

# Dynamic Performance of Fuel Cell Power Module for Mobility Applications

Bouziane Mahmah<sup>1</sup>, Ghania Morsli<sup>1</sup>, Mounia Belacel<sup>1</sup>, Hocine Benmoussa<sup>2</sup>, Soulef Achachera<sup>3</sup>,  
Amina Benhamou<sup>4</sup>, Maïouf Belhamel<sup>1</sup>

<sup>1</sup>Hydrogen Renewable Energy, Centre de Développement des Energies Renouvelables (CDER),  
Algiers, Algeria

<sup>2</sup>Mechanical Engineering Department, Batna University, Batna, Algeria

<sup>3</sup>Photovoltaic Research Division, Algerian Centre for Renewable Energy Development (CDER),  
Algiers, Algeria

<sup>4</sup>Motor's Dynamic and Vibroacoustic Laboratory, University of Boumerdes, Boumerdes, Algeria  
Email: mah2bouziane@gmail.com, mah2@cder.dz

Received October 31, 2012; revised December 10, 2012; accepted December 25, 2012

## ABSTRACT

Fuel cell powered vehicles have been developed as another alternative to internal combustion engine powered vehicles for some applications including passenger cars, buses, trains, motorcycles, forklifts, electric wheelchairs, electric trolleybuses, medical carts, military engines, personal sports craft, mobility devices and other self propelled equipment. Up to now, many researches have focused on the development of the power module in the Fuel cell vehicles (FCVs) and the components of these systems such as membranes, bipolar plates, and electrodes. However, our work in this study focuses on operating the integrated fuel cell power module system efficiently for various operating conditions such as pressure, relative humidity and operating modes. In our validation we have utilized PEMFC single cell, with active area geometry 16 cm<sup>2</sup> and of 120 cm<sup>2</sup>. Some results obtained in our study shown significant performance indicators for PEMFC stack (composed of 2 cells and 4 cells in a series) at different humidification levels.

**Keywords:** Fuel Cell Power Module; Mobility Applications; PEMFC Stack Voltage; Experimental Validation

## 1. Introduction

Fuel cells (FC) are electrochemical devices that convert the chemical energy of the fuel (that is, hydrogen, methanol, and methane) directly into electrical and thermal energy at higher efficiency than internal combustion engines. Fuel cell technologies have received much attention in recent years owing to their high efficiencies, low emissions, and flexible modular structure, compared with conventional power generation systems [1,2].

Since the invention of the very first “gas voltaic battery” by Robert Grove in 1839 [3], several different fuel cell types, which is differentiated by the electrolyte used and the operating temperature, are under investigation and have been developed [4]. The fuel cell became a real option for a wider application base in the late 1980s and early 1990s, although interest efforts increased in fuel cell systems for specific applications [5]. Currently, new Research & Development Programs focusing on large scale testing and on market introduction of fuel cell technology especially in the transport sector, can open an opportunity for fuel cell vehicles (FCVs) in the future [6].

“Over the last five years there has been a twenty-fold increase in shipments of fuel cells with year-on-year growth in both units and megawatts shipped. In 2010, total shipments of fuel cells grew by 40% compared with the previous year, approaching a new high of 230,000 units. Portable fuel cells accounted for 95% of this total but there was substantial growth in other sectors. Over 97% of fuel cells sold worldwide in 2010 used proton exchange membrane fuel cell (PEMFC) technology, and most were hydrogen-fuelled. Europe has been the leading region of adoption for fuel cells since 2009, followed by North America and Asia (including Japan), with all four regions (including the Rest of the World) seeing substantial increases in shipments over that time” [7].

When looking in to the future potential of fuel cell technology in the automotive sector one should be aware of the technological hurdles that have still to be overcome. Large scale demonstration programs are necessary to show and improve fuel cell technology for everyday use. In addition to these programs a lot of basic research on various topics such as material research (membranes, catalysts, bipolar plates...), stack design, system integra-

tion, control strategies, hydrogen storage materials and many more will be necessary [6].

The performance of fuel cell power module in mobility applications is affected with many additional factors (membrane humidification, cells temperature, operation pressure, reactant gas flow rate, impurities...). In this vision the explanation of the effects of operating conditions on the specific ability the cell active surface area, and their relationship with the PEM fuel cell stack performance, reported in terms of its power output per  $\text{cm}^2$  of cell area, is not so widely presented in the scientific literature. Although, Srinivasan *et al.* (1988) have presented the correlation of the electrochemically active surface area with the observed cell potential-current density and also with the values of differential resistances of the cells in the linear regions [8]. After them, Carrette *et al.* (2001) have tried to conclude that an increase in electrochemically active surface area indicates an the increase in fuel cell performance [9], then Meyers *et al.* (2002) have expressed that the geometry of the cell must allow for sufficient surface area of the anode and cathode to facilitate the necessary reactions, as the kinetic losses are a function of current density, so enhancing the surface area of the more sluggish electrode leads to improve its performance. However, one can increase the surface area of the slower electrode, thereby attaining the same total current for a lower current density and, hence, lowering kinetic losses for that electrode [10], Larminie and Dicks (2003) have recognized the very importance of this question, like this, the electrode area is such a vital issue that the performance of a fuel cell design is often quoted in terms of the current per  $\text{cm}^2$  [11]. After this period, the occupation of scientific research is focused on the relationship between the regression in the active surface area and loading of the noble metal catalysts, Yamafuku *et al.* (2004) have shown that the decrease of mass activity seems to be caused by the decrease of active surface area of platinum particle with the crystal growth of platinum particle at higher concentration, they were noticed as a decrease in effective surface area of roughly 60% was found under drying conditions, and a decrease of roughly 40% was found under flooding conditions. The observed decrease in the interfacial capacitance can be converted into an equivalent decrease in the electrochemical active surface area with time, which could be due to the slow approach to steady-state operation [12]. In 2006, Dicks concluded that the ideal support should have a high surface area, good electrical properties, and high electrochemical stability under harsh conditions and the electrocatalyst should be easily accessible to the reactants [13]. Florian Finsterwalder (2007) mentioned that, in order to minimize the surface area loss under potentiodynamic operation the cell voltage should for this type of catalyst be kept within a small window and below ca.

