#### American Journal of Plant Sciences, 2014, 5, 4040-4049

Published Online December 2014 in SciRes. <a href="http://www.scirp.org/journal/aips">http://www.scirp.org/journal/aips</a> <a href="http://dx.doi.org/10.4236/aips.2014.526422">http://dx.doi.org/10.4236/aips.2014.526422</a>



# Leaching Potential of Diuron and Linuron in Gaza Soils

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Received 26 October 2014; revised 21 November 2014; accepted 5 December 2014

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## **Abstract**

This study investigated the leaching potential of diuron and linuron in different soil types in Gaza Strip, Palestine under laboratory and field conditions. Leaching potential was evaluated by tin columns and bioassay technique using Molokhia as test plant. The responses of the test plant were regressed versus concentrations of the herbicide to estimate linear regression equation and the regression coefficient. The obtained results indicate strong positive association between Molokhia and concentrations of diuron or linuron; accordingly it was used in the bioassay and estimation of the concentrations of the tested herbicides in the leaching depth. Leaching potential was very low in North Gaza and Kkan Younis soil, whereas in the Meddle zone soil was very large. Leaching potential under laboratory conditions was larger that under field conditions. The bioestimated concentrations in soil layer under field conditions were sever folds of magnitude lower than those under laboratory conditions. The interesting outcome of this study is that leaching potential is dependent on soil clay fraction, soil organic matter, and soil pH. These results provided answers to the questions raised by farmers in Gaza Strip.

# **Keywords**

Leaching Potential, Diuron, Linuron, Clay Fraction, Soil Organic Matter, Soil pH

## 1. Introduction

Leaching and distribution of herbicides in soil are necessary to achieve proper weed control and to suppress the growth of unwanted plants in the top soil layer (0 - 10 cm). However, movement of herbicides below the root zone may result in loss of herbicidal activity and create problems to plants in the next growth season and con-

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tamination to ground water [1].

Number of studies from Gaza Strip [1]-[6], Europe [7]-[22] Japan [23] and the United States [24]-[27] reported ground water contamination by diuron and/or linuron.

Farmers from Gaza Strip, Palestine claimed losses of herbicidal activity of diuron or linuron and reduction of plant growth following their application. Accordingly, the authors designed this study to investigate the leaching potential of diuron and linuron in different soil types in Gaza Strip and to test the biological responses of different plants to provide answers to the farmers.

# 2. Materials and Methods

Technical diuron and linuron (purity 99%) were purchased from Sigma Aldrich, Germany, commercial emulsifiable formulation of herbicides were purchased from Gaza. The following test plants: Beans (*Phaseolus vulgaris*); mulukhia (*Corochorus olitorius*); Cucumber (*Cucumis sativus*); Peas (*Pisum sativum*) seeds were purchased from certified agricultural products in Gaza Strip.

# 2.1. Soil Sampling

The soil samples were collected from depth of 0 - 30 cm of three agricultural locations in Gaza Governorates (Khan Younis Governorate, Middle Governorate and North Gaza Governorate). The samples were air dried, sieved through 2 mm mesh, transferred to a plastic bags, and storage at the lab until use. The soil properties are shown in [28].

# 2.2. Leaching Potential of Diuron and Linuron

#### 2.2.1. Bioassay Technique

Following the procedure described earlier [1] and modified [29], about 40 mg diuron or linuron were dissolved separately in one liter volumetric flask and used as stock solution to prepare the following concentrations: 0, 0.013, 0.027, 0.055, 0.11, 0.22, 0.44, and 0.88 mg/kg soil. The soils were mixed thoroughly in polyethylene plastic bags to insure complete homogenization and even distribution of diuron or linuron in soil. To avoid losses of herbicides due to possible sorption to the polyethylene bags because of mixing, an appropriate herbicide solution was incorporated in the soil sample, followed by a manual rotary mixing using a glass rod. The treated soils were transferred to five black plastic pots. Each pots have four holes at the bottom covered by tissue paper. Ten seeds of the test plant, were placed in each pot in the lot. The following test plants: beans; Molokhia; cucumber and peas seeds were evaluated for sensitivity to diuron and linuron. Shoot height or fresh weight were determined 2 weeks after treatment, and percent growth inhibition for each test plant was calculated using Equation (1).

