonoids are the 7-glycosides of apigenin and luteolin [4,5]. Among the dihydroxycinnamic derivatives, the oldest known are cynarin (dicaffeoylquinic acid) from artichoke and chicoric acid (dicaffeoyltartaric acid) from chicory [6,7]. Among the 18 medicinal or food species of Asteraceae studied, recent work has helped clarify the caffeoylquinic acid composition or show the presence of these constituents in some of these plants. In the subfamily Cichorioideae, chicory and dandelion leaves but not mouse-ear contain large amounts of chicoric acid [8-10]. In the subfamily

Asterideae, the dicaffeoylquinic acid composition is now relatively well known for arnica, chamomiles, feverfew, giant goldenrod, milfoil, mugwort and tansy, but these compounds have not been revealed in the other plants studied [11-19]. Extensive research on the antioxidant activity of plants was performed in the *Asteraceae* family [20-32]. In some cases, antioxidant capacity was related to the presence of polyphenols, including favonoids [33-41]. To our knowledge, only two studies highlighted important roles of dicaffeoylquinic derivatives in antioxidant activity, one in feverfew and the other in tansy [15,19].

Having previously reported that the major polyphenols in aerial parts of mugwort and yarrow were dicaffeoylquinic acids [17,18], we screened and assayed these constituents in other *Asteraceae* herbs and studied whether they had an impact on the antioxidant activity of these plants. The aim of this study was to specify the qualitative and quantitative composition of five main caffeoyl derivatives, *i.e.* chlorogenic acid, chicoric acid and 1,5-, 3,5- and 4,5-di-O-caffeoylquinic acids in the aerial parts of 18 wild plants of European *Asteraceae* used in traditional medicine or food. In addition to dihydroxycinnamic derivatives, we ran comparative assays on the overall content of flavonoids and phenolics that also demonstrate marked antioxidant activity. We then investigated the correlation between total antioxidant capacity, determined *in vitro* using the 2,2'-diphenyl-1-picrylhydrazyl free radical (DPPH) assay, of each plant and their main antioxidant polyphenolic compounds.

## 2. Materials and Methods

#### 2.1. Plant Material

Aerial parts, *i.e.* flowering tops or leaves (if specified), of 18 medicinal or food taxa of the *Asteraceae* (or *Compositae*) family were examined, *i.e. Achillea millefolium* subsp. *millefolium*, *Arnica montana* subsp. *montana*, *Artemisia absinthium*, *Artemisia verlotiorum*, *Artemisia vulgaris*, *Calendula officinalis*, *Chamaemelum nobile*, *Chamomilla recutita*, *Cichorium intybus* (leaves), *Conyza canadensis*, *Eupatorium cannabinum* subsp. *cannabinum*, *Hieracium pilosella* s.l., *Matricaria perforata*, *Solidago gigantea* subsp. *serotina*, *Solidago virgaurea* subsp. *virgaurea*, *Tanacetum parthenium*, *Tanacetum vulgare*, *Taraxacum officinale* gr. (leaves). Samples of each herb were collected wild at flowering in the Auvergne (France) in summer 2005. All plants were identified and named using Flora Europaea [42].

### 2.2. Chemicals

Standard chlorogenic acid was purchased from Extrasynthese (Genay, France). Chicoric acid (dicaffeoyltartaric acid) and 1,5-, 3,5- and 4,5-di-*O*-caffeoylquinic acids were isolated as described elsewhere [7,15,18]. All other reagents were of pharmacopoeia purity and organic solvents were HPLC grade (Merck, Darmstadt, Germany).

## 2.3. Polyphenolic Compound Extraction Procedure

Samples of herbs desiccated in the lab at room temperature were pulverized in a laboratory crusher (IKA A10 type, Bioblock, Illkirch, France). The powder was sieved (mesh 0.5 mm) and 250 mg of ground sample were carefully extracted with 100 ml of EtOH/H<sub>2</sub>O (50:50 v/v) in a water bath at 80°C for 1 min, stirred for 1 h at room temperature, then filtered. Before HPLC analysis, the samples were filtered through a 0.45  $\mu$ m filter (Acrodisc GPH, Gelman, Ann Arbor, MI, USA).

## 2.4. TLC Identification of Caffeoyl Derivatives

Oualitative analysis of polyphenols by TLC was carried out with a routine technique using Silicagel 60F<sub>254</sub> plates (Merck) [43]. The mobile phase was: toluene/ethyl formate/water/formic acid (10:100:5:10 v/v). A 10 µl aliquot of each sample solution prepared as above and 10 µl of each selected standard solution (0.5 mg/ml) were applied comparatively on the layer. After migration then drying, Neu reagent was sprayed on the plate. Neu reagent was previously prepared with 10 ml of 1% methanolic diphenylboric acid 2-aminoethyl ester (Fluka) and 8 ml of 5% ethanolic polyethylene glycol 4000 (Sigma, St Louis, MO, USA). After 1 h at room temperature, each standard compound studied under UV light at 365 nm appeared with blue fluorescence at frontal ratio (Rf): chlorogenic acid (0.20), 4,5-di-O-caffeoylquinic acid (0.43), 1,5-di-O-caffeoylquinic acid (0.62), 3,5-di-O-caffeoylquinic acid (0.65), chicoric acid (0.66).