850 mV [14], Dhathathreyan *et al.* (2007) have talk about vital importance of the structure optimisation of membrane and electrode assembly (MEA), as the heart of the PEMFC, to minimize all forms of over potential and maximize the power density, minimize the noble metal loading by high utilization of the surface areas, for effective thermal and water management and to attain lifetimes of PEMFCs [15], Lu *et al.* (2008) they noticed that the activity of the catalyst supported on a substrate depends on its large active surface area [16]. Colmenares *et al.* (2009) they noticed that a significant decrease of the catalyst active surface area, in turn provoked a reduction of the performance and operation lifetime of the PEFC, which can aggressively attack and oxidize carbon support and Nafion membrane, causing a rapid deterioration of the MEA [17]. Li *et al.* (2009) they noticed that in region of low current density, the cell performance decreases with increase in the polytetrafluoroethylene content from 0wt% to 30wt%. The reason for this behaviour is related to the active surface area decreases [18]. In 2010, the Platinum Metals Rev., have recognized that the particle size increases with Pt loading leading to a decrease in the electrochemically active surface area per unit mass of Pt. This suboptimal response to increased loading creates challenges for catalyst synthesis [7]. Nart *et al.* (2009) have noticed that in catalysis, the important parameter is the active surface area, which is given by the active material [19]. Than Ticianelli *et al.* (2010) have renowned that for such applications, in order to provide the best performance with minimum content of platinum, the most important metallic material employed for the catalysis, one  $\text{cm}^2$  of the active surface area contain a value of 2 mg of the noble metal catalyst load, can be maximised for supporting about 20 mg of Pt [20] and Ünlü *et al.* (2010) have noticed that it is more difficult to determine the true surface area in a particular operating fuel cell than the superficial area, even they observed the decrease in the active area by flooding due to faster water generation at high current densities and the charge-transfer resistance consists of a purely kinetic term and is inversely proportional to the active surface area [21]. Ho *et al.* (2011) have revealed a new catalyst support, featuring can give a progress: a high surface area and high proton conductivity, efficiency and durability of polymer electrolyte membrane fuel cells [22], Zakrisson (2011) reported that the platinum is expensive and contributes to a large extent to the high price of PEM fuel cells, which makes it desirable to use the platinum as effective as possible. This fact results in small particles supported in a way to maximise the active surface area of the platinum particles [23], Elam *et al.* (2011) have exposed the results in a fivefold increase in the electrochemically active surface area of the solid oxide fuel cells and yielded an increase in the peak power density of the fuel cells to 677

mW/cm<sup>2</sup> and 861 mW/cm<sup>2</sup> at 400°C and 450°C, respectively [24] and in the rapport (2012) of Manufacturing Fuel Cell Manhattan Project, it is noticed that the geometry of the cell must allow for sufficient surface area of the anode and cathode to facilitate the necessary reaction, and the current designs need to be examined for utilization of membrane surface area. There is a significant opportunity to reduce the membrane surface area by designing and assembling the modules for maximum utilization of the available membrane area. Manufacturers of fuel cell systems have not invested the time and effort necessary to conduct statistically significant studies and generate empirical performance data that can be used to develop optimal designs for these modules. The limiting factor for performance is the total surface area on the air side of the heat exchanger since the heat transfer coefficient is much lower on the air side than the coolant side. A folded fin structure that is brazed to the coolant loop is typically used to maximize available air side heat transfer area, much like a radiator for an automobile. They examined different ways; among them for the counting to minimize scrap, one based on minimization of the surface area in the system available for skinning [25]. In 2012, Spornjak *et al.* they noticed, in the absence of a microporous layer, higher water content during potential holds will cause a faster degradation of the cathode catalyst layer, higher kinetic losses and faster decrease of the active surface area [26] and Nishimura *et al.* (2012) have noticed that particularly, at the time of stopping the fuel cell, if the fuel cell voltage is maintained at 0.8 volt per unit cell or higher under a high temperature state, sintering of catalyst takes place to gather catalyst particles thereby to decrease an active surface area and lower a fuel cell voltage. In order to avoid the problem, it is necessary to remove hydrogen and oxygen remaining in the fuel cell [27].

So our contribution is based in an experimental work, with coupling influences of operating conditions (humidification and oxygen pressure) and the increase of the cell active surface area of the PEMFC stack, in order to produce more-efficient fuel cells module.

## 2. Fuel Cell Vehicles Configurations

A fuel cell system vehicle can be represented with three possible configurations [28]:

- Fuel Cell System Hybrid Fuel Cell Vehicles (Fuel Cell and Rechargeable Electric Storage): The hybrid power train consists of a fuel cell system, a storage device for electrical energy, power electronics, and the electric machine at a minimum.
- Fuel Cells for Auxiliary Power.
- Pure Fuel Cell Vehicles (FCVs): The system of fuel cell is the only source of energy, and that can cope with the dynamic power demands of the vehicle.

In this work, our study has opted for the choice of a design of the third configuration of “Pure Fuel Cell Vehicles”, and based on the following baseline studies [29-31]: SAE J 2615 Reference test procedures of fuel cell systems for automotive application, SAE J 2616 Performance test procedures for the fuel processor systems for automotive application, and SAE J 2617 Performance test procedures of PEM fuel cell stack subsystem for automotive application.

The Fuel Cell Vehicles (FCVs) require technologically advanced components not found in today’s conventional vehicles to operate. The major components of typical FCVs include (fuel cell system power module, Power electronics, Electric machines, Energy storage system, and Hydrogen storage system), see **Figure 1**. The most apparent difference is the fuel cell system power module, but many other components are needed for a complete and functional system [33].

## 3. Fuel Cell Power Module

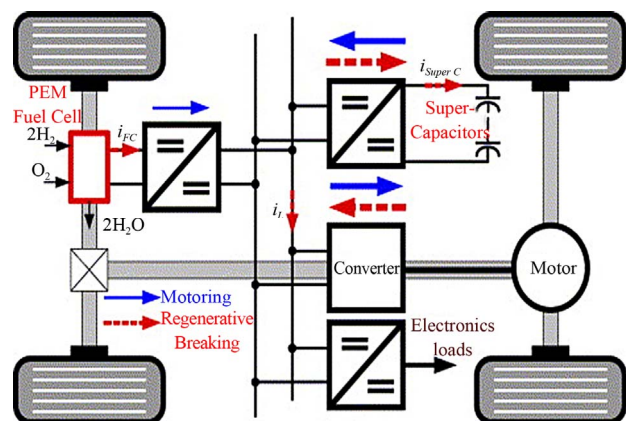
The Fuel cell power module’s, which produces electric power from hydrogen and air, physical, mechanical, and electrical performance characteristics are (Voltage, Current, Power, System Efficiency, Coolant Temperature, Weight, Volume). This power module, shown in **Figure 2**, consists of three subsystems: stack modules, air delivery, and cooling. The support systems for the stacks, such as the air system, water management, and cooling system.

Our objective, in this work, is the testing of fuel cell power module.

## 4. Experimentation Description: Materials and Methods

### 4.1. Experimental System

Two types of PEM fuel cells were used for all experiments in this study:



**Figure 1.** Fuel cell power module in the power train for a hybrid fuel cell vehicle [32].

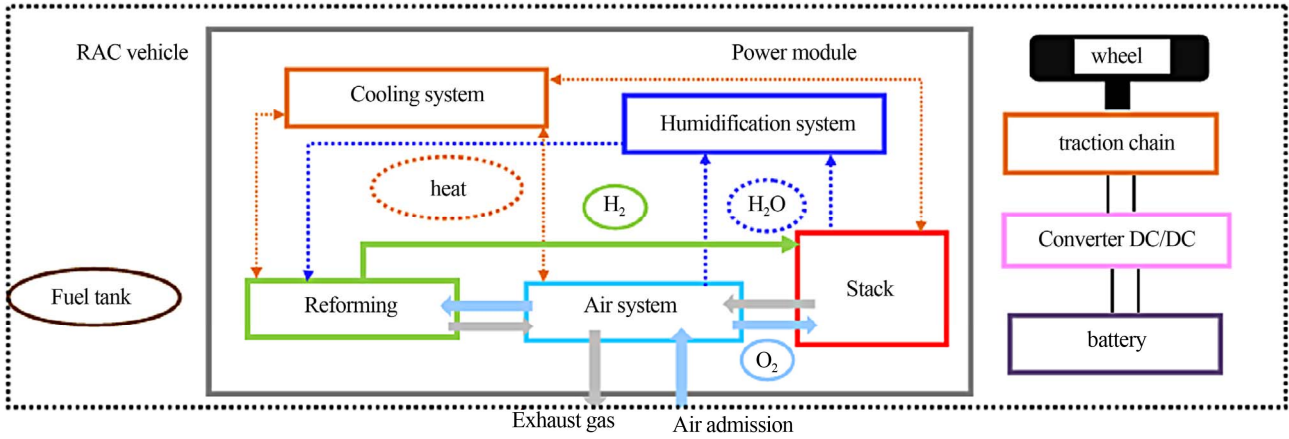


Figure 2. Block diagram of the power module [34].