% Growth Inhibition = 
$$100(Pc - Pt)/Pc$$
 (1)

where *P*c and *P*t are the shoot height of the control and the treated samples, respectively. The growth inhibition data were regressed versus corresponding concentrations and used to establish the regression equation. This enabled the calculation of LC50. The lowest LC50 value indicates the high sensitivity of the test plant.

The sensitive test plant was used to estimate the leaching potential of the herbicides in soil by estimating the growth inhibition or relative concentration of diuron or linuron at different soil layers using bioassay and column techniques (Figure 1).

## 2.2.2. Leaching Potential of Herbicides at Laboratory Columns

The columns were made from tin with  $10 \text{ cm} \times 10 \text{ cm}$  surface area and 25 cm height. The bottom of the columns contained several micro-holes of 2 mm mesh. The columns were filled with 2 letter of an air-dried, sandy soil, sieved through a 2 mm screen. The soil surfaces were sprayed with the field rate (active ingredient 2.0 kg/ha) of either diuron or linuron commercial EC formulation. The columns were carefully irrigated with 500 m<sup>3</sup>/ha applied in portions over 4 h at 30 min intervals. The columns were left for 48 h for equilibration and sliced along their lengths, hence forming two pots. The pots were used for bioassay in the greenhouse, the test plant (Molokhia) was sown in two rows in each column half. The pots were left for 2 days without irrigation to insure absorption soil solution which may have threshold concentration of diuron or linuron due to potential leaching. The



Figure 1. Columns techniques used for leaching experiments: 1, empty half column; 2, complete column full of soil; 3, special spatula to divide the column; 4, a half column ready for sowing test plants; 5, stoppers [1].

sprinkle irrigation was supplied as needed. Relative concentration of diuron or linuron could be determined at various soil layers by a reduction in tissue dry weight or height of the test plant using Equation [1] and application of a linear regression equation generated from the standard curve Equation (2).

$$Y = aX + c \tag{2}$$

Y represents growth inhibition of the test plant determined by Equation (1), and X represents herbicide concentration (mg/kg soil), a and c are the slope and the intercept of the regression line and are constants related to the experimental conditions.

## 2.2.3. Leaching Behavior in the Field Plots

In the field plots, beds (1 m²) were prepared using a rotary tiller. Emulsifiable concentrate of diuron or linuron at 2.0 kg/ha active ingredient, were applied pre-emergence using a back pack motorized sprayer calibrated to discharge 300 L/ha. A sprinkler irrigation system was used to water the field plots at 500 m³/ha. One soil sample was taken from each plot using the one half of the tin column (**Figure 1**) gently inserted to a depth of 25 cm and supported with a wide spatula to ensure the complete removal of the soil in the column. The soil columns were carefully transferred to the greenhouse for bioassay as described above.

## 2.2.4. Influence of Various Soils on Leaching Potential

Soil samples were collected from different 3 agricultural locations in Gaza Governorate and used for leaching potential. In this technique 3 soil columns represent an area and were sprayed with the herbicide concentrations and irrigated with the required amount of water as mentioned above.

This experiment enabled us to generate information about the possible interaction between herbicide and various soils.

# 2.3. Statistical Analysis

The growth inhibition data was analyzed for variance, and main effects and interactions was tested for significance using repeated measures ANOVA. Univariate comparisons of mean growth inhibition at different depths were performed by T-test ( $\alpha = 0.05$ ), the statistical analysis was performed by using Microsoft Excel software.

## 3. Results and Discussion

#### 3.1. Soil Properties

Properties of soil used in this study are shown in Table 1. It can be seen that pH values of the tested soils ranged

between 7.32 - 8. Furthermore the EC values ranged from 1.81 - 2.63 dS/m, the total organic matter range from 0.801% - 0.254%. This indicates that soil is nearly poor with organic matters. The clay fraction of soils range between 10% - 47%.

## 3.2. Bioassay Studies

Growth inhibition of the test plants are shown in **Figure 2** and the corresponding linear regression equations are shown in **Table 2**. The presented results indicate that the test plant Molokhia, (*Corochorus olitorius*) is the most sensitive one among all, followed by cucumber. The data in **Table 2** indicate strong positive correlation. These data agree with Safi *et al.* [30] who found that Molokhia was sensitive to diuron.

Accordingly Molokhia was used in this study as test plant to determine the leaching potential of herbicides in columns and field plots. Responses of Molokhia to different concentrations of diuron and linuron in different Gaza soils were linear (Data not shown). The equation and the correlation coefficient of the regression lines are shown in **Table 3**.

