

A Validated High-Performance Thin-Layer Chromatography (Hptlc) Method for the Quantitative Determination of Tricin in Two *Spergularia* Species

Omnia Gamal El-Dien, Eman Shawky*, Amal H. Aly, Rokia M. Abdallah, Nabil A. Abdel-Salam
Pharmacognosy Department, Faculty of Pharmacy, University of Alexandria, Alexandria, Egypt
Email: *shawkyeman@yahoo.com

Received September 24, 2013; revised October 26, 2013; accepted November 8, 2013

Copyright © 2013 Omnia Gamal El-Dien *et al.* This is an open access article distributed under the Creative Commons Attribution License, which permits unrestricted use, distribution, and reproduction in any medium, provided the original work is properly cited.

ABSTRACT

A simple, sensitive, precise and robust high-performance thin-layer chromatographic (HPTLC) method for analysis of triclin has been developed and validated for the determination of triclin in genus *Spergularia*. Analysis of triclin was performed on pre-coated TLC aluminum plates with silica gel as the stationary phase. Linear ascending development was carried out in double twin trough glass chamber saturated with a mobile phase consisting of chloroform: methanol (37:3 v/v). Spectro-densitometric scanning was performed by TLC scanner III (CAMAG) in absorbance mode at the wavelength of 270 nm. The system was found to give compact spots for triclin (R_f value of 0.34 ± 0.04). The linear regression analysis data for the calibration plots showed a good linear relationship with r² = 0.9981 in the concentration range 2 - 15 µl with respect to peak area. According to the International Conference on Harmonization (ICH) guidelines, the method was validated for precision, recovery, and robustness. The triclin content quantified from *Spergularia marina* and *S. daiandra* was found to be 0.398 mg/100g and 1.642 mg/100g fresh weight, respectively. Statistical analysis of the data showed that the method is reproducible and selective for the estimation of triclin.

Keywords: *Spergularia marina*; *Spergularia daiandra*; HPTLC; Quantitative; Tricin

1. Introduction

Tricin possesses several biological properties that are superior to other flavonoids and polyphenols in terms of agricultural, nutraceutical and pharmaceutical values [1]. It has long been credited for its beneficial effects: as an antioxidant due to its potent inhibition of lipoperoxidation and its sparing effect on vitamin E in erythrocyte membranes; as an antiviral; antihistaminic; immunomodulatory and anti-tubercular activities [2-4]. The promising anti-ulcerogenic activity of triclin and its 7-glucoside was recently reported with a curative ratio of 77 and 79%, respectively [5]. Tricin is a potent COX (Cyclooxygenase) inhibitor. The inhibition of COX enzyme activity by triclin was apparently isoenzyme-nonselective, which renders it more comparable with non-steroidal anti-inflammatory drugs, such as sulindac and aspirin, rather than to coxibs [6-8].

One of the most prominent and well documented features of triclin is its potential antitumor/anticancer activity.

Based on a study of structure-activity relationships of flavonoid-induced cytotoxicity on human leukemia cells, triclin is considered one of the most potent anti-cancer agents tested so far and this is most probably due to the stability of its structure [9].

Recently a simple HPLC method for the quantitation of triclin in human plasma has been reported and cross-validated in mouse plasma, liver and small intestinal mucosa. The work was designed to contribute to the development of triclin as a potential chemo-preventive agent in humans [10].

Being the dominant flavone pigment in whole wheat meal, triclin has prompted the elaboration of a method for triclin isolation and quantification, based on ultrasonic-assisted extraction and ultra-performance HPLC purification, and guided by electrospray ionization MS [11].

A recent HPLC method for triclin isolation and quantification in the pressed juice of *Medicago sativa* was established [1]. In addition, a simple, precise, and rapid method was developed for the determination of triclin and 4'-methoxy-tricin in the Tibetan herbal medicine of *Py-*

*Corresponding author.

rethrum tatsienense by the high-performance liquid chromatographic technique coupled with photodiode array detection [12].

Among the modern Analytical tools, HPTLC is a powerful analytical method equally suitable for qualitative and quantitative analytical tasks. HPTLC is playing an important role in today's analytical world, not in competition to HPLC but as a complementary method [13].