#### 2.5. HPLC Analysis of Caffeoyl Derivatives

Qualitative and quantitative HPLC analysis was carried out with an apparatus comprising two 510 pumps, a 680 solvent programmer and a 991 photodiode array detector (Waters Associates, Milford, MA, USA). A 25 µl aliquot was injected onto a Lichrocart 125-4 Superspher RP8-E 5 µm column (Merck). The mobile phase consisted of solvent A:  $H_2O/H_3PO_4$  85% (100:0.3 v/v) and solvent B: MeCN/H<sub>2</sub>O/H<sub>3</sub>PO<sub>4</sub> 85% (80:20:0.3 v/v). Separation was performed by a quadriconcave gradient of B in A at a flow rate of 2 ml/min as follows: 0 - 5 min, 12% - 15% B; 5 - 30 min, 15% - 30% B; 30 - 40 min, 30% - 50% B; 40 -45 min, 50% - 70% B. Under these conditions, standard compounds were correctly separated and eluted at approximate retention times  $(R_t)$ , as follows: chlorogenic acid (7.1 min), 3,5-di-O-caffeoylquinic acid (32.5 min), 1,5-di-O-caffeoylquinic acid (33.0 min), chicoric acid (33.5 min), 4,5-di-O-caffeoylquinic acid (34.2 min). The specificity of the method was verified for each hydroxycinnamic constituent using a photodiode array detector to compare their UV spectra with those of standard compounds. UV absorption maxima were 217, 238 and 325 nm for chlorogenic acid, 218, 241 and 328 for 3,5-di-Ocaffeoylquinic acid, 218, 240 and 328 for 1,5-di-O-caffeoylquinic acid, 218, 243 and 329 for chicoric acid and 217, 241 and 326 for 4,5-di-O-caffeoylquinic acid. Detection was therefore carried out at 340 nm for all compounds. Linearity and reliability standard deviations of caffeovl derivatives were < 5%. Linearity correlation coefficient was greater than 0.99 (5 points; 3 assays). All samples were run in triplicate and quantification was carried out using external standards. The content of each compound was calculated and expressed as g/kg on dry

matter (DM) [17].

### 2.6. Determination of Total Dihydroxycinnamic Derivatives

Total dihydroxycinnamic acid contents (including caffeoyl derivatives) were expressed as chlorogenic acid as previously described in the European Pharmacopoeia (6th ed. 2008) for Fraxini folium [44]. The extract (1 ml) was added to 2 ml 0.5 M HCl, 2 ml Arnow's reagent (10 g sodium nitrite and 10 g sodium molybdate made up to 100 ml with distilled water), 2 ml NaOH (at a concentration of 2.125 M) and 3 ml of water. Each solution was compared with the same mixture without Arnow's reagent. Absorbance was read at 525 nm. The content of each plant was calculated and expressed as g/kg DM.

## 2.7. Determination of Total Flavonoids

Total flavonoid contents were determined and expressed as luteolin 7-glucoside as previously described in the European Pharmacopoeia (6th ed. 2008) for Passiflorae herba [44]. The extract (1 ml) was evaporated to dryness under reduced pressure, and the residue dissolved in 10 ml of MeOH/acetic acid (10:100 v/v). The solution was added to 10 ml oxaloboric reagent (25 g boric acid and 20 g sodium molybdate made up to 1000 ml with anhydrous formic acid) and adjusted to 25 ml with anhydrous acetic acid. Each solution was compared with the same mixture without reagent. After 30 min, absorbance was read at 401 nm. The content of each plant was calculated and expressed as g/kg DM.

#### 2.8. Determination of Total Polyphenolic Compounds

The amount of total phenolics was determined according to the Folin-Ciocalteu procedure [45]. The Folin reagent (diluted 1:10 in water, 750  $\mu$ l) and aqueous Na<sub>2</sub>CO<sub>3</sub> (75 g/l, 600  $\mu$ l) were successively added to the herb extract (150  $\mu$ l). The mixture was kept in a water bath at 50°C for 5 min, then chilled on ice before reading absorbance at 760 nm. Pyrogallol was used to establish the calibration curve, and total polyphenolic content was expressed as g/kg DM.

#### 2.9. Determination of Antioxidant Activity

Extraction was carried out using 250 mg (*m*) of plant powder in 100 ml of EtOH/H<sub>2</sub>O (50:50 v/v). The mixture was carefully heated for 1 min in a water bath at 80°C, then stirred for 1 h at room temperature and filtered. The absorbance of a blank (2.5 ml of 25 mg/l DPPH (colored radical) in MeOH + 10 µl of EtOH/H<sub>2</sub>O (50:50 v/v)) was compared with the absorbance of the sample extract (2.5 ml of 25 mg/l DPPH in MeOH + 10 µl of extractive solution). After 30 min, absorbance (A) was determined at 517 nm [46].