### 4.2. Experimental Procedure

The laboratory PEM fuel cell stack workbench, shown in Figure 3 can control the fuel cell temperature, humidification, temperatures and pressures on both the anode and cathode sides through a computer connected with Omega OM-3000 series portable data logger, that is make taking measurements and downloading them to our PC, so that with the T6V4 module allows up to six channels of temperature measurement ranging from (-250°C) to (+1370°C) depending on the thermocouple type, relative humidity (RH) and up to four channels of voltage measurement with user-selectable. In the platform tested, reactant gases are humidified by passing through external water tanks. Regulating the water temperature controls the humidification of the reactant gases and the pressures are controlled by backpressure regulators.

## 5. Modeling

### 5.1. Mathematical Cell Model

#### 5.1.1. Ideal Cell Model

The ideal performance of a fuel cell depends on the electrochemical reactions that occur with different fuels. It is defined by its Nernst potential, and represented as cell voltage. In reality, the output voltage of the fuel cell is defined as the function of the thermodynamic potential, the concentration overvoltage, the activation overvoltage, and the ohmic overvoltage. The fuel cell voltage is represented with many models.

#### 5.1.2. Empirical Cell Model

In our work [35-38] based mainly on three research works of Amphlett *et al.*, (1993, 1994, 1996, 1997, 2000 and 2002) [39-45] developed a semi-empirical model for the Ballard Mark cell. Their model combines performance losses in parametric equations based on cell operating conditions and incorporated the voltage degradation into the generalized steady-state model (GSSEM) by

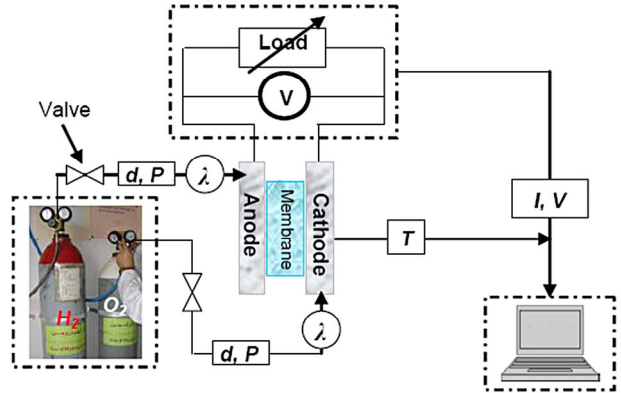


Figure 3. Schematic diagram of the experimental device (P: pressure, d: gas flow, λ: humidification) [Hydrogen Laboratory; CDER—Algeria, 2011].

expressing the change of model parameters in semi-empirical equations. The output voltage expression of a single cell can be written:

$$E_{PAC} = E_{Nernst} - \eta_{act} - \eta_{ohm} - \eta_{conc} \quad (1)$$

The reversible voltage of the cell ( $E_{Nernst}$ ),

$$E_{Nernst} = 1.229 - 0.85 \times 10^{-3} \cdot (T - 298.15) + 4.31 \times 10^{-5} \cdot T \cdot \left[ \ln(P_{H_2}^*) + \frac{1}{2} \ln(P_{O_2}^*) \right] \quad (2)$$

The activation voltage loss,

$$\eta_{act} = - \left[ \xi_1 + \xi_2 \cdot T + \xi_3 \cdot T \cdot \ln(C_{O_2}^*) + \xi_4 \cdot T \cdot \ln(I_{PAC}) \right] \quad (3)$$

The ohmic voltage loss,

$$\eta_{ohm} = I_{PAC} (R_M + R_C) \quad (4)$$

where  $R_C$  is the equivalent contact resistance to the electrons conduction and  $R_M$  the equivalent membrane resistance to the protons conduction. This resistance is calculated from the relation,

$$R_M = \frac{r_M \cdot l}{A} \tag{5}$$

where ( $l$ ) is the membrane thickness (cm), ( $A$ ) is the active surface of the fuel cell (cm<sup>2</sup>) and  $r_M$  is the membrane specific resistivity for the flow of hydrated protons ( $\Omega \cdot \text{cm}$ ),  $r_M$  can be obtained by the following expression:

$$r_M = \frac{181.6 \cdot \left[ 1 + 0.03 \cdot \left( \frac{I_{PAC}}{A} \right) + 0.062 \cdot \left( \frac{T}{303} \right)^2 \cdot \left( \frac{I_{PAC}}{A} \right)^{2.5} \right]}{\left[ \lambda - 0.634 - 3 \cdot \left( \frac{I_{PAC}}{A} \right) \right] \cdot \exp \left[ 4.18 \left( \frac{T - 303}{T} \right) \right]} \tag{6}$$

The term  $\frac{181.6}{\lambda - 0.634}$  represents the specific resistivity ( $\Omega \cdot \text{cm}$ ) at  $I_{PAC} = 0$  and the temperature  $T = 30^\circ\text{C}$ , where ( $T$ ) is the cell absolute temperature in Kelvin degrees. From Springer *et al.* (1991-1993) [46-48] the ( $\lambda$ ) term represents an empirical correction to the specific resistivity to bring in two other factors that affect the average water content of the membrane, protonic current density and cell temperature, and the parameter ( $\lambda$ ) can have a value as high as 14 under ideal, 100% relative humidity conditions and has had reported values as high as 22 and 23 under supersaturated conditions.

From the expressions (4) to (6), the ohmic polarisation can be written as (see the Equation (7) below),

### 5.1.3. Polarization Global Expression

The detailed of the global polarization expression de-

scribed the electrochemical model with essential parametric empirical coefficients, which is represented by the output voltage expression of a single cell in Equation (1), is as follows (see the Equation (8) below);

And the voltage of the module is obtained by the multiplication of the number of cells by the voltage of a single cell;

$$E_{FC} = n \cdot E_{cFC} \tag{9}$$

## 5.2. Predicting Performance Model

The mathematical model proposed to modeling the PEMFC Power Module of Fuel Cell Vehicles is based on the electrochemical work [35-38] which present a modified version of a semi-empirical model developed by Mann *et al.*, (2000), description of these input parameters is presented in the **Table 1** [45] and Amphlett *et al.*, (1993) [40]. That they have developed a practical flexible model of prediction to calculate the cell voltage output in terms of complex relationships between current, stack temperature, and inlet partial pressure of hydrogen and oxygen. In the modified version we include a new value of empirical coefficient in the formulation of the activation polarization related to the specific physical model of the laboratory.