HPTLC produces visible chromatograms complex information about the entire sample available at a glance. Multiple samples are seen simultaneously, so that reference and test samples can be compared for identification. Similarities and differences are immediately apparent and with the help of the image comparison. Several chromatograms can be compared directly, even from different plates. They can be evaluated either by the image based software video-scan or by scanning densitometry with TLC scanner, measuring the absorption and/or fluorescence of the substances on the plate. TLC is an offline technique: the subsequent steps are relatively independent, allowing parallel treatment of multiple samples during chromatography, derivatization and detection.

We managed to isolate tricin for the first time from genus *Spergularia*. Tricin was isolated from the chloroform fraction of *Spergularia marina*; meanwhile it was found in the chloroformic fraction of *Spergularia diandra* in a large amount.

In view of the fore-mentioned points, the aim of this work is to develop a simple, non expensive, accurate, specific and repeatable HPTLC method for the determination of tricin in Genus *Spergularia*. Literature review revealed that only one HPLC method has been reported for the analysis of tricin among other five flavonoids, but the method was not thoroughly validated [14].

2. Experimental

2.1. Chemicals and Reagents

HPTLC analyses were performed on Merck 20 cm × 10 cm (0.25 mm) plates. Tricin used as standard material was obtained from the Department of Pharmacognosy, Faculty of Pharmacy, University of Alexandria and tested for its purity by TLC and spectroscopic method. All the reagents used in the experiment were of analytical grade and were supplied by Merck, Darmstadt, Germany.

2.2. Preparation of Standard Solution

A weight of 4.0 mg of standard tricin was accurately weighed, quantitatively transferred into a 10 ml volumetric flask, dissolved in methanol and the volume was adjusted with the same solvent.

2.3. Preparation of Sample Solutions

Fresh aerial parts, 500 g of each of *Spergularia marina*

and *Spergularia diandra* were accurately weighed and exhaustively extracted using (2 L) 70% ethanol at room temperature. Each alcoholic extract was concentrated to 200 ml volume, fractionated successively by (600 ml) petroleum ether, then by (600 ml) chloroform. The chloroform fraction was evaporated under reduced pressure to give 440 mg of extract for *S. marina* and 770 mg for *S. diandra*. A weight of 225 mg of each of the chloroformic fraction of *Spergularia marina* and *Spergularia diandra* were accurately weighed, dissolved in a mixture of methanol and chloroform (8:2), transferred quantitatively to 10 ml volumetric flask, adjusted to volume with the same mixture and shaken to mix thoroughly to give samples A and B, respectively.

2.4. HPTLC—Densitometric Procedure

2.4.1. Instrumentation

Sample solutions for HPTLC analyses were applied by means of a CAMAG (Wilmington, NC) Linomat IV automated spray-on band applicator. Zones were quantified by linear scanning at 270 nm with a CAMAG TLC Scanner 3 with a deuterium source in the reflection mode, slit dimension settings of length 6 and width 0.1, monochromator bandwidth 20 nm, and a scanning rate of 15 mm·s⁻¹. The peak areas of chromatograms were determined using CATSTLC software (version 4.X).

2.4.2. Chromatographic Procedure

Sample and standard solutions were applied in the form of bands on pre-coated HPTLC silica gel plates 60 F-254 (20 cm × 10 cm with 250 μm thickness) by means of Linomat IV automated spray-on band applicator operated with the following settings: band length 6 mm, application rate 15 s/μl, distance between each two bands 4 mm, distance from the plate side edge 1 cm, distance from the bottom of the plate 1 cm.

The volumes applied for routine analysis were duplicate 1.0 μl, 3.0 μl and 5.0 μl of the TLC tricin standard (0.4, 1.2 and 2.0 μg) and triplicate 6 μl and 4 μl aliquots of samples solutions A and B respectively. A volume of 20 ml of mobile phase (system IV) were used for development.

Linear ascending development of the plates was carried out in 20 cm × 20 cm Camag HPTLC twin trough chamber saturated with the mobile phase. The optimized chamber saturation time for the mobile phase was 30 min at room temperature. Plates were developed to a distance of 8 cm beyond the origin. The development time was 17 min. After development, the plates were air-dried for 5 min. Densitometric scanning was performed on Camag TLC scanner III in the reflectance-absorbance mode at λ 270 nm and operated by WINCATS software (V. 3.1). The source of radiation utilized was deuterium lamp

emitting a continuous UV spectrum between 190 and 400 nm. The slit dimension was kept at 6 mm × 0.1 mm. Concentrations of the samples and standards chromatographed were determined from the intensity of diffusely reflected light. Evaluation was done by peak area measurement with linear regression.