Relative antioxidant activity on DPPH (total capacity %) was expressed as a percentage decrease of the absorbance of origin ( $A_0$ ):

Total Capacity 
$$\% = (A_0 - A) \times 100 / A_0$$

Absolute antioxidant activity was expressed as the amount of plant yielding a 50% decrease in the original color under the test conditions, so that half maximal effective concentration ( $EC_{50}$ ) in µg of plant by ml of reaction mixture was given by:

$$EC_{50} = m \times 50 \times 10/2.5 \times 100 \times (\text{total capacity \%})$$
$$= m \times 2/(\text{total capacity \%})$$

For the compounds isolated,  $EC_{50}$  were expressed in  $\mu$ mol/l.

## 2.10. Calculation of the Contribution of Caffeoyl Derivatives to Total Antioxidant Activity

After determining the amount of each caffeoyl derivative by HPLC, their contribution to the total antioxidant capacity (%) of each plant was calculated as described previously for chicoric acid (dicaffeoyltartaric acid) and chlorogenic acid [47].

"A" was the amount of chicoric acid determined by HPLC (g/kg DM). First, we calculated the amount of chicoric acid in 10  $\mu$ l of extraction solution:

 $A \times m \times 10/1000 \times 100$ 

The antioxidant activity of chicoric acid solution used as reference ( $A_0$ : amount of chicoric acid expressed as mg/ml) was 34.44%.

Then, the percentage activity due to chicoric acid from plant was calculated as:

$$X\% = A \times m \times 10 \times 34.44 / A_0 \times 1000 \times 100 \times 10$$

The contribution of chicoric acid to the total antioxidant capacity of each plant was calculated as:

 $X\% \times 100/(\text{total capacity}\%)$ 

The same method was applied to determine the contributions of chlorogenic acid and 1,5-, 3,5-, and 4,5- dicaffeoylquinic acids.

## 2.11. Statistical Analysis

Results of triplicate analyses were expressed as mean  $\pm$  SD. Relationships between phenolic compound contents and antiradical efficiency were established using the Pearson correlation test (*P* < 0.05).

## 3. Results and Discussion

## 3.1. Polyphenolic Composition of the 18 Asteraceae Herbs

In the 18 Asteraceae samples, 5 major caffeoyl derivatives

were individually investigated. Each component was cochromatographed (TLC and HPLC) with its corresponding standard substance, and the spectral characteristics were compared for identification. Ubiquitous chlorogenic acid (3-caffeoylquinic acid) was present in all herbs, confirming previous results [17]. Chicoric acid (dicaffeoyltartaric acid) was present only in two studied species of the subfamily Cichorioideae, i.e. C. intybus and T. officinale, as previously reported [8,9]. 3,5-dicaffeoylquinic acid was found in all other studied Asterideae subfamily plants but in none of the studied Cichorioideae except H. pilosella. HPLC showed chicoric acid is largely predominant in chicory and dandelion, with lesser amounts of chlorogenic acid in the absence of dicaffeoylquinic derivatives. The genus Hieracium (Cichorioideae) is clearly distinguishable from its subfamily as its composition is typical of the Asteridae subfamily [10]. In almost all representatives of the Asteridae subfamily, as well as in the mouse-ear, the principal component is 3,5-dicaffeoyl-quinic acid. In some cases, the major compound is 1,5-dicaffeoylquinic acid, as in A. verlotiorum, or chlorogenic acid, as in S. gigantea, but the presence of 1,5-dicaffeoylquinic acid was inconsistent within the species. Most of the 18 species except chicory and dandelion contain 4,5-dicaffeoylquinic acid, but in relatively low proportions. Characteristic HPLC profiles of the caffeoyl derivatives from H. pilosella (containing chlorogenic acid, 1,5-, 3,5- and 4,5-dicaffeoylquinic acids) and T. officinale (chlorogenic and chicoric acids) are shown in Figure 1.

Table 1 gives quantitative data on polyphenolic compound levels in the 18 Asteraceae herbs. Among individual caffeoyl derivatives, large amounts of chicoric acid were found in C. intybus ( $(38.67 \pm 1.65)$  mg/kg DM) and T. officinale ((34.08  $\pm$  1.90) mg/kg DM). Other plants studied generally contain significant amounts of 3,5-dicaffeoylquinic acid, including A. millefolium, A. absinthium, A. vulgaris, E. cannabinum, H. pilosella, S. virgaurea, and T. parthenium which was the richest at approximately 20 to 30 g/kg. Levels were fairly variable depending on species, ranging from 0.78 to 30.08 g/kg with a mean of  $(13.27 \pm 9.36)$  g/kg. High quantities of 1,5-dicaffeoylquinic acid were only found in A. verlotiorum, at 34.85 mg/kg. Significant quantities of 20 to 25 g/kg of chlorogenic acid were found in A. verlotiorum, H. pilosella and S. gigantea, with a main and SD of (8.07  $\pm$  7.29) g/kg indicating a high variability among species (0.55 to 25.54 g/kg). In the Asteridae herbs and H. pilosella, 4,5-dicaffeoylquinic acid was found in only minor amounts, *i.e.* 0.80 to 6.05 g/kg with a mean of  $(2.59 \pm 1.90)$  g/kg, with C. canadensis having the highest content. For the 18 Asteraceae species, overall rates