The model utilizes an iterative computer solution which could calculate any variable in terms of the others. The use of this model can make it possible to determine and to achieve various levels of power required for different vehicle power supplies, in different applications of automobiles (buses, locomotives, ships, submarines, and unmanned underwater vehicles) with large power supplies of 3 - 3000 kW.

$$\eta_{ohm} = \frac{I_{PAC}}{A} \left( \frac{181.6 \cdot \left[ 1 + 0.03 \cdot \left( \frac{I_{PAC}}{A} \right) + 0.062 \cdot \left( \frac{T}{303} \right)^2 \cdot \left( \frac{I_{PAC}}{A} \right)^{2.5} \right]}{\left[ \lambda - 0.634 - 3 \cdot \left( \frac{I_{PAC}}{A} \right) \right] \cdot \exp \left[ 4.18 \left( \frac{T - 303}{T} \right) \right]} \cdot L + A \cdot R_C \right) \tag{7}$$

$$E_{cFC} = 2.282 - 0.85 \times 10^{-3} (T - 298.15) + 4.031 \times 10^{-5} \cdot T \left[ \ln \left( \frac{1}{2} P_{anod} \right) + \frac{1}{2} \ln (P_{cath}) \right] - 2.86 \times 10^{-3} \cdot T$$

$$- \left[ 2 \times 10^{-3} \cdot \ln (A) + 4.3 \times 10^{-5} \cdot \ln \left( \frac{\frac{1}{2} P_{anod}}{1.09 \times 10^5 e^{\left( \frac{77}{T} \right)}} \right) \right] \cdot T - 7.8 \times 10^{-5} \cdot T \cdot \ln \left( \frac{P_{cath}}{5.08 \times 10^6 \cdot e^{-\left( \frac{498}{T} \right)}} \right) \tag{8}$$

$$+ 1.97 \times 10^{-4} \cdot T \cdot \ln (I_{FC}) - I_{FC} \left[ \frac{l}{A} - \frac{181.6 \cdot \left[ 1 + 0.03 \cdot \left( \frac{I_{FC}}{A} \right) + 0.062 \cdot \left( \frac{T}{303} \right)^2 \cdot \left( \frac{I_{FC}}{A} \right)^{2.5} \right]}{\left[ \lambda_{\frac{H_2O}{SO_3}} - 0.634 - 3 \cdot \left( \frac{I_{FC}}{A} \right) \right] \cdot \exp \left[ 4.18 \left( \frac{T - 303}{T} \right) \right]} + R_C \right] + B \left( 1 - \frac{J}{J_{max}} \right)$$



**Table 1. Parameters of Mann electrochemical model.**

Mann (2000)	
$R_M = \frac{r_M \cdot l}{A}$	
Ballard Mark—IV	Ballard Mark—V
$\xi_1 = -0.9514$	$\xi_1 = -0.944$
$\xi_2 = 3.12 \times 10^{-3}$	$\xi_2 = 3.54 \times 10^{-3}$
$\xi_3 = 0.74 \times 10^{-4}$	$\xi_3 = 0.78 \times 10^{-4}$
$\xi_4 = -1.87 \times 10^{-4}$	$\xi_4 = -1.96 \times 10^{-4}$
Model: GSSEM	
Generalised Steady-State Electrochemical Model	
$\xi_1 = -0.948 \pm 0.004$	
$\xi_2 = 0.00286 + 0.0002(\ln A) + (4.3 \times 10^{-5}) \ln C_{H_2}^*$	
$\xi_3 = (7.6 \pm 0.2) \times 10^{-5}$	
$\xi_4 = -(1.93 \pm 0.05) \times 10^{-4}$	

**6. Results and Discussion**

The water uptake ratio ( $\lambda [H_2O/SO_3^-]$ ) of Nafion 117 membrane as a function of relative humidity RH [49-51], is shown in **Figure 4**.

The produced **Figures 5** and **6** will show the I-V characteristic for two typical cells of PEMFC according to the active area respectively 16 cm<sup>2</sup> and 120 cm<sup>2</sup>.

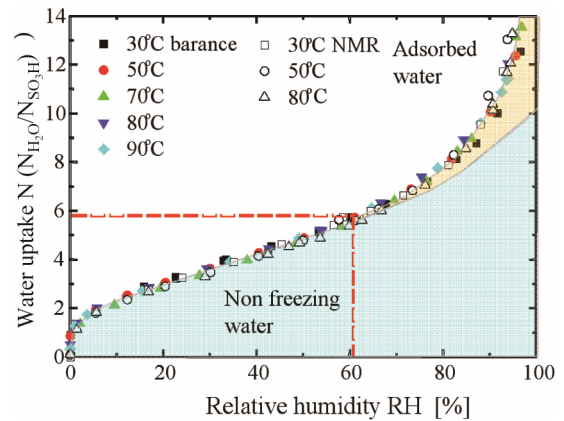
The I-V characteristic presented in **Figure 6** according to the active area of 120 cm<sup>2</sup>, showing a good performance with the different water uptake ratio ( $\lambda$ ), so we went opted for the assembly of cells with this surface.

Therefore **Figures 7** and **8** show polarization curves obtained with two and four cells assembled in series, but we observed an anomaly with the low humidification (water uptake ratio ( $\lambda$ ) = 1 and  $\lambda$  = 3), where we applied equal pressure on both sides of the anode and cathode, although for ( $\lambda$  = 23) is validated in the numerical model (Equations (8) and (9)) with input parameters presented in **Table 2**.

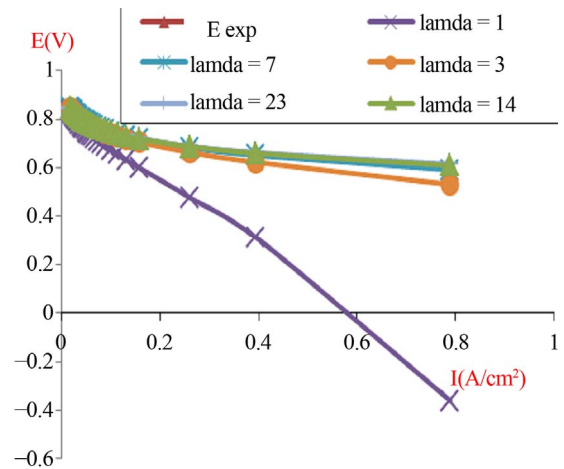
And the power density of the fuel cell module power P (W·cm<sup>-2</sup>) expressed by the formula (P = V × I), composed of 4 cells in series with an active area equal to 120 cm<sup>2</sup>, its variation is represented by the **Figure 9**, Our model presented with his parameters in **Table 3**, with different values of water uptake ratio ( $\lambda$ ), was experimentally validated with various operating conditions presented in **Table 2**, but with ( $\lambda$  = 23) a numerical validation is applied.

To check stability and performance with the optimal value ( $\lambda$  = 14), we have studied in **Figure 10** the pressure-gradient effect between along the two sides of stack consisting of four cells with surface active area equal to 120 cm<sup>2</sup>.