# 3.3. Leaching Potential in Soil Columns of Diuron

Leaching potential of diuron and linuron in Gaza Soils are shown in **Figure 3**, **Figure 4**. It can be seen that the leaching potential is similar in three soil type and different only in Middle Governorate soil in both cases. However, the only difference in the leaching potential is that the part of the threshold concentration of both herbicides disappeared from Middle Governorate soil and accumulated at deeper depths. Whereas in the other soils the threshold concentration remained in the top soil layer (0 - 5 cm) then considerable amount of diuron and linuron moved down to different depths. More details are obvious in **Figure 3**, **Figure 4**. Nevertheless, statistical

Table 1. Some physic-chemical properties of soils used in the study.

Site	pН	EC dS/m	OM %	TOC %	Sand %	Silt %	Clay %	Texture
KHY Z	8	1.592	0.762	0.442	62.5	17.5	20	Sandy loam
MZ	7.32	1.813	0.254	0.147	87.5	1.25	11.25	Loamy sand
NZ	7.33	2.36	0.571	0.331	60	12.5	27.5	Clay soil
Field exp	7.63	2.044	0.317	0.184	85	5	10	Loamy sand
Clay soil	7.36	2.63	0.801	0.465	23	30	47	Clayey silt

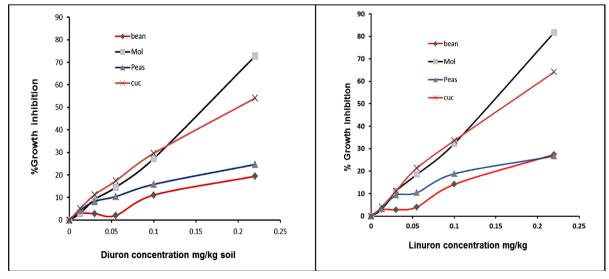


Figure 2. Effects of diuron and linuron concentrations on the growth of the test plants; bean, mul, cuc, and peas represent beans, mulukhia, cucumber and peas respectively.

Table 2. Linear regression equation of the sensitivity plants to herbicide concentrations.

Test plant	Equation	$R^2$
Mulukhiya	Y = 317.46X	0.9896
Cucumber	Y = 259.45X	0.9762
Bean	Y = 126.23X	0.8452
Peas	Y = 89.812X	0.9423

Y and X represent % growth inhibition and concentration respectively.

Table 3. Linear regression equation and regression coefficient of diuron and linuron using Molokhia as test plant.

Area	Diuron		Linuron		
	Equation	$R^2$	Equation	$R^2$	
KHYZ	Y = 233.63X	0.95	Y = 275.73X	0.91	
MZ	Y = 662.07X	0.95	Y = 265.21X	0.98	
NZ	Y = 238.34X	0.93	Y = 229.07X	0.97	

Y and X represent % growth inhibition and concentration respectively.

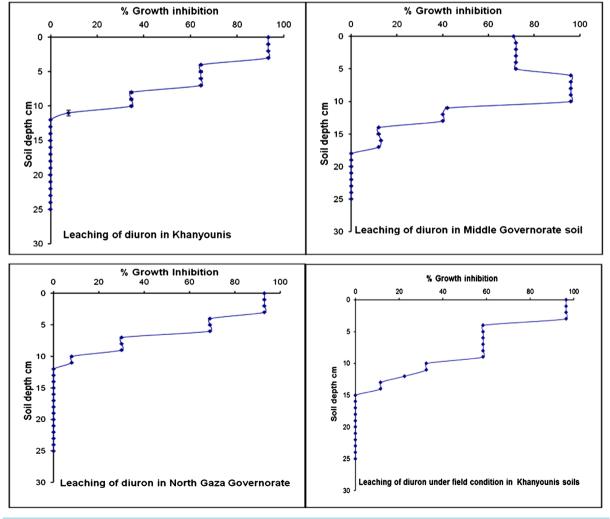


Figure 3. Leaching of diuron in different soils. Error bars represent standard deviation. Using Mulukhia as test plant.