calculated by summing individual caffeoyl derivative levels determined by HPLC (T1 in **Table 1**) ranged from 1.33 g/kg for *C. officinalis* to 62.85 g/kg for *H. pilosella*, with an average value of  $(32.18 \pm 15.80)$  g/kg. A few species were poor in these constituents as *C. recutita*, about 5 g/kg, while most are much richer as *A. verlotiorum* 

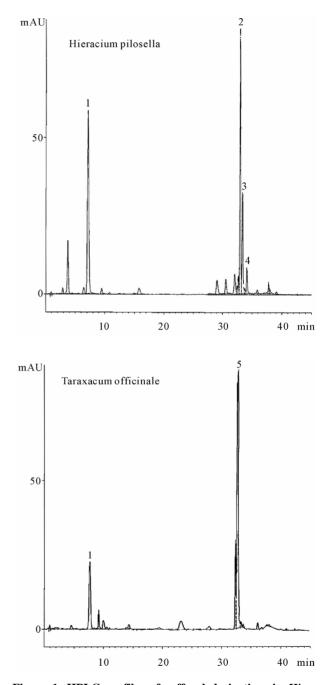


Figure 1. HPLC profiles of caffeoyl derivatives in *Hieracium pilosella* and in *Taraxacum officinale*. Peaks: (1) Chlorogenic acid, (2) 3,5-Dicaffeoylquinic acid, (3) 1,5-Dicaffeoylquinic acid, (4) 4,5-Dicaffeoylquinic acid, (5) Chicoric acid.

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	Polyphenolic compounds (g/kg DM)										
Taxon	Chlorogenic acid	3,5-DCQA	1,5-DCQA	4,5-DCQA	Chicoric acid	T1	T2	Т3	T4	Т5	
Achillea millefolium	$8.12\pm0.40$	$21.59 \pm 1.08$	$8.88 \pm 0.44$	$3.31\pm0.22$	ND	41.90	$52.67 \pm 2.90$	$12.92\pm0.68$	65.59	$63.06\pm3.16$	
Arnica montana	$3.63\pm0.18$	$7.52\pm0.37$	$11.80\pm0.59$	ND	ND	22.95	39.32 ± 2.01	$8.12\pm0.42$	47.44	$57.26\pm2.79$	
Artemisia absinthium	$6.37\pm0.31$	22.14 ± 1.11	$2.41\pm0.12$	$3.02\pm0.17$	ND	33.94	43.41 ± 2.16	$4.62\pm0.22$	48.03	$57.22\pm2.73$	
Artemisia verlotiorum	$19.44\pm0.97$	$5.43\pm0.41$	$34.85 \pm 1.57$	$0.80\pm0.11$	ND	60.52	$67.78\pm3.87$	$2.43\pm0.17$	70.21	$72.51\pm4.07$	
Artemisia vulgaris	$4.51\pm0.22$	$18.89\pm0.94$	$4.67\pm0.23$	$4.04\pm0.28$	ND	32.11	52.33 ± 2.55	$2.07\pm0.12$	54.40	$62.27\pm2.97$	
Calendula officinalis	$0.55\pm0.02$	$0.78\pm0.04$	ND	ND	ND	1.33	$7.54\pm0.39$	$5.12\pm0.24$	12.66	$24.97 \pm 1.20$	
Chamaemelum nobile	$4.21\pm0.21$	$13.13\pm0.65$	ND	$2.60\pm0.19$	ND	19.94	$43.29\pm2.27$	$22.33 \pm 1.11$	65.62	$64.08\pm2.94$	
Chamomilla recutita	$1.16\pm0.06$	$2.92\pm0.14$	ND	$1.61\pm0.12$	ND	5.69	$15.89\pm0.76$	$9.48\pm0.46$	25.37	$36.79 \pm 1.74$	
Cichorium intybus	$3.34\pm0.16$	ND	ND	ND	38.67 ± 1.90	42.01	$61.07\pm2.97$	$5.07\pm0.25$	66.14	$63.22\pm3.24$	
Conyza canadensis	$5.29\pm0.26$	$14.55\pm0.72$	$1.89\pm009$	$6.05\pm0.38$	ND	27.78	$60.05\pm2.98$	$10.34\pm0.50$	70.39	$76.41 \pm 3.78$	
Eupatorium cannabinum	$14.67\pm0.73$	22.74 ± 1.13	ND	$4.23\pm0.23$	ND	41.64	$65.72\pm3.37$	$8.10\pm0.41$	73.82	$81.47\pm3.75$	
Hieracium pilosella	$25.54 \pm 1.27$	24.17 ± 1.21	$10.32\pm0.49$	$2.82\pm0.16$	ND	62.85	$67.68 \pm 3.41$	$12.51\pm0.60$	80.19	$82.44 \pm 4.44$	
Matricaria perforata	$5.23\pm0.26$	$13.72\pm0.68$	$1.66\pm0.86$	$1.86\pm0.15$	ND	22.47	30.83 ± 1.62	23.44 ± 1.13	54.27	$60.85\pm3.11$	
Solidago gigantea	$20.33 \pm 1.01$	$7.56\pm0.38$	ND	$2.65\pm0.21$	ND	30.54	$62.34 \pm 3.10$	$16.17\pm0.79$	78.51	$83.27\pm4.42$	
Solidago virgaurea	$11.45\pm0.57$	$21.78 \pm 1.09$	ND	$4.23\pm0.18$	ND	37.46	57.90 ± 2.77	$6.73\pm0.32$	64.63	$73.76\pm3.72$	
Tanacetum parthenium	$6.45\pm0.32$	30.08 ± 1.49	ND	$5.61\pm0.27$	ND	42.14	57.21 ± 2.81	$13.97\pm0.70$	71.18	81.12 ± 4.29	
Tanacetum vulgare	$4.12\pm0.21$	$11.83\pm0.59$	ND	$3.54\pm0.20$	ND	19.49	41.53 ± 2.07	11.40 ± 0.59	52.93	$65.86 \pm 3.17$	
Taraxacum officinale	$0.84\pm0.04$	ND	ND	ND	$34.08 \pm 1.65$	34.92	53.92 ± 2.80	$1.62\pm0.08$	55.54	$53.59\pm2.47$	