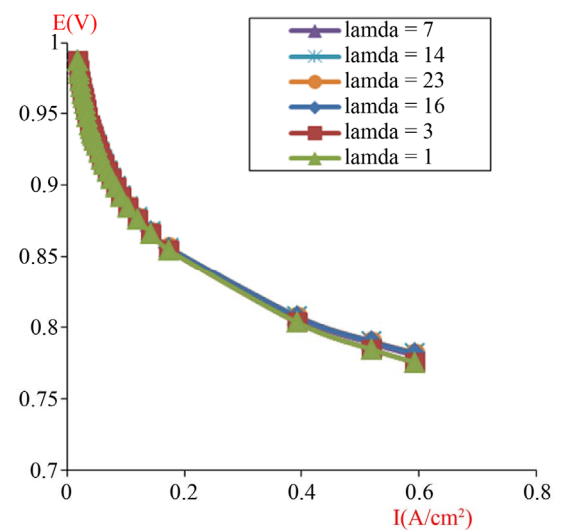
Even with the small active surface area (**Figures 11** and **12**); we have seen an improvement in the PEMFC



**Figure 4. Water uptake ratio ( $\lambda$ ) of Nafion 117 membrane as a function of relative humidity (RH) [49].**



**Figure 5. The simulated I-V characteristics of PEMFC single cell, with active area geometry (A = 16 cm<sup>2</sup>), at different humidification levels (Lambda).**



**Figure 6. The simulated I-V characteristics of PEMFC single cell, with active area geometry (A = 120 cm<sup>2</sup>), at different humidification levels ( $\lambda$ ).**

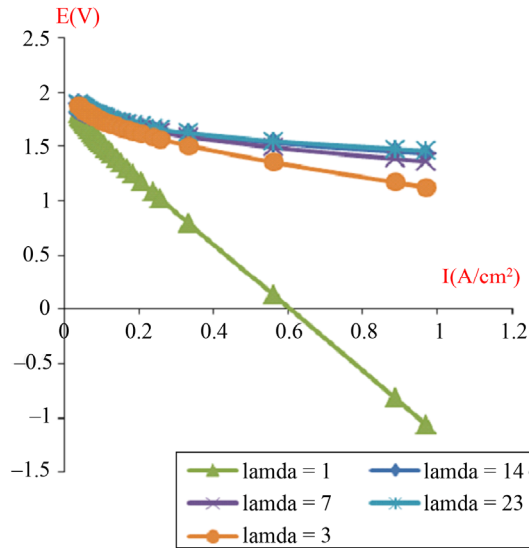


Figure 7. The simulated I-V characteristics of PEMFC stack (composed of 2 cells in a series), with active area geometry ( $A = 120 \text{ cm}^2$ ), at different humidification levels ( $\lambda$ ).

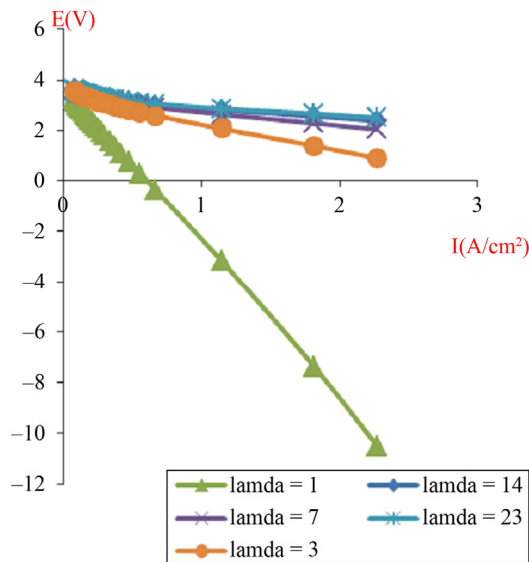


Figure 8. The simulated I-V characteristics of PEMFC stack (composed of 4 cells in a series), with active area geometry ( $A = 120 \text{ cm}^2$ ), at different humidification levels ( $\lambda$ ).

performance with an application of significant pressure at the cathodic side (for oxygen) in comparison with the case of Figure 5.

In addition, we have observed an important influence of oxygen pressure on the performance of the stack; we used the best characteristic value obtained in the experiment as a reference and we have successively increased the pressure in the cathode-outside (Figure 13).

### 7. Conclusions and Outlook

The comparisons and analyses of some results obtained

Table 2. Operating/design parameters of fuel cells.

Parameters	PEM fuel cell 1	PEM fuel cell 2
$A \text{ (cm}^2\text{)}$	16	120
Membrane	Nafion 117	Nafion 117
$l \text{ (membrane thickness) (}\mu\text{m)}$	230	230
$n \text{ (number of cells)}$	4	4
$B \text{ (V)}$	0.016	0.016
$T$	298.15	K
$P_{anod} \text{ (atm)}$	0.01 - 1	0.01 - 1
$P_{cath} \text{ (atm)}$	0.02 - 2	0.02 - 2
$RC \text{ (}\Omega\text{)}$	0.0003	0.0003
$\lambda \text{ (water content)}$ $mol(\text{H}_2\text{O})/mol(\text{SO}_3)$	0 - 23	0 - 23
$\text{H}_2$	N55 $\equiv$ 99.999	N55 $\equiv$ 99.999

Table 3. Parameters of Mahmah electrochemical model.

Mahmah (2012)
PEM <sup>Power1-FuelCell H-TEC</sup>
$A = 16 \text{ cm}^2$
&
Laboratory PEM fuel cell stack workbench development similar of the Ballard Nexa system
$A = 120 \text{ cm}^2$ With Membrane (Nafion 117)
$R_M = \frac{r_M \cdot l}{A}$
$\eta_{con} = -B \left( 1 - \frac{J}{J_{mac}} \right)$
$\xi_1 = -1.053$
$\xi_2 = 0.00286 + 0.0002(\ln A) + (4.3 \times 10^{-5}) \ln C_{H_2}^*$
$\xi_3 = 7.8 \times 10^{-5}$
$\xi_4 = -1.97 \times 10^{-4}$

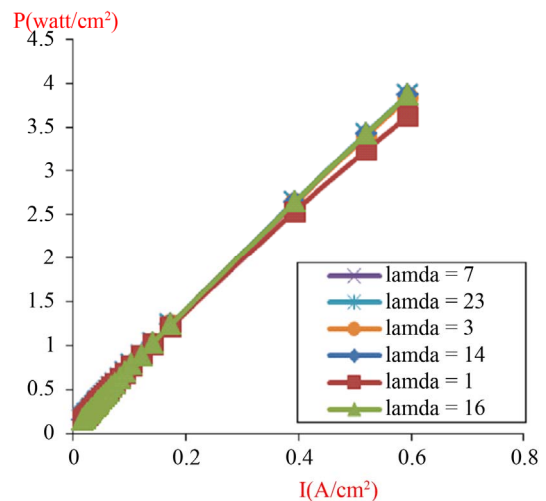


Figure 9. Influence of the humidification levels (Lambda) on the module power (composed of 4 cells in series), with active area ( $A = 120 \text{ cm}^2$ ).

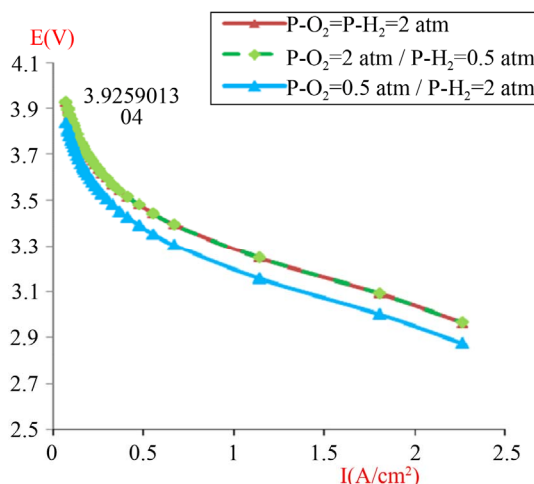


Figure 10. The simulated I-V characteristics of PEMFC stack (composed of 4 cells in a series), with active area geometry ( $A = 120 \text{ cm}^2$ ), with different values of pressure gradient (anode/cathode), and at the humidification level [ $\lambda = 14$ ].