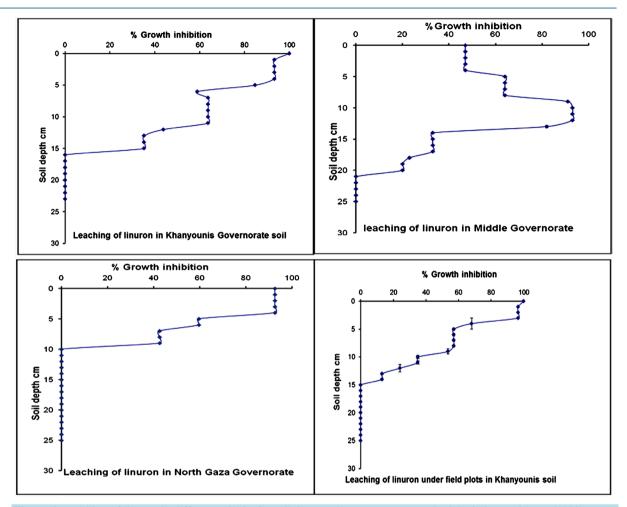


Figure 4. Leaching of linuron in different soils in Gaza Strip. Error bars represent standard deviation. Using Mulukhia as test plant.

analysis found significant difference among leaching potential. Leaching potential of diuron and linuron in the meddle Governorate were significantly different from others.

The differences in leaching potential of diuron and linuron in the tested soils are due to the differences in clay fraction, soil organic matter and soil pH (**Table 2**). Those factors affect the adsorption potential of diuron and linuron in soil [28] which affect the leaching potential in soil. These results are in agreement with previous studies which reported that diuron or linuron did not reach deeper than the 16 cm depth [20] [31] [32].

However, our results showed that the leached amounts of diuron or linuron were very small (**Table 4**), indicating a little risk of groundwater contaminating at currently field rates, moreover, the amount of used irrigation (500 m³/ha) is not so heavy irrigation and can saturate the root zone only but at higher rate or irrigation or rainfall the potential leaching zone would become larger and pose threats to shallow ground water aquifer. Our results agree with previous that studies showed groundwater contamination by diuron or linuron in other countries [19] [20]. In different study [33] diuron and linuron were detected at depths of 20 and 40 cm.

However, from bioestimation of the remaining concentrations at laboratory conditions, one can be concluded that leaching potential of diuron and/or linuron was the lowest in North Gaza soil followed by Khan Younis soil, the larger leaching zone was observed in Meddle zone soil (Table 4). Moreover, the remaining concentrations in the top soil layer (0 - 5 cm) reached about  $0.36 \pm 0.02$  mg/kg soil in North and Khan Younis soil. This concentration may reduce the growth of the plants in the next growing season as claimed by the farmers. In addition, the concentrations of Meddle zone were also high in both soil layers (0 - 5) and (6 - 10). This is also in accord with claim of the farmers in Gaza. The potential solution to this problem is to use lower applied rate of diuron and linuron for weed control.

However, the leaching potential of both herbicides at Meddle zone soil are the largest among all soils, under this condition of leaching potential, phytotoxicity may occur to a deep rooted plants that may grow in the next growing season. Nevertheless, diuron and linuron can be classified with high potential leaching in soil profile in MZ soil. The greater mobility of diuron and linuron in to the soil columns might be due to its lower adsorption. Furthermore the adsorption results in MZ soil was the lowest among all soil [28], whereas the increase in sand content in Middle Governorate soil let diuron move easily. APVMA [34], reported that diuron moved into the soil more than 17 cm in sandy soil. In addition, the clay fraction in KHYZ soil is nearly 2 times higher than MZ soil (Table 1). Similar results were previously shown the clay fraction in soil has greater effect in reducing the leaching potential of herbicides [35].

# 3.4. Leaching Potential of Diuron and Linuron in Field Plots

Leaching potential of diuron and linuron under field conditions is shown in **Figure 5**. The presented results clearly show the herbicidal activity of both herbicides are restricted in the top soil layer (0 - 4 cm), followed by moderate growth inhibition restricted to deeper layers (5 - 10 cm), then low herbicidal activity in the layer (10 - 15 cm). Normal growth of test plant is observed in the layer (15 - 25 cm), these data indicate that diuron or linuron did not leach to this depth. The explanation of this mobility is similar to that under laboratory conditions. However, comparing the leaching depth under laboratory and field condition, one can realize that the leaching depth under field condition is lower than under laboratory condition. The explanation of these results is that under field condition herbicide moves in horizontal and perpendicular directions due to large surface area whereas

Table 4. Remaining concentration (mg/kg soil) determined by bioassay using Mulukhia as test plant.