Table 1. Polyphenolic compound levels of the 18 Asteraceae herbs (g/kg on dry matter).

(DCQA) Dicaffeoylquinic acid, (T1) Total caffeoyl derivatives (determined by HPLC), (T2) Total dihydroxycinnamic derivatives (determined by colorimetry), (T3) Total flavonoids (colorimetry), (T4) Total dihydoxycinnamic derivatives + Total flavonoids (calculated by summing), (T5) Total polyphenolic compounds (colorimetry). Data are mean  $\pm$  SD of triplicate analysis. ND, not determined.

about 60 g/kg, A. millefolium, C. intybus, E cannabinum. and T. parthenium, all about 40 g/kg. The HPLC results on total caffeoyl derivatives can be usefully compared with the results obtained by colorimetry with Arnow's reagent for total dihydroxycinnamic derivatives or total phenolic acids. Colorimetry results (T2 in **Table 1**) ranged from 7.54 g/kg (C. officinalis) to 67.78 for A. verlotiorum and 67.68 g/kg for H. pilosella. Most of the 18 Asteraceae species had very high levels of around 50 to 60 g/kg, with an average value of (48.92  $\pm$  17.11) g/kg. A good correlation was found between total phenolic acids determined by colorimetry (T2 in **Table 1**) and the sum (T1 in **Table 1**) of the caffeoyl derivatives determined individually by HPLC for the 18 herbs ( $R^2 = 0.8697$ ) (**Figure 2**). For comparative purposes, we also determined total flavonoids concentrations of the aerial parts of the 18 *Asteraceae* using colorimetry with oxaloboric reagent (T3 in **Table 1**). The values ranged from 1.62 g/kg for *T* officinale to 22.33 and 23.44 g/kg for *C. nobile* and *M. perforata*, respectively. Compared to results for phenolic

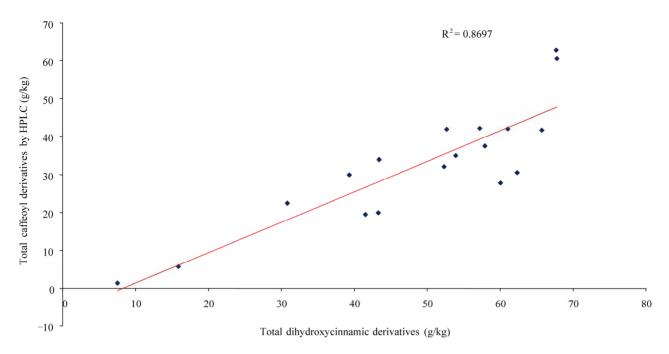


Figure 2. Correlation between total phenolic acids determined by colorimetry (T2 in Table 1) and the sum (T1 in Table 1) of caffeoyl derivatives individually determined by HPLC for the 18 *Asteraceae* herbs.