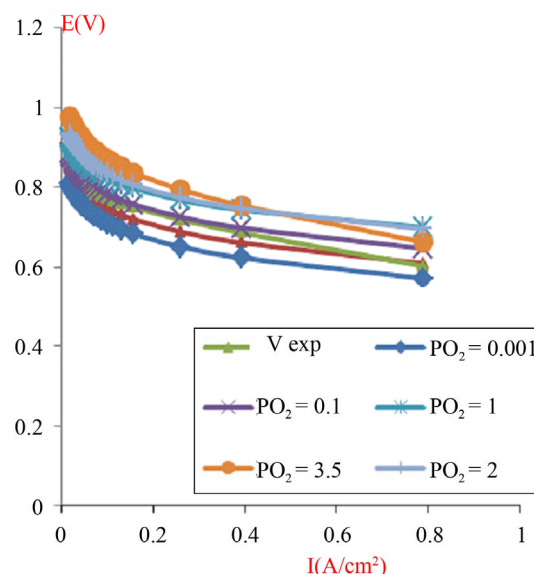


Figure 11. The simulated I-V characteristics of PEMFC single cell, with active area geometry ( $A = 16 \text{ cm}^2$ ), with different values of the cathode pressure, and at the humidification level ( $\lambda = 14$ ).

in our present study for power module in the fuel-cell vehicles previously shown significant performance indicators:

- The advantages of operating PEMFCs at elevated pressures are often debated. The question to pressurize the system is in relation with certain elements (costs, size, weight, etc.). The increase of pressure is helpful to increasing the voltage of the cell, in our case, a motor powered with fuel cell power modules would be functioned with compressed intake air, which would be necessary to supply an adequate

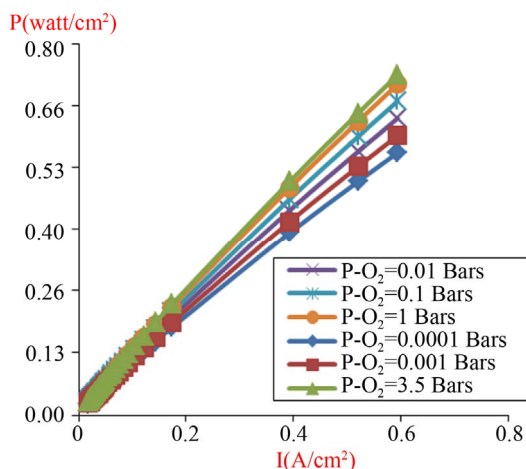


Figure 12. Influence of the pressure gradient (anode/cathode) on the module power of PEMFC single cell, with active area geometry ( $A = 16 \text{ cm}^2$ ), and humidification level ( $\lambda = 14$ ).

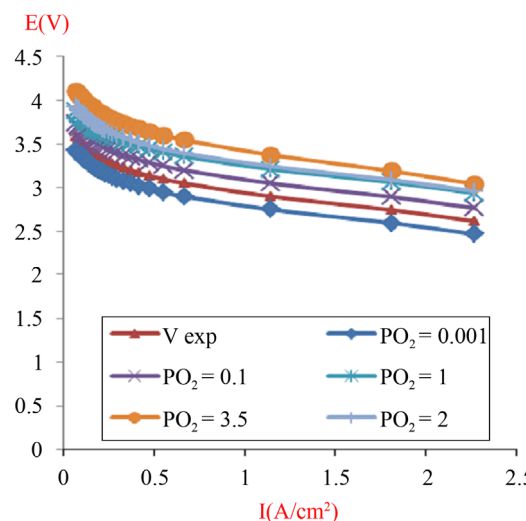


Figure 13. The simulated I-V characteristics of PEMFC stack (composed of 4 cells in a series), with active area geometry ( $A = 120 \text{ cm}^2$ ), with different values of the cathode pressure, and at the humidification level [ $\lambda = 14$ ].

amount of  $\text{O}_2$  and satisfy water concerns.

- The humidification level of the cell is lower, thus leading to lower conductivity. The cell performance is therefore low. The higher humidification levels are best for high loads.
- The output current of a fuel cell is proportional to its active area. Consequently the fuel cell power module geometry can be tailored to provide the desired output voltage, current and power characteristics.
- In the outlook, we are interested to look further into the subject raised by a predictive modeling, enables improved understanding of the most important factors that causes poor performances of PEM fuel cell power modules under different conditions.



## 8. Acknowledgements

This preliminary study was conducted as part of a research project “Solar Fuel”. The authors acknowledge the Center for Renewable Energies Development (CDER), The General Directorate for Scientific Research and Technological Development (DG-RSDT), as well as The National Observatory of the Environment and the Durable Development (ONEDD) and the Ministry of Town and Country Planning and the Environment for their supports.