2.5. Validation

The method was validated according to the ICH guidelines on the validation of analytical methods [15]. All results were expressed as percentages, where n represents the number of values. For the statistical analysis, Excel 2003 (Microsoft Office) was used. A 5% level of significance was selected.

2.5.1. Calibration and Quantification

As recommended by the International Committee on Harmonization (ICH) guidelines [15], a calibration curve was established using six analyte concentrations (1.0, 2.0, 3.0, 4.0, 5.0 and 6.0 µl/zone) of the TLC standard applied in triplicate, representing 0.4 - 2.4 µg of triclin. For routine analytical procedures, a three-point calibration curve within this range was used, produced by applying duplicate 1.0, 3.0 and 5.0 µl (0.4, 1.2, 2.0 µg) of the HPTLC standard on each plate.

2.5.2. Specificity

The specificity of the method was determined by analyzing standard drug and test samples. The spots for triclin and samples were confirmed by comparing the R_f and spectrum of the spot with that of a standard. The peak purity of triclin was assessed by comparing the spectra at three different levels, that is, peak start (S), peak middle (M), and peak end (E) positions of the spot.

2.5.3. Accuracy

The accuracy of the method was validated by a standard addition analysis. The sample solutions were spiked with three different known concentrations of triclin. An aliquot of 2 ml of sample B solution of known previously determined concentration (0.24 mg/ml) was mixed with 3 ml of standard triclin solution (0.12 mg/ml) to give mixture 1, another 2 ml aliquot of the same sample solution was mixed with 3 ml of standard triclin solution (0.24 mg/ml) to give mixture 2, and a third 2 ml aliquot of the same sample solution was mixed with 3 ml of standard triclin solution (0.36 mg/ml) to give mixture 3. The original and the three fortified sample solutions (mixtures 1, 2 and 3) were analyzed on the same plate by application of triplicate 4.0 µl, 6.0 µl, 4.0 µl and 4.0 µl volumes, respectively, in addition to the three standards described earlier for routine analyses which were applied in duplicate. The difference between the expected concentrations and the found ones was calculated to determine the accuracy of

the method.

2.5.4. Precision

In accordance with the ICH guidelines [15], precision was determined by independent repeated analysis of six different triclin standard solutions by the use of the same equipment and the same analytical procedure in the same laboratory and on the same plate. The assay was repeated in the same day for intra-day precision and on different days to obtain inter-day precision.

2.5.5. Limit of Detection (LOD) and Limit of Quantification (LOQ)

According to the ICH guidelines, the detection limit (DL) may be expressed as $DL = 3.3 \sigma/S$, while, the quantitation limit (QL) may be expressed as $QL = 10 \sigma/S$, where σ = the standard deviation of the response S = the slope of the calibration curve. The slope S is estimated from the calibration curve. The estimate of σ was carried out by studying the calibration curves, where the standard deviations of y-intercepts of regression lines were used as the standard deviation.

3. Results and Discussion

Mixtures of several mobile phases were tried to separate spot of triclin from other spots and get stable peak. The solvent system used was System chloroform: methanol (37:3 v/v) was selected for analysis of triclin, which gave good resolution. Good chromatogram in **Figures 1-3** was attained with R_f value 0.38 ± 0.04 . The wavelength of 270 nm was used for quantification of sample as it gave the best sensitivity.

The selectivity of the separation and the specificity of the detection were shown by the densitograms and comparison of the superimposed UV spectra, respectively (**Figure 4**). The peak purity of triclin was assessed by

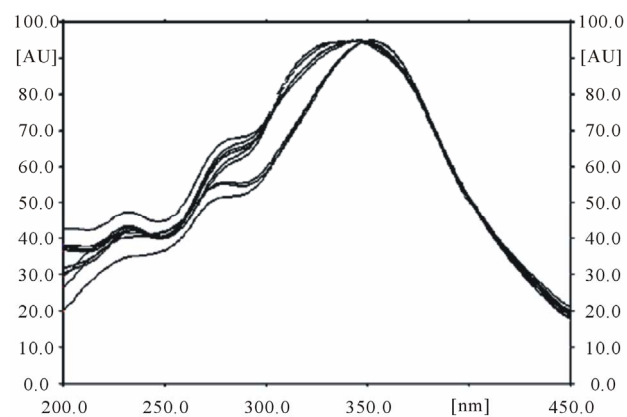


Figure 1. Overlay of *in-situ* absorption spectra obtained from the band of standard triclin and the corresponding band obtained from chloroform extracts of *S. marina* and *S. diandra* scanning from 200 to 400 nm.