acid derivatives, the 18 Asteraceae species showed 3-4-fold higher contents of caffeoyl derivatives than flavonoids. Total polyphenols were also determined by colorimetry with Folin reagent (T5 in **Table 1**). All taxa contained high quantities of total polyphenolic compounds, from about 25 g/kg for C. officinalis to more than 80 g/kg for E. cannabinum, H. pilosella, S. gigantea and T. parthe*nium*, with a mean value of  $(64.45 \pm 15.55)$  g/kg. These values of total polyphenols obtained by direct colorimetry (T5) were compared against those calculated by summing the contents of total dihydoxycinnamic derivatives and total flavonoids (T4 = T2 + T3 in **Table 1**). The values were comparable, with an average calculated for all 18 species at (58.72  $\pm$  17.57) g/kg. We found a very good correlation between total phenolics determined by colorimetry (T5) and the sum (T4) of total phenolic acids and total flavonoids for the 18 Asteraceae herbs ( $R^2 = 0.9564$ ) (Figure 3). The presence of other phenolic compounds such as tannins may help explain some differences [48,49]. Besides C. officinalis and C. *recutita*, which were also distinguished by significantly lower levels of total polyphenols, derivatives of phenolic acids and flavonoids represent almost all the polyphenolic compounds for the studied species. As dihydroxycinnamic derivatives are clearly preponderant in relation to flavonoids, these constituents including caffeoyl derivatives can be considered the major phenolic compounds in the studied Asteraceae.

## 3.2. Antioxidant Activity of the 18 Compositae Herbs

Table 2 presents data on the antioxidant activity of the 18 Asteraceae species determined in vitro by DPPH radical scavenging activity. These results are expressed as percentage of total antioxidant capacity for each plant. Species that have the highest antioxidant activities are the richest in polyphenols, while the least active species are less rich in polyphenols, exemplified by C. officinalis and C. recutita at 1.52% and 2.78%, respectively. The remaining 16 species showed relatively homogeneous activities between 5.08% to 11.57%, with an average value of  $8.37\% \pm 1.82\%$  (7.68%  $\pm 2.65\%$  for all species). The two most active plants, S. gigantea and H. pilosella, had an absolute antioxidant activity expressed as EC<sub>50</sub> of  $(44.01 \pm 0.76)$  and  $(44.67 \pm 0.81)$  µg/ml, respectively. Such activities are comparable to figures reported in the literature for Asteraceae species [19,21]. On the whole, the antioxidant activities of the 18 species compared to their different polyphenolic levels showed good correlations between total phenolics (T5 in Table 1) or total dihydroxycinnamic derivatives (T2 in Table 1) and plant antioxidant capacity, *i.e.*  $R^2 = 0.8904$  (Figure 4) and  $R^2 =$ 0.8529 (Figure 5), respectively, and a relatively good correlation between total caffeoyl derivatives determined by HPLC (T1 in **Table 1**) and antioxidant activity, at  $R^2$ = 0.7172 (Figure 6). Determined individually, the antioxidant activities of chlorogenic acid (EC<sub>50</sub> =  $(13.80 \pm$ 

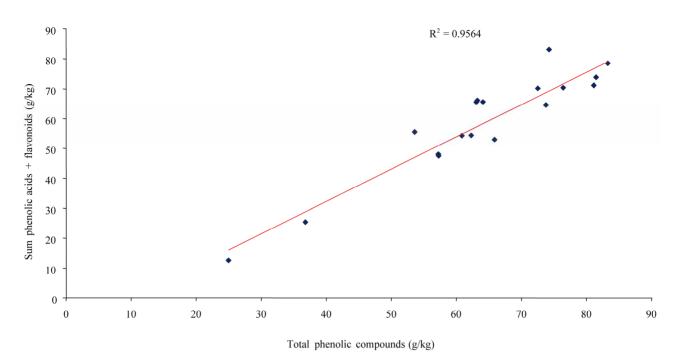


Figure 3. Correlation between total phenolics determined by colorimetry (T5 in Table 1) and the sum (T4 = T2 + T3 in Table 1) of total phenolic acids (T2) and total flavonoids (T3) for the 18 *Asteraceae* herbs.

Table 2. Antioxidant capacity (%) of the 18 Asteraceae herbs and contributions of main caffeoyl derivatives to total antio	-
xidant power.	