## REFERENCES

- [1] Y. Wang, K. S. Chen, J. Mishler, S. C. Cho and X. C. Adroher, “A Review of Polymer Electrolyte Membrane Fuel Cells: Technology, Applications, and Needs on Fundamental Research,” *Applied Energy*, Vol. 88, No. 4, 2011, pp. 981-1007. [doi:10.1016/j.apenergy.2010.09.030](https://doi.org/10.1016/j.apenergy.2010.09.030)
- [2] C. Wang and M. H. Nehrir, “Load Transient Mitigation for Stand-Alone Fuel Cell Systems,” *IEEE Transactions on Energy Conversion*, Vol. 22, No. 4, 2007, pp. 887-897.
- [3] W. R. Grove, “On a Gaseous Voltaic Battery,” *Philosophical Magazine and Journal of Science*, Vol. 21, No. 140, 1842, pp. 417-420.
- [4] M. L. Perry and T. F. Fuller, “A Historical Perspective of Fuel Cell Technology in the 20th Century,” *Journal of the Electrochemical Society*, Vol. 149, No. 7, 2002, pp. S59-S67. [doi:10.1149/1.1488651](https://doi.org/10.1149/1.1488651)
- [5] S. M. J. Zaidi, “Research Trends in Polymer Electrolyte Membranes for PEMFC,” In: S. M. Javid Zaidi and T. Matsuura, Eds., *Polymer Membranes for Fuel Cells*, Springer Science + Business Media, LLC., New York, 2009, pp. 7-25. [doi:10.1007/978-0-387-73532-0](https://doi.org/10.1007/978-0-387-73532-0)
- [6] B. Egger, P. H. Dietrich, A. Dorda and A. Pesaran, “Worldwide Promotion and Deployment of Fuel Cell Vehicles,” *24th International Battery, Hybrid and Fuel Cell Electric Vehicle Symposium (EVS24)*, Vol. 3, Stavanger, 13-16 May 2009, pp. 1864-1868.
- [7] J. Matthey, “Catalyst Preparation for the 21st Century: Controlled Catalyst Synthesis to Match form to Function,” *Platinum Metals Review*, Vol. 54, No. 3, 2010, pp. 162-165. [doi:10.1595/147106710X509621](https://doi.org/10.1595/147106710X509621)
- [8] S. Srinivasan, E. A. Ticianelli, C. R. Derouin and A. Redondo, “Advances in Solid Polymer Electrolyte Fuel Cell Technology with Low Platinum Loading Electrodes,” *Journal of Power Sources*, Vol. 22, No. 3-4, 1988, pp. 359-315. [doi:10.1016/0378-7753\(88\)80030-2](https://doi.org/10.1016/0378-7753(88)80030-2)
- [9] L. Carrette, K. A. Friedrich and U. Stimming, “Fuel Cells: Fundamentals and Applications,” *Fuel Cells*, Vol. 1, No. 1, 2001, pp. 5-39. [doi:10.1002/1615-6854\(200105\)1:1<5::AID-FUCE5>3.0.CO;2-G](https://doi.org/10.1002/1615-6854(200105)1:1<5::AID-FUCE5>3.0.CO;2-G)
- [10] J. P. Meyers and H. L. Maynard, “Design Considerations for Miniaturized PEM Fuel Cells,” *Journal of Power Sources*, Vol. 109, No. 1, 2002, pp. 76-88. [doi:10.1016/S0378-7753\(02\)00066-6](https://doi.org/10.1016/S0378-7753(02)00066-6)
- [11] J. Larminie and A. Dicks, “Fuel Cell Systems Explained,” 2nd Edition, John Wiley & Sons Ltd., Chichester, 2003.
- [12] T. Yamafuku, K. Totsuka, S. Hitomi, H. Yasuda and M. Yamachi, “Optimization of Polymer Electrolyte Distribution of Ultra-Low Platinum Loading Electrode for PEFC,” *Technical Report*, Vol. 63, No. 1, 2004, pp. 23-37.
- [13] A. L. Dicks, “The Role of Carbon in Fuel Cells,” *Journal of Power Sources*, Vol. 156, No. 1, 2006, pp. 128-141. [doi:10.1016/j.jpowsour.2006.02.054](https://doi.org/10.1016/j.jpowsour.2006.02.054)
- [14] F. Finsterwalder (DaimlerChrysler AG), “Degradation of the Active Catalyst Surface in Automotive Fuel Cells,” *Proceedings of the 212th Electrochemical Society Meeting*, Vol. 1-3, Washington DC, October 7-12, 2007, p. 400.
- [15] K. S. Dhathathreyan and N. Rajalakshmi, “Polymer Electrolyte Membrane Fuel Cell,” In: S. Basu, Ed., *Recent Trends in Fuel Cell Science and Technology*, Anamaya Publishers, New Delhi, 2007, pp. 40-115. [doi:10.1007/978-0-387-68815-2\\_3](https://doi.org/10.1007/978-0-387-68815-2_3)
- [16] P. Lu and B. Ding, “Applications of Electrospun Fibers,” *Recent Patents on Nanotechnology*, Vol. 2, No. 3, 2008, pp. 169-182. [doi:10.2174/187221008786369688](https://doi.org/10.2174/187221008786369688)
- [17] L. C. Colmenares, A. Wurth, Z. Jusys and R. J. Behm, “Model Study on the Stability of Carbon Support Materials under Polymer Electrolyte Fuel Cell Cathode Operation Conditions,” *Journal of Power Sources*, Vol. 190, No. 1, 2009, pp. 14-24. [doi:10.1016/j.jpowsour.2009.01.078](https://doi.org/10.1016/j.jpowsour.2009.01.078)
- [18] Y. S. Li, T. S. Zhao and Z. X. Liang, “Effect of Polymer Binders in Anode Catalyst Layer on Performance of Alkaline Direct Ethanol Fuel Cells,” *Journal of Power Sources*, Vol. 190, No. 2, 2009, pp. 223-229. [doi:10.1016/j.jpowsour.2009.01.055](https://doi.org/10.1016/j.jpowsour.2009.01.055)
- [19] F. C. Nart and W. Vielstich, “Normalization of Porous Active Surfaces,” In: W. Vielstich, H. A. Gasteiger, A. Lamm and H. Yokokawa, Eds., *Handbook of Fuel Cells—Fundamentals, Technology and Applications*, Vol. 2, John Wiley & Sons Ltd., Chichester, 2009, pp. 302-315.
- [20] E. A. Ticianelli and E. R. Gonzalez, “Fundamental Kinetics/Transport Processes in MEAs,” In: W. Vielstich, H. A. Gasteiger and A. Lamm, Eds., *Handbook of Fuel Cells—Fundamentals, Technology and Applications*, Vol. 2, John Wiley & Sons Ltd., Chichester, 2003, pp. 490-501.
- [21] M. Ünlü, J. Zhou and P. A. Kohl, “Study of Alkaline Electrodes for Hybrid Polymer Electrolyte Fuel Cells,” *Journal of the Electrochemical Society*, Vol. 157, No. 10, 2010, pp. B1391-B1396. [doi:10.1149/1.3468700](https://doi.org/10.1149/1.3468700)
- [22] V. T. T. Ho, B.-J. Hwang and J.-F. Lee, “New Catalyst Support Improves Efficiency and Durability of Fuel Cells,” NSRRC Activity Report, National Synchrotron Radiation Research Center, Taiwan, 2011, pp. 44-45.
- [23] E. Zakrisson, “The Effect of Start/Stop Strategy on PEM Fuel Cell Degradation Characteristics. Master of Science,” Master of Science Thesis, Chalmers University of Technology, Gothenburg, 2011.
- [24] J. W. Elam, N. P. Dasgupta and F. B. Prinz, “ALD for Clean Energy Conversion, Utilization, and Storage,” *Materials Research Society MRS Bulletin*, Vol. 36, No. 11,