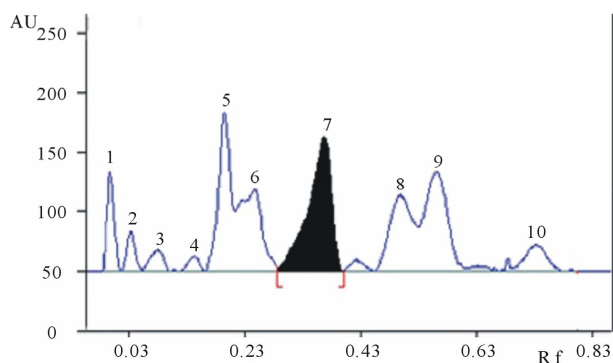


Figure 2. HPTLC scan-densitogram showing the separation of triclin (7) from other components in the sample A at 270 nm.

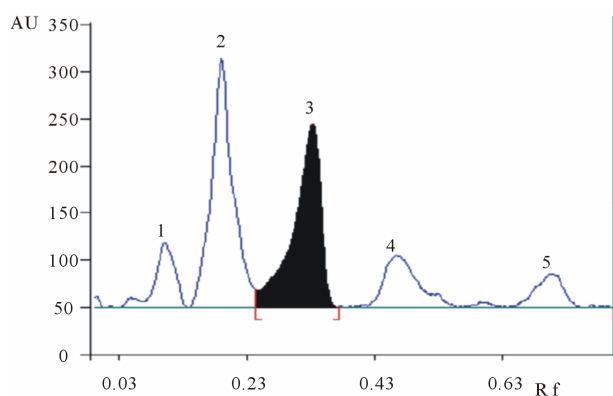


Figure 3. HPTLC scan-densitogram showing the separation of triclin (3) from other components in the sample B at 270 nm.

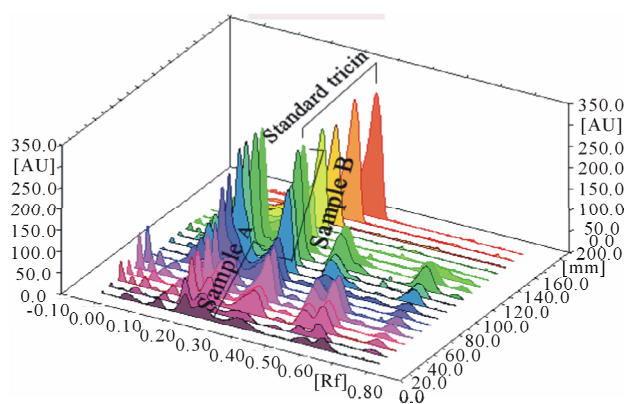


Figure 4. A 3-dimentional of a series of sample A, B and standard triclin applied on the same plate with increasing concentrations showing an increase in the area of successive peaks in samples and standard.

comparing its respective spectra at the peak start, apex, and peak end.

Visual observation of the calibration curves was done which reveals that the calibration curve is linear. The intercept of the calibration lines is slightly different from zero. From these we can say that, this kind of methods

shows saturation on the spots and because of that it produces curved calibrations. However, here linearity can fit the calibration responses at narrow range but intercept deviate from zero. The results showed linear relationship between the peak areas and the triclin concentrations. The plot was linear in the range of 0.4 - 2.4 $\mu\text{g}/\text{spot}$ ($y = 7463x + 583.6$, $r^2 = 0.998$). The three-point calibration was repeated many times and was also found to have a linear regression correlation coefficient of 0.999.

LOD and LOQ were found to be 0.0513 and 0.155 $\mu\text{g}/\text{spot}$ respectively. Results are shown in **Table 1**.

The results of repeatability and intermediate precision study of triclin are depicted in **Table 2**. The developed method was found to be precise as the %RSD values for repeatability and intermediate precision studies were less than 2%.

Results of recovery studies by addition of triclin standard in triplicate are depicted in **Table 3**.