	Total capacity (%)	Contribution of main caffeoyl derivatives to total antioxidant power (%)						
Taxon		Chlorogenic Acid	3,5-Dicaffeoyl -quinic acid	1,5-Dicaffeoyl -quinic acid	4,5-Dicaffeoyl -quinic acid	Chicoric acid	Total caffeoyl derivatives (sum)	
Achillea millefolium	$8.29\pm0.16$	10.01	33.17	13.63	4.99	ND	61.80	
Arnica montana	$5.08\pm0.10$	7.28	18.70	29.52	ND	ND	55.20	
Artemisia absinthium	$6.29\pm0.11$	10.33	45.31	4.92	6.08	ND	66.64	
Artemisia verlotiorum	$10.12\pm0.15$	16.41	5.69	36.13	0.81	ND	59.04	
Artemisia vulgaris	$7.22\pm0.14$	6.37	33.51	8.03	7.52	ND	55.43	
Calendula officinalis	$1.52\pm0.02$	3.15	6.90	ND	ND	ND	10.05	
Chamaemelum nobile	$9.20\pm0.17$	4.67	18.04	ND	3.57	ND	26.28	
Chamomilla recutita	$2.78\pm0.05$	4.31	13.66	ND	7.53	ND	25.50	
Cichorium intybus	$8.56\pm0.18$	3.97	ND	ND	ND	58.52	62.49	
Conyza canadensis	$9.63\pm0.18$	5.71	19.52	2.49	8.19	ND	35.91	
Eupatorium cannabinum	$7.53\pm0.13$	19.92	38.24	ND	7.05	ND	65.21	
Hieracium pilosella	$11.24\pm0.20$	23.22	27.40	11.65	3.17	ND	65.44	
Matricaria perforata	$7.53\pm0.14$	7.30	23.50	2.92	3.08	ND	36.80	
Solidago gigantea	$11.57\pm0.20$	18.15	8.38	ND	2.86	ND	29.39	
Solidago virgaurea	$10.04\pm0.17$	11.75	27.68	ND	5.31	ND	44.74	
Tanacetum parthenium	$7.89\pm0.15$	8.49	48.92	ND	9.12	ND	66.53	
Tanacetum vulgare	$7.42\pm0.13$	5.66	19.81	ND	5.88	ND	31.35	
Taraxacum officinale	$6.38\pm0.10$	1.26	ND	ND	ND	68.96	70.22	

Data are mean  $\pm$  SD of triplicate analysis. ND, not determined (compound not determined by HPLC).

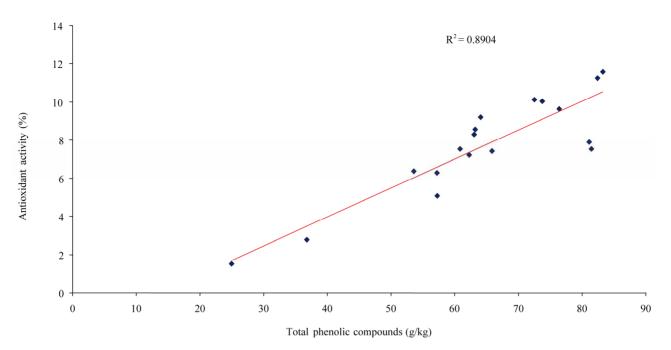


Figure 4. Correlation between total phenolics of the 18 Asteraceae herbs and their antioxidant activity.

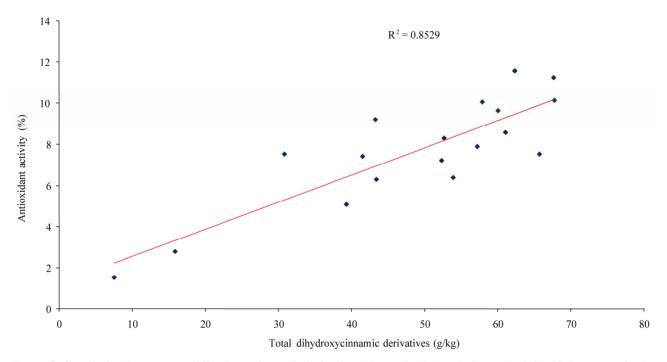


Figure 5. Correlation between total dihydroxycinnamic derivatives (determined by colorimetry) of the 18 Asteraceae herbs and their antioxidant activity.

0.36)  $\mu$ mol/l), 3,5-dicaffeoylquinic acid (EC<sub>50</sub> = (7.62 ± 0.22)  $\mu$ mol/l), 1,5-dicaffeoylquinic acid (EC<sub>50</sub> = (7.85 ± 0.23)  $\mu$ mol/l), 4,5-dicaffeoylquinic acid (EC<sub>50</sub> = (7.99 ± 0.31)  $\mu$ mol/l), and chicoric acid (EC<sub>50</sub> = (8.24 ± 0.27)  $\mu$ mol/l) were comparable to those of L-ascorbic acid (EC<sub>50</sub> = (15.66 ± 0.19)  $\mu$ mol/l) and quercetin (EC<sub>50</sub> =

 $(8.88 \pm 0.21) \mu mol/l)$  and to figures reported in the literature [19,50].