- 2011, pp. 899-906. [doi:10.1557/mrs.2011.265](https://doi.org/10.1557/mrs.2011.265)
- [25] The Benchmarking and Best Practices Center of Excellence (B2PCOE), "Manufacturing Fuel Cell—Manhattan Project," ACI Technologies Inc., Philadelphia, 2012, 283 Pages.
- [26] D. Spornjak, J. D Fairweather, T. Rockward, K. C. Rau, J. S. Spendelow, R. L. Borup, R. Mukundan and D. Hussey, "Influence of *in Situ* and *ex Situ* Aging of Gas Diffusion Layers on Fuel Cell Performance Degradation," *Conference: ECS Meeting*, Honolulu, 7-10 October 2012, pp. 1672-1678.
- [27] K. Nishimura, J. Imahashi, M. Komachiya and K. Takahashi, "Method of Stopping a Solid Polymer Type Fuel Cell System," Patent No. US 8173313, 2012.
- [28] O. Ellabban, J. Van Mierlo, P. Lataire and P. Van den Bossche, "Z-Source Inverter for Vehicular Applications," *Vehicle Power and Propulsion Conference (VPPC)*, Chicago, 6-9 September 2011, pp. 1-6.
- [29] United Nations Economic Commission for Europe, "Informal Documents for the 53rd GRB Session," 15-17 February 2011.  
<http://www.unece.org/trans/main/wp29/wp29wgs/wp29grb/grbinf53.html>
- [30] G. Brusaglino, "Energy and Operation Performance of Hydrogen (H2) Fuelled Vehicles," 16 September 2009.  
<http://www.unece.org/fileadmin/DAM/trans/doc/2009/wp29grpe/H2SGE-04-IP-02e.pdf>
- [31] Industry Standards & Regulations, Retrieved 30 August 2012.  
<http://www.ihs.com/products/industry-standards/org/sae/j-series/page18.aspx>
- [32] P. Thounthong, S. Raël and B. Davat, "Test of a PEM Fuel Cell with Low Voltage Static Converter," *Journal of Power Sources*, Vol. 153, No. 1, 2006, pp. 145-150.  
[doi:10.1016/j.jpowsour.2005.01.025](https://doi.org/10.1016/j.jpowsour.2005.01.025)
- [33] J. Ryu, Y. Park and M. Sunwoo, "Electric Power Train Modeling of a Fuel Cell Hybrid Electric Vehicle and Development of a Power Distribution Algorithm Based on Driving Mode Recognition," *Journal of Power Sources*, Vol. 195, No. 17, 2010, pp. 5735-5748.  
[doi:10.1016/j.jpowsour.2010.03.081](https://doi.org/10.1016/j.jpowsour.2010.03.081)
- [34] N. Romani, "Modélisation et Commande du Système d'Alimentation en Air pour le Module de Puissance d'un Véhicule à Pile à Combustible avec Reformeur Embarqué," Thèse de Doctorat en Physique, Faculté des Sciences d'Orsay, Université Paris-Sud, Orsay, 2007.
- [35] D. Haddad, H. Benmoussa, N. Bourmada, K. Oulmi, B. Mahmah and M. Belhamel, "One Dimensional Transient Numerical Study of the Mass Heat and Charge Transfer in a Proton Exchange Membrane for PEMFC," *International Journal of Hydrogen Energy*, Vol. 34, No. 11, 2009, pp. 5010-5014. [doi:10.1016/j.ijhydene.2008.12.033](https://doi.org/10.1016/j.ijhydene.2008.12.033)
- [36] B. Mahmah, A. M'Rouï, M. Belhamel and H. Benmoussa, "Experimental Study and Modelling of a Fuel Cell PEMFC Fed Directly with Hydrogen/Oxygen," *The 16th World Hydrogen Energy Conference (WHEC 16)*, Lyon, 13-16 June 2006, 10 Pages.
- [37] F. Amrouche, B. Mahmah, M. Belhamel and H. Benmoussa, "Modélisation d'une Pile à Combustible PEMFC Alimentée Directement en Hydrogène-Oxygène et Validation Expérimentale," *Revue des Energies Renouvelables*, Vol. 8, No. 2, 2005, pp. 109-121.
- [38] B. Mahmah, A. M'Rouï, M. Belhamel and H. Benmoussa, "Simulation Numérique Unidimensionnelle du Phénomène de Transfert de Chaleur et de Transport de Masse et Charge dans une PEMFC," *Revue des Energies Renouvelables*, Special Issue, 2003, pp. 103-110.
- [39] M. W. Fowler, R. F. Mann, J. C. Amphlett, B. A. Peppley and P. R. Roberge, "Incorporation of Voltage Degradation into a Generalized Steady State Electrochemical Model for a PEM Fuel Cell," *Journal of Power Sources*, Vol. 106, No. 1-2, 2002, pp. 274-283.  
[doi:10.1016/S0378-7753\(01\)01029-1](https://doi.org/10.1016/S0378-7753(01)01029-1)
- [40] J. C. Amphlett, R. M. Baumert, R. F. Mann, B. A. Peppley and P. R. Roberge, "A Performance Model for PEM Fuel Cells," *Proceedings of the 28th Intersociety Energy Conversion Engineering Conference (IECEC-93)*, Vol. 1, Atlanta, 8-13 August 1993, pp. 1215-1220.
- [41] J. C. Amphlett, R. M. Baumert, R. F. Mann, B. A. Peppley, P. R. Roberge and A. Rodrigues, "Parametric Modelling of the Performance of a 5 kW Proton-Exchange Membrane Fuel Cell Stack," *Journal of Power Sources*, Vol. 49, No. 1-3, 1994, pp. 349-356.  
[doi:10.1016/0378-7753\(93\)01835-6](https://doi.org/10.1016/0378-7753(93)01835-6)
- [42] J. C. Amphlett, R. F. Mann, B. A. Peppley, P. R. Roberge and A. Rodrigues, "A Practical PEM Fuel Cell Model for Simulating Vehicle Power Sources," *Proceedings of the 10th Annual Battery Conference on Applications and Advances*, Long Beach, 10-13 January 1995, pp. 221-226.
- [43] J. C. Amphlett, R. F. Mann, B. A. Peppley, P. R. Roberge and A. Rodrigues, "A Model Predicting Transient Responses of Proton Exchange Membrane Fuel Cells," *Journal of Power Sources*, Vol. 61, No. 1-2, 1996, pp. 183-188. [doi:10.1016/S0378-7753\(96\)02360-9](https://doi.org/10.1016/S0378-7753(96)02360-9)
- [44] J. C. Amphlett, E. H. de Oliveira, R. F. Mann, P. R. Roberge, A. Rodrigues and J. P. Salvador, "Dynamic Interaction of a Proton Exchange Membrane Fuel Cell and a Lead-Acid Battery," *Journal of Power Sources*, Vol. 65, No. 1-2, 1997, pp. 173-178.  
[doi:10.1016/S0378-7753\(97\)02472-5](https://doi.org/10.1016/S0378-7753(97)02472-5)
- [45] R. F. Mann, J. C. Amphlett, M. A. I. Hooper, H. M. Jensen, B. A. Peppley and P. R. Roberge, "Development and Application of a Generalised Steady-State Electrochemical Model for a PEM Fuel Cell," *Journal of Power Sources*, Vol. 86, No. 1-2, 2000, pp. 173-180.  
[doi:10.1016/S0378-7753\(99\)00484-X](https://doi.org/10.1016/S0378-7753(99)00484-X)
- [46] T. E. Springer, T. A. Zawodzinski and S. Gottesfeld, "Polymer Electrolyte Fuel Cell Model," *Journal of The Electrochemical Society*, Vol. 138, No. 8, 1991, pp. 2334-2342. [doi:10.1149/1.2085971](https://doi.org/10.1149/1.2085971)
- [47] T. E. Springer, M. S. Wilson and S. Gottesfeld, "Modeling and Experimental Diagnostics in Polymer Electrolyte Fuel Cells," *Journal of the Electrochemical Society*, Vol. 140, No. 12, 1993, pp. 3513-3526.  
[doi:10.1149/1.2221120](https://doi.org/10.1149/1.2221120)
- [48] T. A. Zawodzinski, T. E. Springer, F. Uribe and S. Gottesfeld, "Characterization of Polymer Electrolytes for

- Fuel-Cell Applications,” *Solid State Ionics*, Vol. 60, No. 1-3, 1993, pp. 199-211. [doi:10.1016/0167-2738\(93\)90295-E](https://doi.org/10.1016/0167-2738(93)90295-E)
- [49] J. Kawamura, N. Kuwata, K. Hattori and J. Mizusaki, “Ionic Transport in Nano-Heterogeneous Structured Materials,” *Reports of the Institute of Fluid Science*, Vol. 19, 2007, pp. 67-72.
- [50] C. Yang, P. Costamagna, S. Srinivasan, J. Benziger and A. B. Bocarsly, “Water Uptake and Conductivity of Composite Membranes Operating at Reduced Relative Humidity,” *Journal of Power Sources*, Vol. 103, No. 1, 2001, pp. 1-9. [doi:10.1016/S0378-7753\(01\)00812-6](https://doi.org/10.1016/S0378-7753(01)00812-6)
- [51] E. M. W. Tsang, Z. Zhang, A. C. C. Yang, Z. Shi, T. J. Peckham, R. Narimani, B. J. Frisken and S. Holdcroft, “Nanostructure, Morphology, and Properties of Fluorous Copolymers Bearing Ionic Grafts,” *Macromolecules*, Vol. 42, No. 24, 2009, pp. 9467-9480. [doi:10.1021/ma901740f](https://doi.org/10.1021/ma901740f)