Soil depth cm	North Gaza		Meddle zone		Kh Y		Kh Y field exp	
	Diuron	Linuron	Diuron	Linuron	Diuron	Linuron	Diuron	Linuron
0 - 5	$0.36 \pm 0.02$	$0.38 \pm 0.06$	$0.12\pm0.03$	$0.18 \pm 0.00$	$0.36 \pm 0.06$	$0.34 \pm 0.01$	$0.13 \pm 0.00$	$0.11 \pm 0.03$
6 - 10	$0.14 \pm 0.03$	$0.16 \pm 0.11$	$0.16\pm0.05$	$0.28 \pm 0.05$	$0.20 \pm 0.07$	$0.23 \pm 0.00$	$0.10\pm0.03$	$0.06 \pm 0.02$
11 - 15	$0.03\pm0.00$	BD	$0.04 \pm 0.05$	$0.23 \pm 0.12$	$0.06 \pm 0.08$	$0.15\pm0.04$	$0.04 \pm 0.03$	$0.01 \pm 0.01$
16 - 20	BD	BD	$0.01 \pm 0.00$	$0.08 \pm 0.05$	BD	BD	BD	BD
21 - 30	BD							

 $Average \pm standard \ deviation. \ Note: BD = below \ detection \ limit, \ average \pm standard \ deviation.$ 

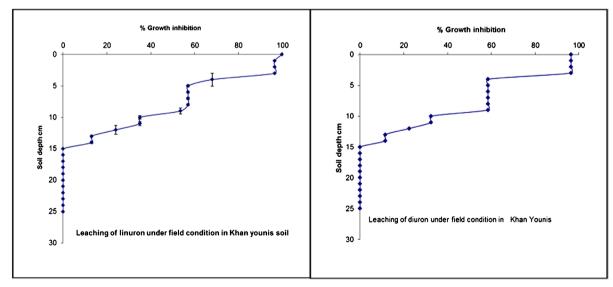


Figure 5. Leaching of linuron and diuron under field conditions in Khan Younis soils. Error bars represent standard deviation. Using Mulukhia as test plant.

under laboratory condition, mobility is restricted to downward movement due to narrow soil surface. In addition, under filed condition, due to evaporation and high temperature in summer time upward movement may occur. Accordingly the leaching zone under field condition is smaller than under laboratory. This explanation was previously used to explain similar phenomena [1]. Bio-estimation of the concentrations of diuron and linuron under filed condition in Kh Y field exp showed that the leaching potential were in the top 5 cm with considerable concentration whereas below this depth nearly low concentration was observed. Comparison between field and laboratory bio-estimation (Table 4) one can realize that the bioestimated concentrations under field conditions is several times lower that under laboratory conditions. The explanation of these results is that under field condition some evaporation and/or photochemical degradation may occur. The explanation is supported by previous results [35] [36].

## 4. Conclusions

This study describes the leaching potential of linuron and diuron in selected soil from Gaza Strip and provides information on the distribution of diuron and linuron concentrations in soil profile. Consequently, the study advances our understanding about the losses of herbicidal activity of the tested herbicides under field conditions; and provides answers to the questions raised by farmers in Gaza. Regardless of the similarity in some points, the leaching potential of both herbicides in Gaza soils varies from one soil to another and considerably is affected by three factors:

- 1) Clay fraction which is negatively charged can influence the adsorption of herbicides.
- 2) Soil pH affects the solubility of herbicides in soil solution.
- 3) Organic matter which acts as a non-polar phase or surface and consequently is the major sorbent for organic herbicides.

The leaching potential of both herbicides in the selected soils was strongly retained in the top soil layers except in the Middle Governorate soil. The interesting results of this study are both herbicides loss considerable concentration under field conditions. The rational of this work comes from the fact that soil pH and organic matter have strongly effluences in adsorption and mobility. Field experiments demonstrated less leaching potential than that in the lab experiments.

## **Acknowledgements**

Dr. Y. El-Nahhal acknowledges Alexander von Humboldt Stiftung Foundation Fellowship Grant no IV-PAL/1104842 STP, Germany. Special thanks go to Prof. Dr. G. Lagaly at Kiel University, Prof. Shourrmann Germany.

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