The method was found to be robust by studying the effect of alteration of mobile phase composition. The results of change in mobile phase composition in terms of standard deviation of the peak areas calculated as the % RSD was found to be less than 2% (**Table 4**).

The method was employed for the determination of triclin content in the total extract of *Spergularia marina* and *S. diandra*, where % w/w triclin in fresh total weight was found to be 0.398 mg/100g fresh weight and 1.642 mg/100g fresh weight for *Spergularia marina* and *S. diandra*, respectively.

4. Conclusion

TLC and TLC-densitometry have been largely investigated for purity testing, pharmaceutical dosage form assay and herbal fingerprinting [16]. These methods are flexible and cost-effective and present the advantage of the simultaneous processing of standards and samples with versatile detection possibilities, including a great variety of post-chromatographic derivatization reagents due to the short analysis time and low solvent consumption. The results of the analysis of marketed formulation using the proposed HPTLC method are highly reproducible and reliable. HPTLC method is found to be simple, sensitive, accurate, precise, specific, robust and economi-

Table 1. Linear regression data for the calibration curve of triclin.

	Triclin
Linearity range ($\mu\text{g}/\text{spot}$)	0.32 - 2.88
Intercept (a)	583.6
Slope (b)	7463
Correlation coefficient (r)	0.999172
LOD ($\mu\text{g}/\text{spot}$)	0.051238
LOQ ($\mu\text{g}/\text{spot}$)	0.155266

Table 2. Precision of the HPTLC method for the determination of tricrin (repeatability).

Experiment number	Intra-day precision			Inter-day precision		
	Concentration ($\mu\text{g}/\text{spot}$)	Mean \pm SD	RSD %	Concentration ($\mu\text{g}/\text{spot}$)	Mean \pm SD	RSD %
1	0.399	0.401	1.108	0.401	0.404	1.046
	0.406	\pm		0.402	\pm	
	0.398	0.004		0.409	0.004	
2	0.808	0.809	1.624	0.809	0.807	0.253
	0.823	\pm		0.805	\pm	
	0.797	0.013		0.807	0.002	
3	1.218	1.218	0.207	1.218	1.208	0.969
	1.216	\pm		1.211	\pm	
	1.221	0.003		1.195	0.012	
4	1.583	1.615	1.815	1.615	1.620	0.546
	1.622	\pm		1.615	\pm	
	1.640	0.029		1.630	0.009	
5	1.984	2.008	1.369	2.008	2.015	0.504
	2.002	\pm		2.027	\pm	
	2.038	0.028		2.009	0.010	
6	2.273	2.297	1.941	2.422	2.377	1.662
	2.269	\pm		2.362	\pm	
	2.348	0.045		2.348	0.040	

Table 3. Results of the standard addition experiments.

Mix.	Concentration of tricrin in sample (mg/ml)	Concentration of tricrin added (mg/ml)	Concentration of tricrin found in mixture (mg/ml) ^a	Recovery (%) \pm SD	RSD %
1	0.24	0.12	0.166	98.95 \pm 1.56	1.59
2	0.24	0.24	0.242	100.66 \pm 1.614	1.6
3	0.24	0.36	0.317	101.48 \pm 1.44	1.41

^aEach concentration is the average of three determinations.

Table 4. Robustness testing.

Parameter	S.D.	R.S.D. %
Mobile phase composition	0.072	1.3
Amount of mobile phase	0.03	0.82
Time from spotting to chromatography	0.05	0.97
Time from chromatography to scanning	0.06	0.74

cal and it can be used for the routine analysis of tricrin in genus *Spergularia* L.