As caffeoyl derivatives were generally the major phenolic compounds of the studied *Asteraceae*, we investigated their contributions to the total antioxidant capacity of each plant. Individual and overall values for the

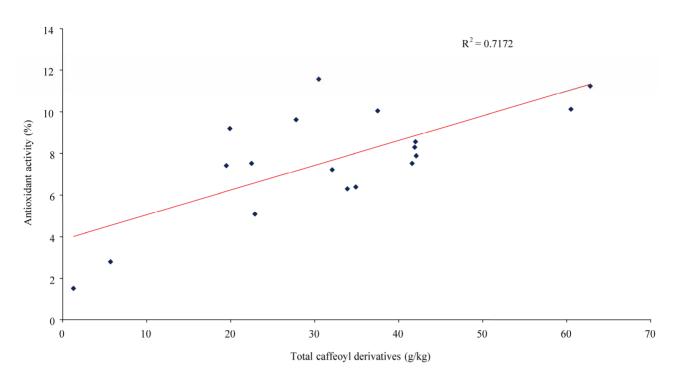


Figure 6. Correlation between total caffeoyl derivatives (determined by HPLC) of the 18 Asteraceae herbs and their antioxidant activity.

5 major caffeovl derivatives are shown in Table 2. Globally, for all 18 species, the mean contributions of chlorogenic acid, 3,5-dicaffeoylquinic acid, 1,5-dicaffeoylquinic acid and 4,5-dicaffeoylquinic acid to antioxidant activity were  $9.33\% \pm 6.25\%$ ,  $21.58\% \pm 14.56\%$ , 6.07% $\pm$  10.68% and 4.18%  $\pm$  3.12%, respectively, varying similarly with the mean levels of these constituents, *i.e.*  $(8.07 \pm 7.29)$ ,  $(13.27 \pm 9.36)$ ,  $(4.25 \pm 8.57)$  and  $(2.59 \pm$ 1.90) g/kg, respectively. Thus, for each species, the major constituents proved to have the largest contributions to antioxidant activity. Chlorogenic acid contributed 18.15 % to the antioxidant activity of S. gigantea  $((20.33 \pm 1.01) \text{ g/kg})$ , 3,5-dicaffeoylquinic acid contributed 48.92% to the antioxidant activity of T. parthenium ( $(30.08 \pm 1.49)$  g/kg) and 45.31% to the antioxidant activity of A. absinthium ((22.14  $\pm$  1.11) g/kg), and 1.5-dicaffeoylquinic contributed 36.13 % to the antioxidant activity of A. verlotiorum ((34.85  $\pm$  1.57) g/kg). The contribution of the 4,5-dicaffeoylquinic acid in smaller quantities is therefore minor. In contrast, chicoric acid, which was found in large amounts in C. intybus ((38.67  $\pm$ 1.90) g/kg) and T. officinale ((34.08  $\pm$  1.65) g/kg) has a very strong contribution, at 58.52% and 68.96%, respectively. For example, the levels of major caffeoyl derivatives of H. pilosella, i.e. chlorogenic acid and 3,5-, 1,5- and 4,5-dicaffeoylquinic acids ((25.54  $\pm$  1.27),  $(24.17 \pm 1.21)$ ,  $(10.32 \pm 0.49)$  and  $(2.82 \pm 0.16)$  g kg<sup>-1</sup>,

respectively) can be insightfully compared to their contributions to total antioxidant activity (23.22%, 27.40%, 11.65% and 3.17%, respectively). For another typical plant, C. intybus, chlorogenic acid and chicoric acid levels were  $(3.34 \pm 0.16)$  and  $(38.67 \pm 1.90)$  g/kg and their contributions to total antioxidant activity were 3.97% and 58.52%, respectively. For each plant, summing the contribution values for caffeoyl derivatives gives the total contribution of these constituents to antioxidant capacity. The values ranged from 10.05% for C. officinalis to 70.22% for T. officinale with an average for the 18 taxa of  $48.22\% \pm 18.37\%$ . For some species, such as C. nobile, the relatively low antioxidant contribution of caffeoyl derivatives (26.28%) can be explained by the presence of larger quantities of flavonoids such as chamaemeloside [13]. For all 18 studied Asteraceae, caffeoyl derivatives determined by HPLC (T1 in Table 1) averaged about 50% (mean 32 vs. 64 g/kg) of total phenolics (T5 in Table 1). Their average contribution to the antioxidant activity of these species is also close to 50%. In addition, for 10 of the 18 plants studied, these compounds contributed over 50%, widely. On the other hand, total dihydroxycinnamic derivatives (T2 in Table 1) averaged about 75% (mean 49 vs. 64 g/kg) of total phenolics (T5 in Table 1). These constituents, comprising the main caffeoyl derivatives, can thus be considered the major antioxidant compounds in the 18 Asteraceae studied.

# 4. Conclusions

This work has clarified the polyphenolic composition of wild herbs of *Asteraceae* studied, and has provided the first evidence of the presence of dicaffeoylquinic derivatives in some of these species. Highly active constituents such as chicoric acid or 3,5-dicaffeoylquinic acid which were present in large amounts in most species contribute very significantly to the antioxidant capacity of these medicinal or food plants. We conclude that, ahead of flavonoids, the main caffeoyl derivatives among polyphenols can be considered the major antioxidant compounds in the aerial parts of studied *Asteraceae*. The search for such active compounds could be extended to many other medicinal or food plants of the broad *Asteraceae* family.

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