REFERENCES

- [1] J.-M. Zhou and R. K. Ibrahim, "Tricrin—A Potential Multifunctional Nutraceutical," *Phytochemistry Reviews*, Vol. 9, No. 3, 2010, pp. 413-424. <http://dx.doi.org/10.1007/s11101-009-9161-5>
- [2] M. Watanabe, "Antioxidative Phenolic Compounds from Japanese Barnyard Millet *Echinochloa crus-galli* Grains," *Journal of Agricultural and Food Chemistry*, Vol. 47, No. 11, 1999, pp. 4500-4505. <http://dx.doi.org/10.1021/jf990498s>
- [3] Y. S. Kwon, E. Y. Kim, W. J. Kim, et al., "Antioxidant Constituents from *Setaria viridis*," *Archives of Pharmacal Research*, Vol. 25, No. 3, 2002, pp. 300-305. <http://dx.doi.org/10.1007/BF02976630>
- [4] Y. S. Kwon and C. M. Kim, "Antioxidant Constituents from the Stem of *Sorghum Bicolor*," *Archives of Pharmacal Research*, Vol. 26, No. 7, 2003, pp. 535-539. <http://dx.doi.org/10.1007/BF02976877>
- [5] A. S. Awaad, N. H. Mohamed, D. J. Maitland, et al., "Antiulcerogenic Activity of Extract and Some Isolated Flavonoids from *Desmostachia bipinnata* L. Stapf," *Records of Natural Products*, Vol. 2, No. 3, 2008, pp. 76-82.
- [6] H. Cai, M. Al-Fayez, R. G. Tunstall, et al., "The Rice Bran Constituent Tricin Potently Inhibits Cyclooxygenase Enzymes and Interferes with Intestinal Carcinogenesis in Apc Minmice," *Molecular Cancer Therapeutics*, Vol. 4, No. 9, 2005, pp. 1287-1292. <http://dx.doi.org/10.1158/1535-7163.MCT-05-0165>
- [7] H. Cai, S. Sale, R. Schmid, et al., "Flavones as Colorectal Cancer Chemopreventive Agents: Phenol-O-Methylation Enhances Efficacy," *Cancer Prevention Research*, Vol. 2, No. 8, 2009, pp. 743-750. <http://dx.doi.org/10.1158/1940-6207.CAPR-09-0081>
- [8] J. Yan, L. Sun, X. Zhang, et al., "A New Flavone from *Lycopodium japonicum*," *Heterocycles*, Vol. 65, No. 3, 2005, pp. 661-666. <http://dx.doi.org/10.3987/COM-04-10314>
- [9] K. Plochmann, G. Korte, E. Koutsilieri, et al., "Structure-Activity Relationships of Flavonoid-Induced Cytotoxicity on Human Leukemia Cells," *Archives of Biochemistry and Biophysics*, Vol. 460, 2007, pp. 1-9. <http://dx.doi.org/10.1016/j.abb.2007.02.003>

- [10] H. Cai, W. P. Steward and A. J. Gescher, "Determination of the Putative Cancer Chemopreventive Flavone Tricin in Plasma and Tissues of Mice by HPLC with UV-Visible Detection," *Biomedical Chromatography*, Vol. 19, No. 7, 2005, pp. 518-533. <http://dx.doi.org/10.1002/bmc.473>
- [11] S. Zhao, L.-J. Zhao, X.-H. Gu, X.-Z. Wang and J. C. Tian, "Determination of Tricin in Wheat Grain by Ultra Performance Liquid Chromatography Tandem Mass Spectrometry," *Chinese Journal of Analytical Chemistry*, Vol. 37, No. 6, 2009, pp. 873-876.
- [12] X. Liu, A.-M. Yang and Y.-P. Shi, "Determination of Flavonoids in Tibetan Herbal Medicine of *Pyrethrum tatsienense* by HPLC with Photodiode Array Detection," *Journal of Analytical Chemistry*, Vol. 63, No. 5, 2008, pp. 472-476. <http://dx.doi.org/10.1134/S1061934808050122>
- [13] "High Performance Thin Layer Chromatography (HPTLC): A Modern Analytical Tool for Biological Analysis," *Nature and Science*, Vol. 8, No. 10, 2010, pp. 58-61.
- [14] "HPTLC Analysis of the Flavonoids in Eight Species of *Indocalamus* Leaves," *Journal of Planar Chromatography*, Vol. 24, No. 5, 2011, pp. 394-399.
- [15] ICH-Guidelines Q2(R1), Validation of Analytical Procedures: Text and Methodology. http://www.ich.org/fileadmin/Public_Web_Site/ICH_Products/Guidelines/Quality/Q2_R1/Step4/Q2_R1_Guideline.pdf
- [16] H. Hajimehdipoor, M. Khanavi, H. Zahedi, Z. Abedi, N. Kalantari Khandani, N. Adib and M. Pirali Hamedani, "Fingerprint Study of *Thymus* spp. by TLC," *Journal of Medicinal Plants*, Vol. 8, No. 31, 2009, pp. 19-24.